

# From Chemistry to Biology: Getting the RNA World started

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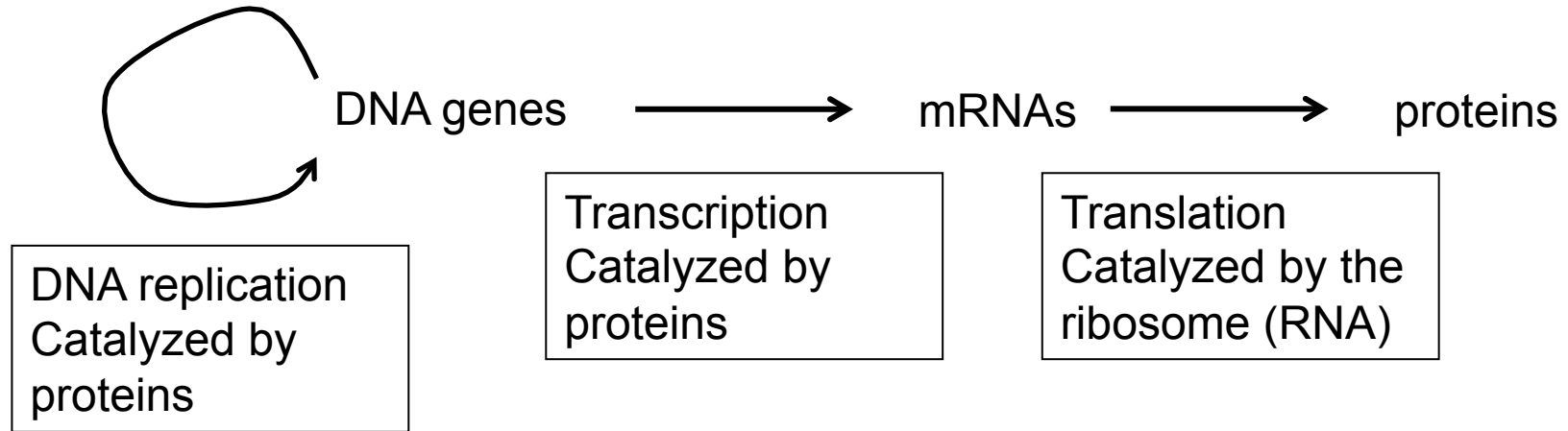
Supported by:

- NSERC
- Harry Lonsdale (Origin of Life Research Award)

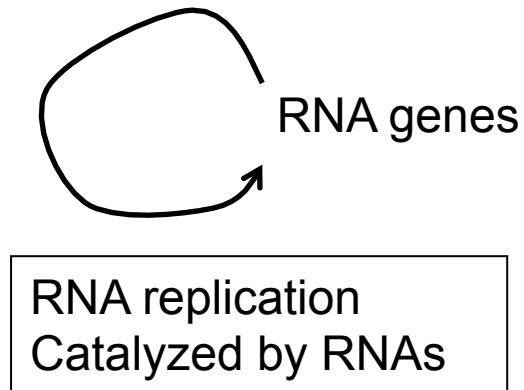


# Definition of the RNA World

Modern Organisms.



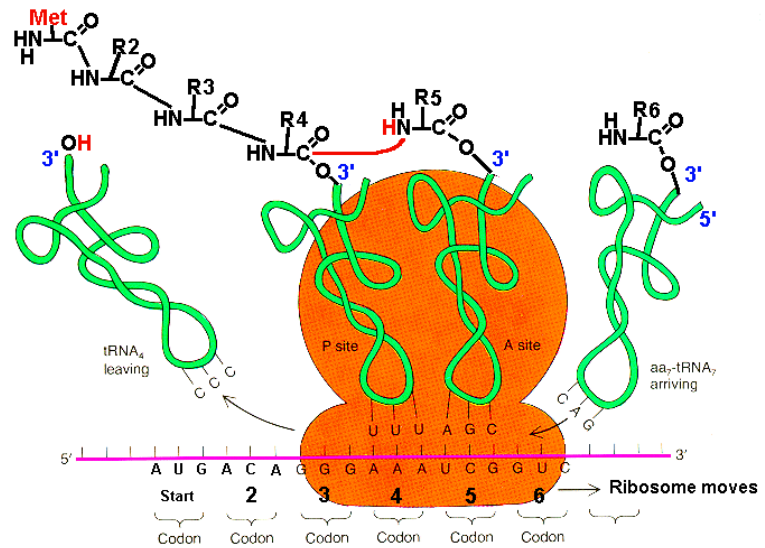
Hypothetical RNA World Organisms.



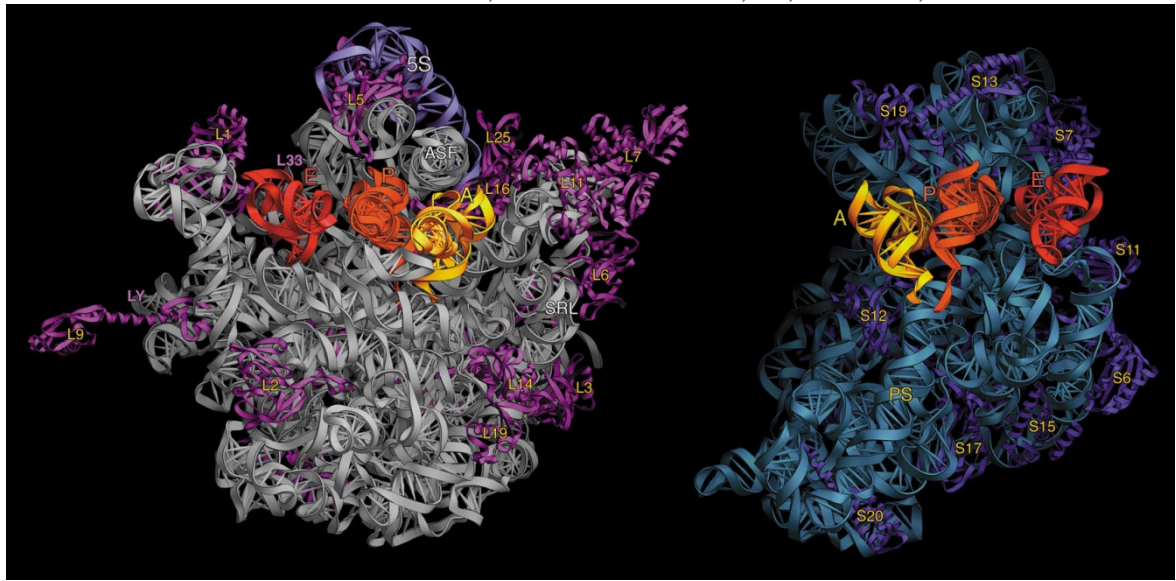
**ribozyme** =  
catalytic RNA  
strand

Evidence from current biology

Protein synthesis in modern cells relies on rRNA, tRNA, and mRNA

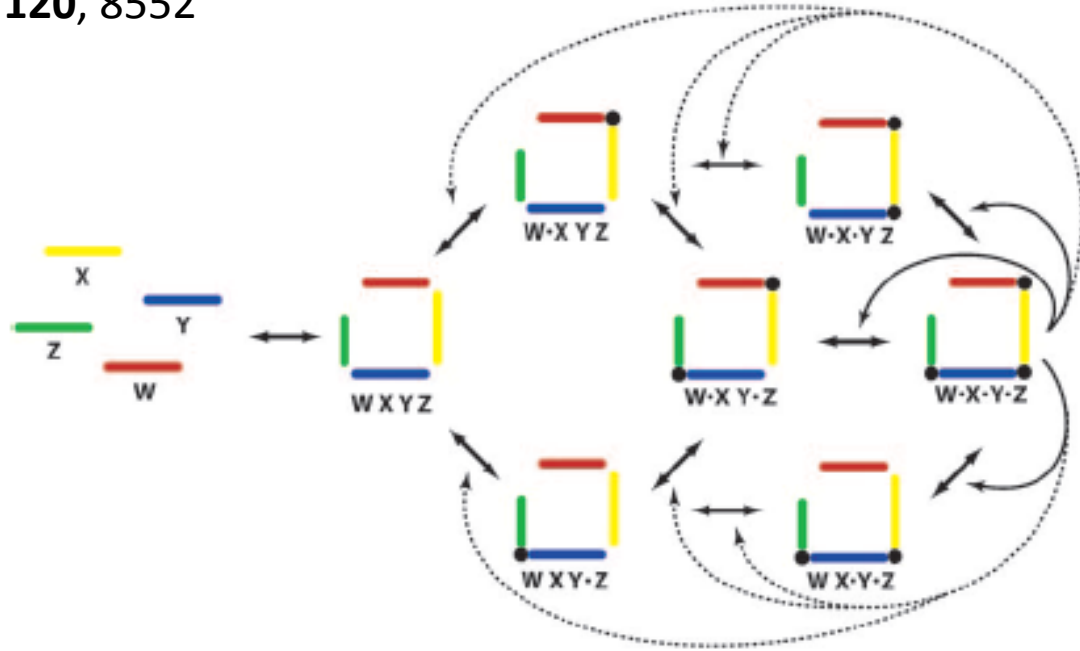


Modified from Griffiths et al., AN INTRODUCTION TO GENETIC ANALYSIS, 6th Ed., W.H. Freeman & Co., 1996.

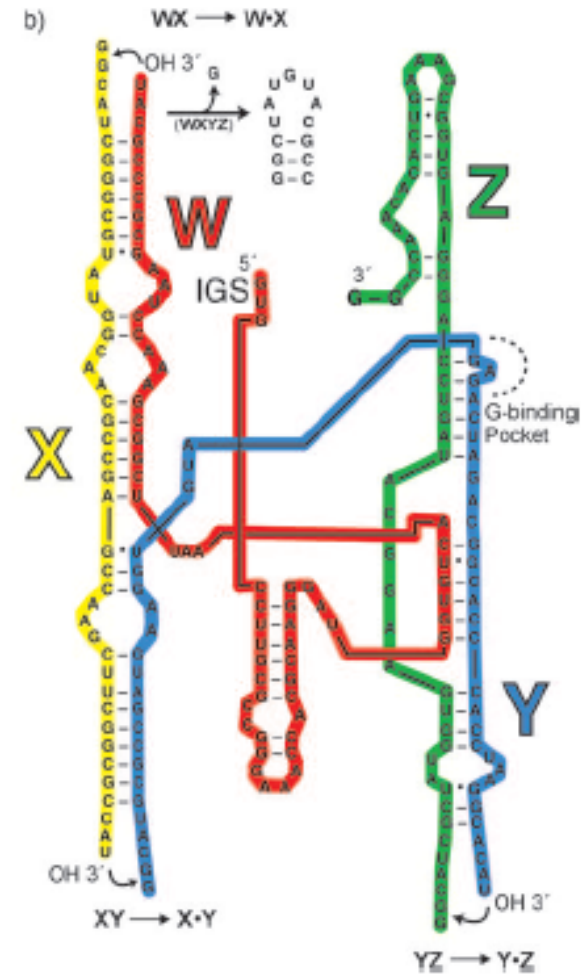


# Laboratory Ribozymes: Recombinases

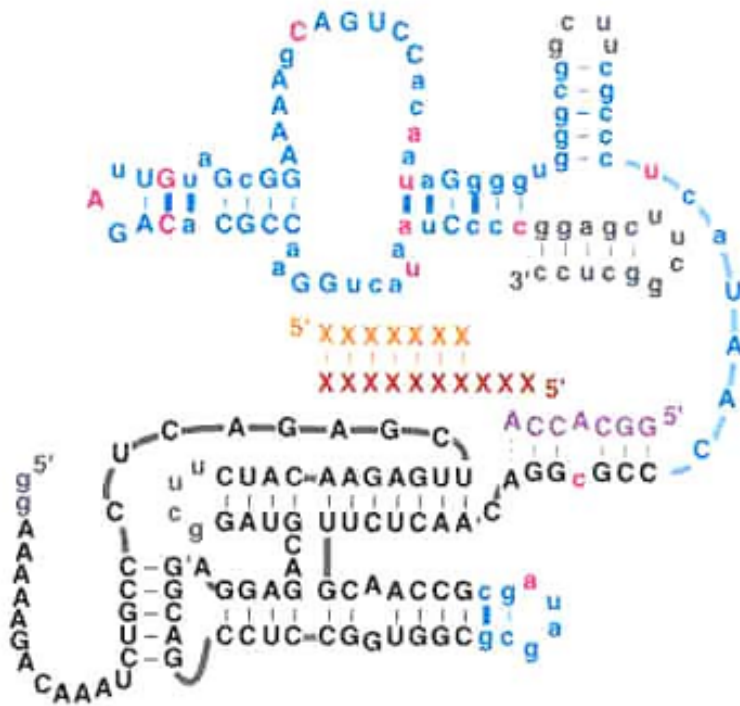
E.J. Hayden, G.v. Kiedrowski & N.  
Lehman, *Angew. Chem. Int. Edit.* (2008)  
**120**, 8552



Catalyst is autocatalytic given a supply of W X Y Z.  
The non-covalent assembly is also a catalyst.

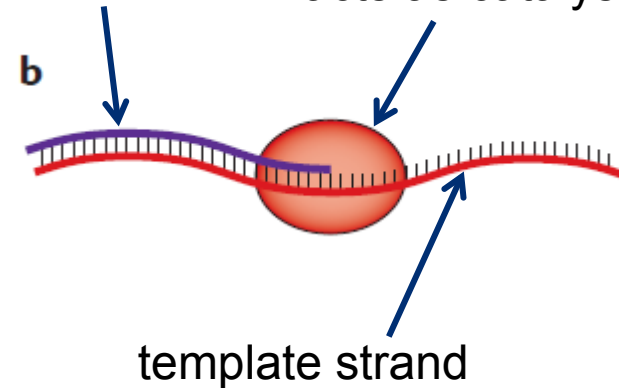


# Laboratory Ribozymes: Polymerases



new strand is complementary

folded strand acts as catalyst



Johnstone et al. (2001) *Science* - Primer extended by up to 14 nucleotides

Wochner et al. (2011) *Science* - up to 95 nucleotides

Attwater et al. (2013) *Nature Chemistry* - up to 200 nucleotides

We're getting there ...

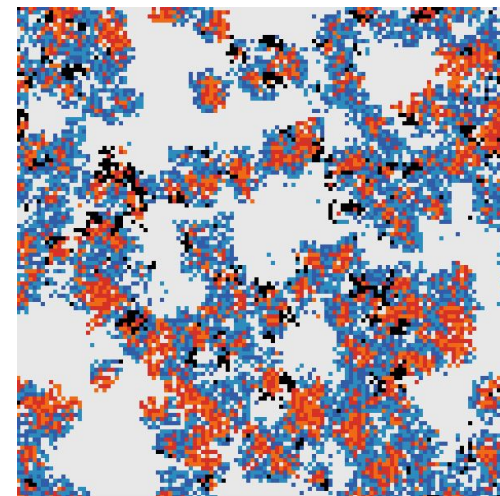
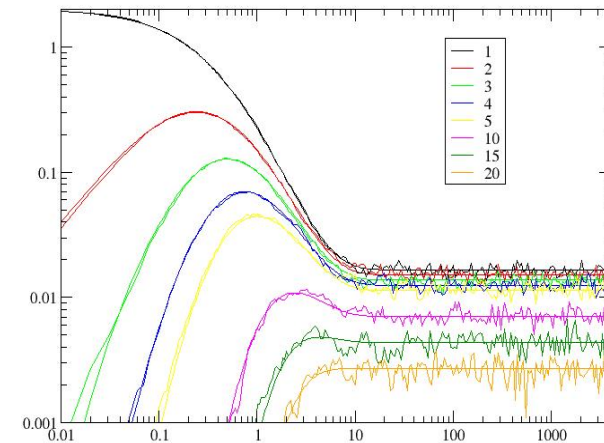
## Two key questions for the RNA World (that a theoretical biophysicist can address)

How was RNA synthesized abiotically?

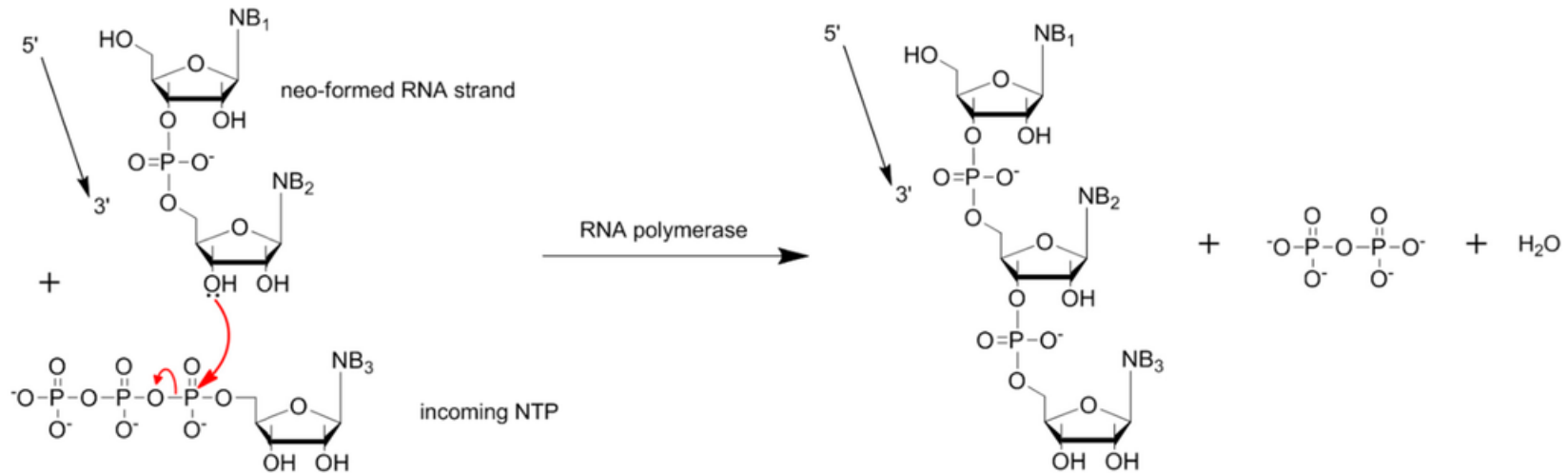
- Wet-dry cycling drives formation of long strands

How did replicating sequences emerge from a mixture of random strands?

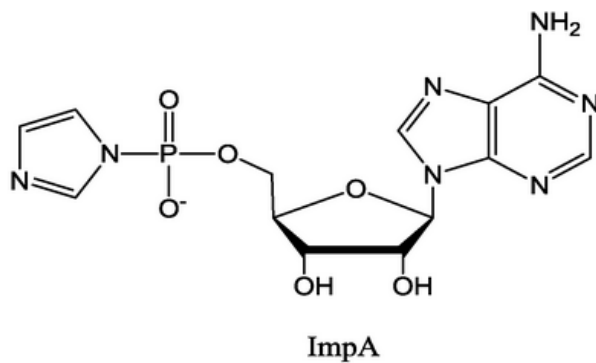
- Evolutionary models for replication in the RNA World.



## How to make RNA polymers? – Activated nucleotides



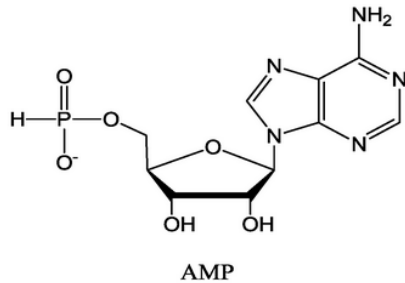
Modern organisms do it using tri-phosphates and a protein catalyst



Ferris (2002) *Orig. Life Evol. Biosph.*  
Montmorillonite catalyzed synthesis of  
RNA oligonucleotides (30-50 mers)

Uses imidazole-activated nucleotides

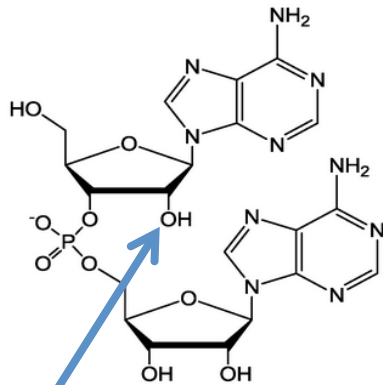
## Can we make RNA polymers from mono-phosphates?



Polymerization tends to be up-hill thermodynamically.

Hydrolysis is faster than polymerization.

Equilibrium is dominated by monomers and very short oligomers.



2' OH initiates hydrolysis in RNA (but not DNA)

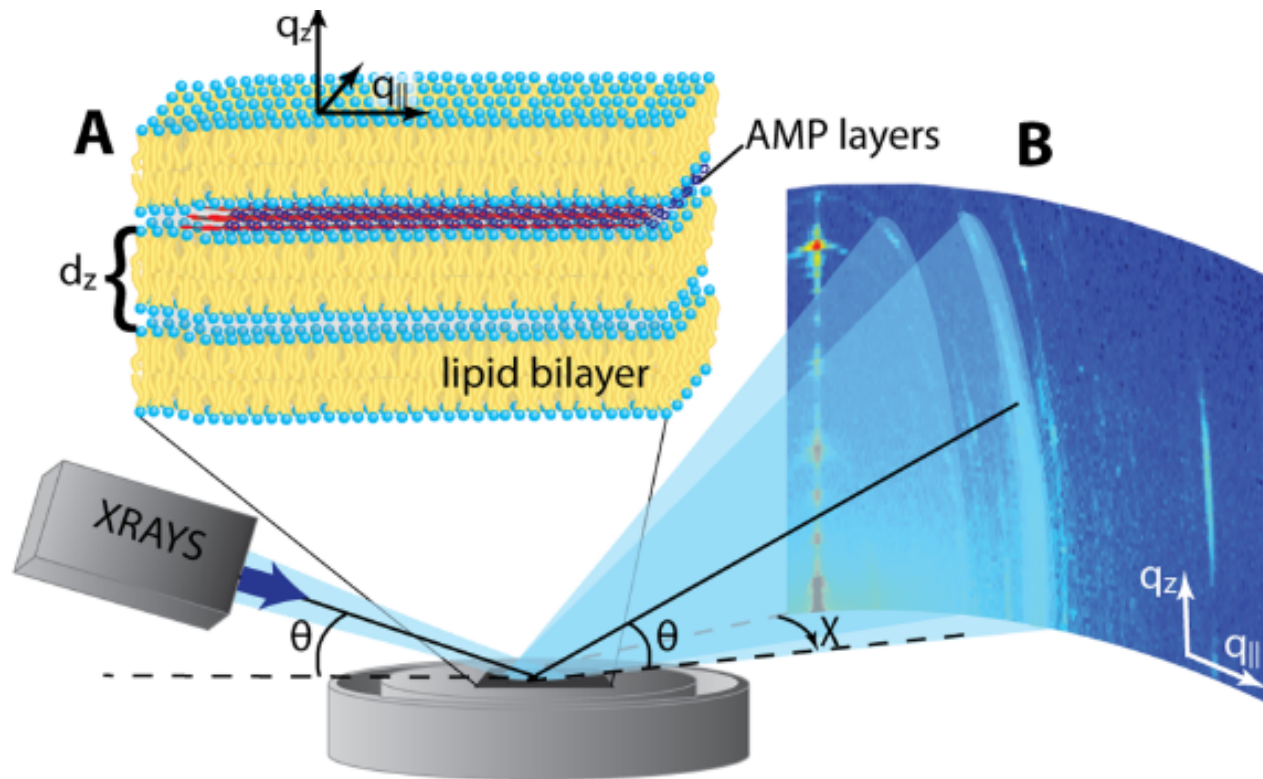
Lipid assisted synthesis of RNA-like polymers from mononucleotides.

Drying AMPs between lipid bilayers creates favourable conditions for polymerization

Rajamani ... Deamer et al. (2008) *Orig. Life Evol. Biosph.*

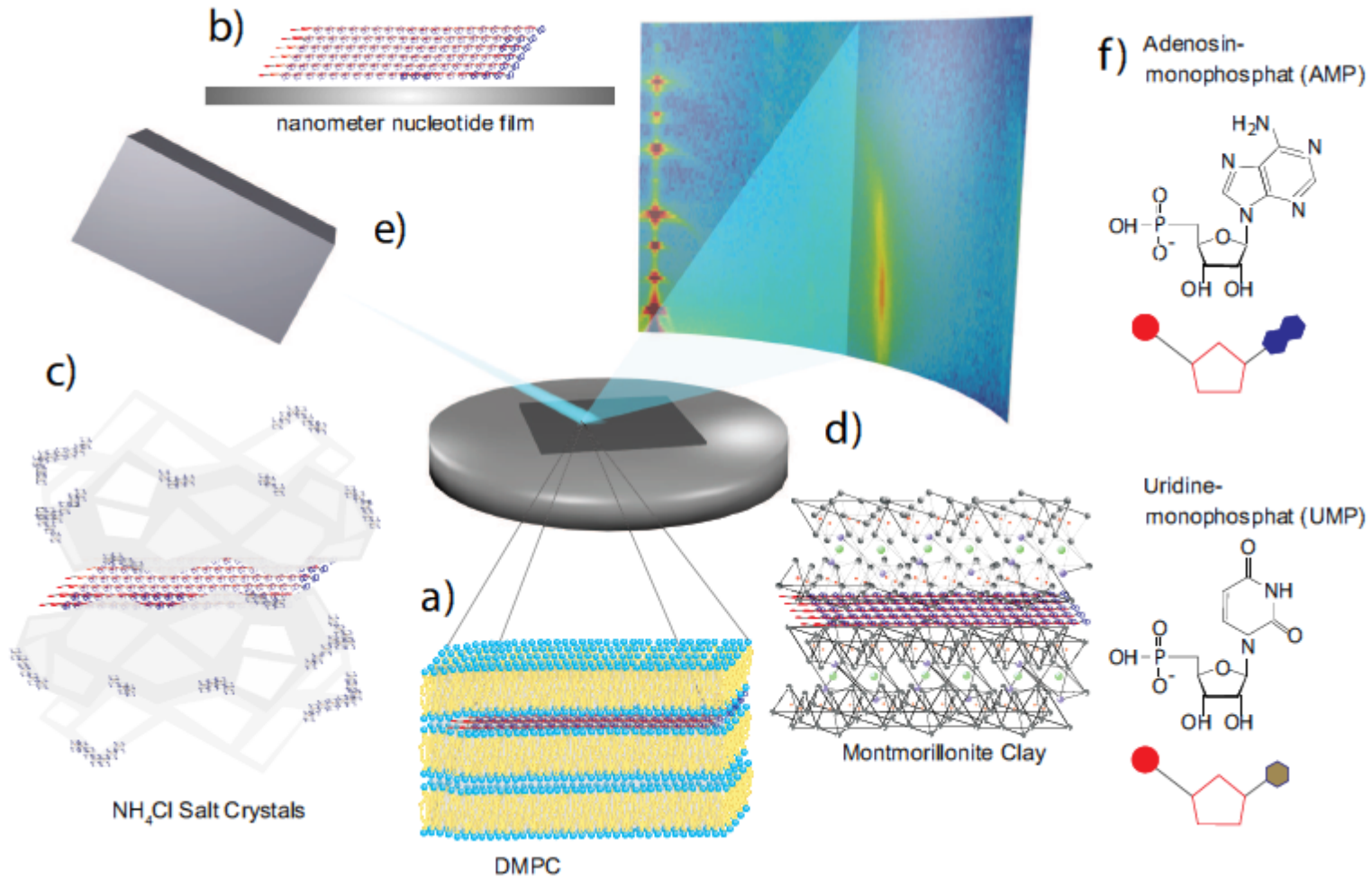


X-ray scattering shows ordered structures of nucleotides between lipid bilayers in lamellar phases



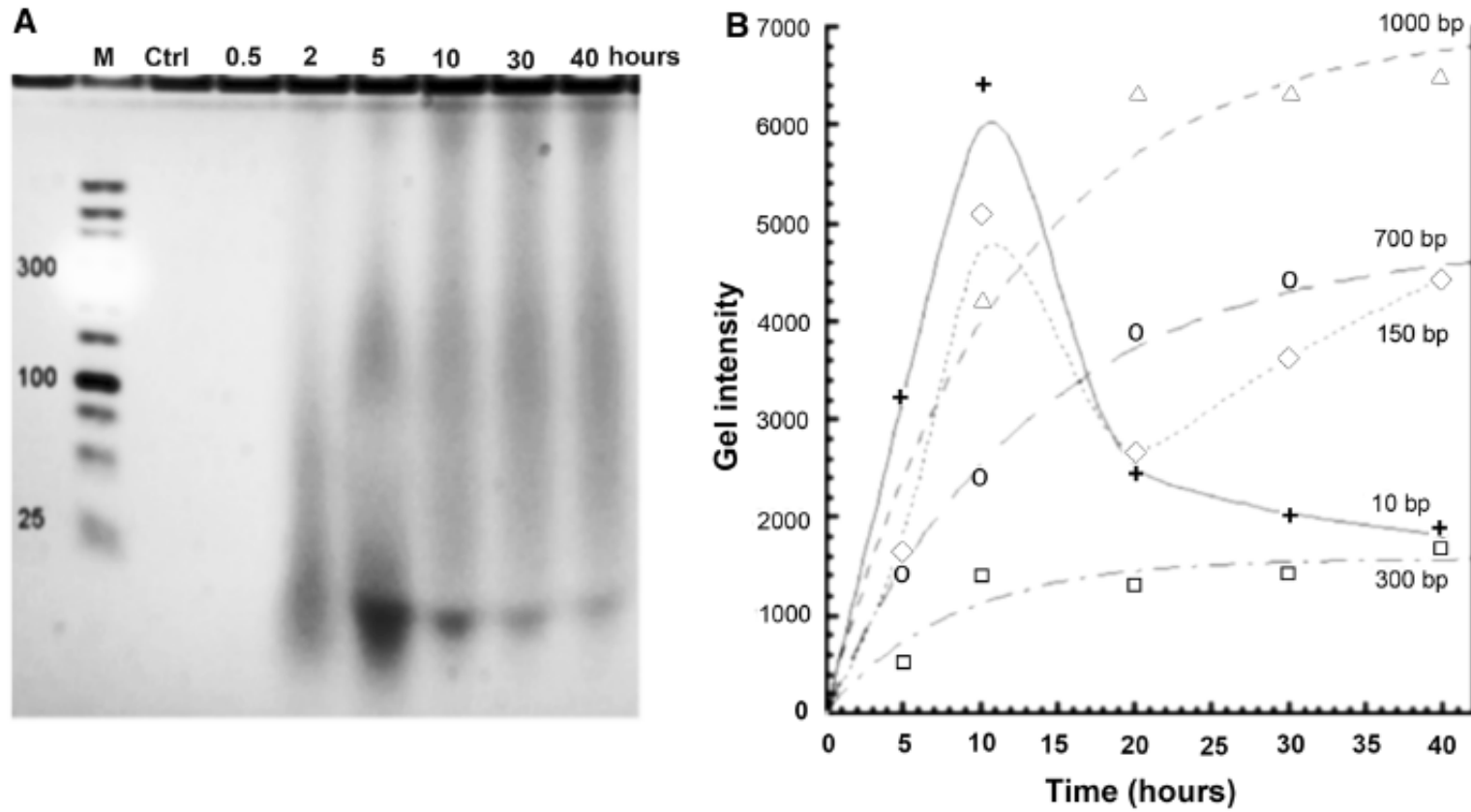
Topozini, Dies, Deamer, Rheinstadter – PLoS ONE 2013

Drying nucleotides in different environments  
 Low water content favours polymerization.  
 Molecules may align in an arrangement favourable to bond formation.

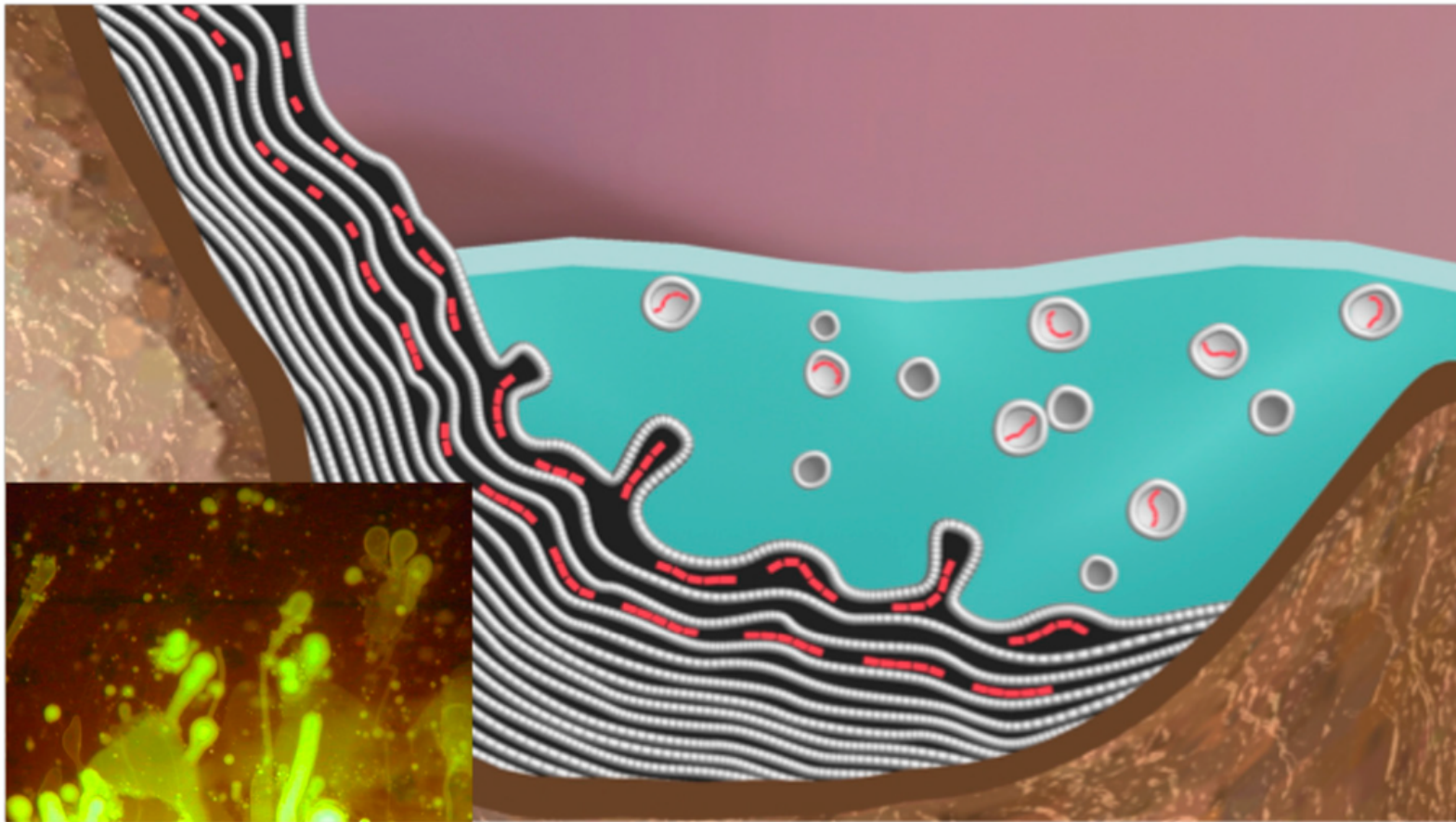


Himbert, Chapman, Deamer, Rheinstadter – 2016 submitted

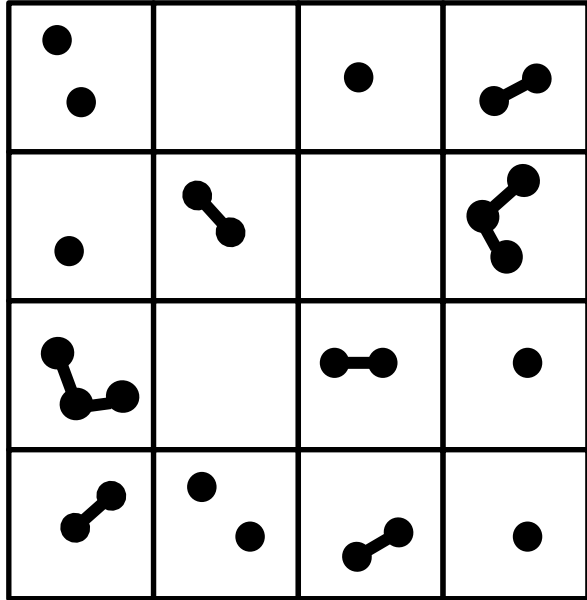
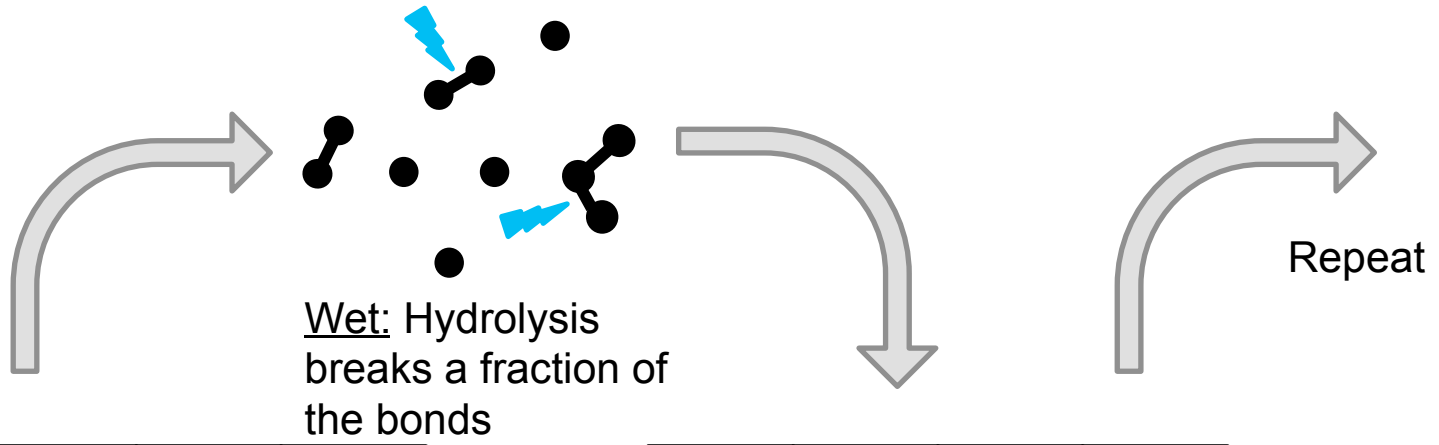
Wet-dry cycling in the presence of salt (e.g. ammonium chloride) generates a broad range of long strands



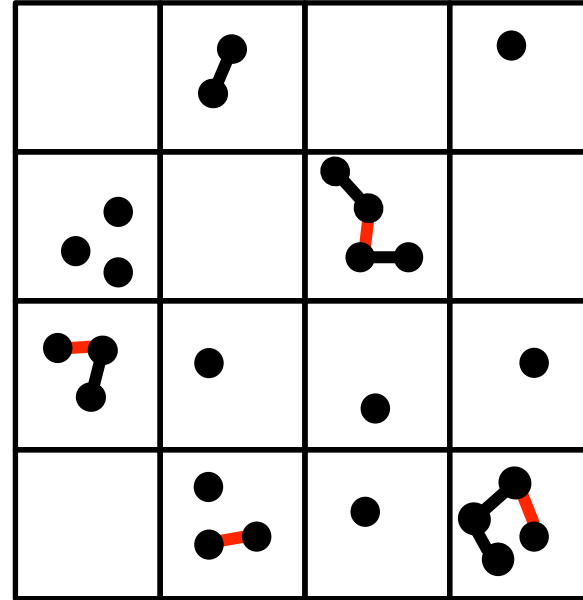
Da Silva, Maurel, Deamer – J. Mol. Evol. 2015



Bruce Damer and David Deamer envisage wetting and drying cycles leading to the Origin of Life in volcanic hydrothermal pools on land.



Dry: Bond formation is favoured if monomers are close



Dry: Repositioning of molecules allows **new** bond formation and creates longer polymers.

## Polymerization model – Reaction kinetics

$$\frac{dC_1}{dt} = -2k^+ C_1^2 + 2k^- C_2 - 2k^+ C_1 \sum_{n=2}^{L-1} C_n + 2k^- \sum_{n=3}^L C_n$$

$$\frac{dC_n}{dt} = k^+ \sum_{m=1}^{n-1} C_m C_{n-m} - (n-1)k^- C_n + 2k^- \sum_{m=n+1}^L C_m - 2k^+ C_n \sum_{m=1}^{L-n} C_m$$

Equilibrium length distribution is exponential. Can be solved exactly.

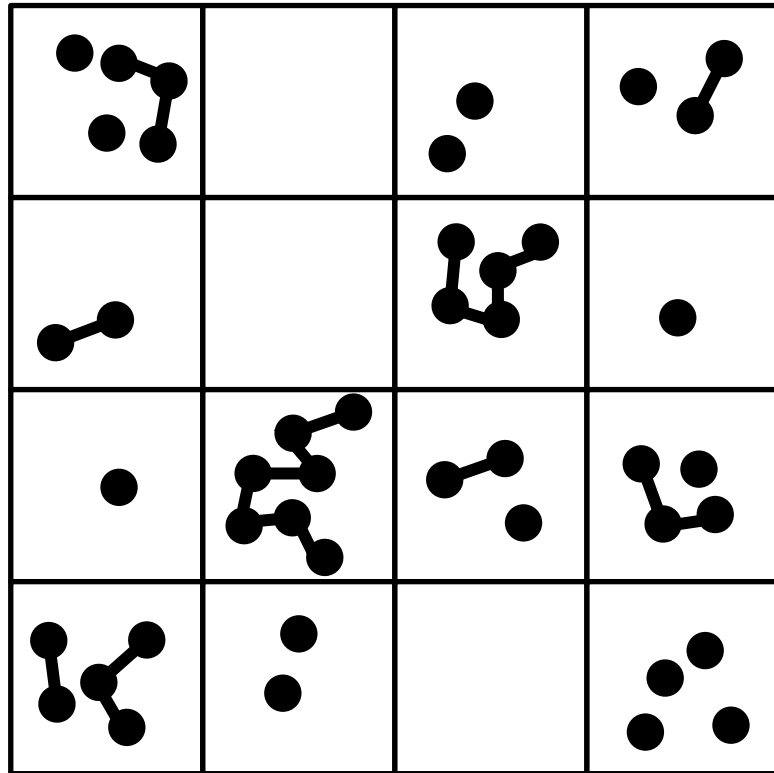
$$C_n = ax^{n-1}$$

$a$  and  $x$  are functions of the initial monomer concentration and the  
equil constant

$$K = k^+ / k^-$$

mean length is  $1/(1-x)$

# Polymer model



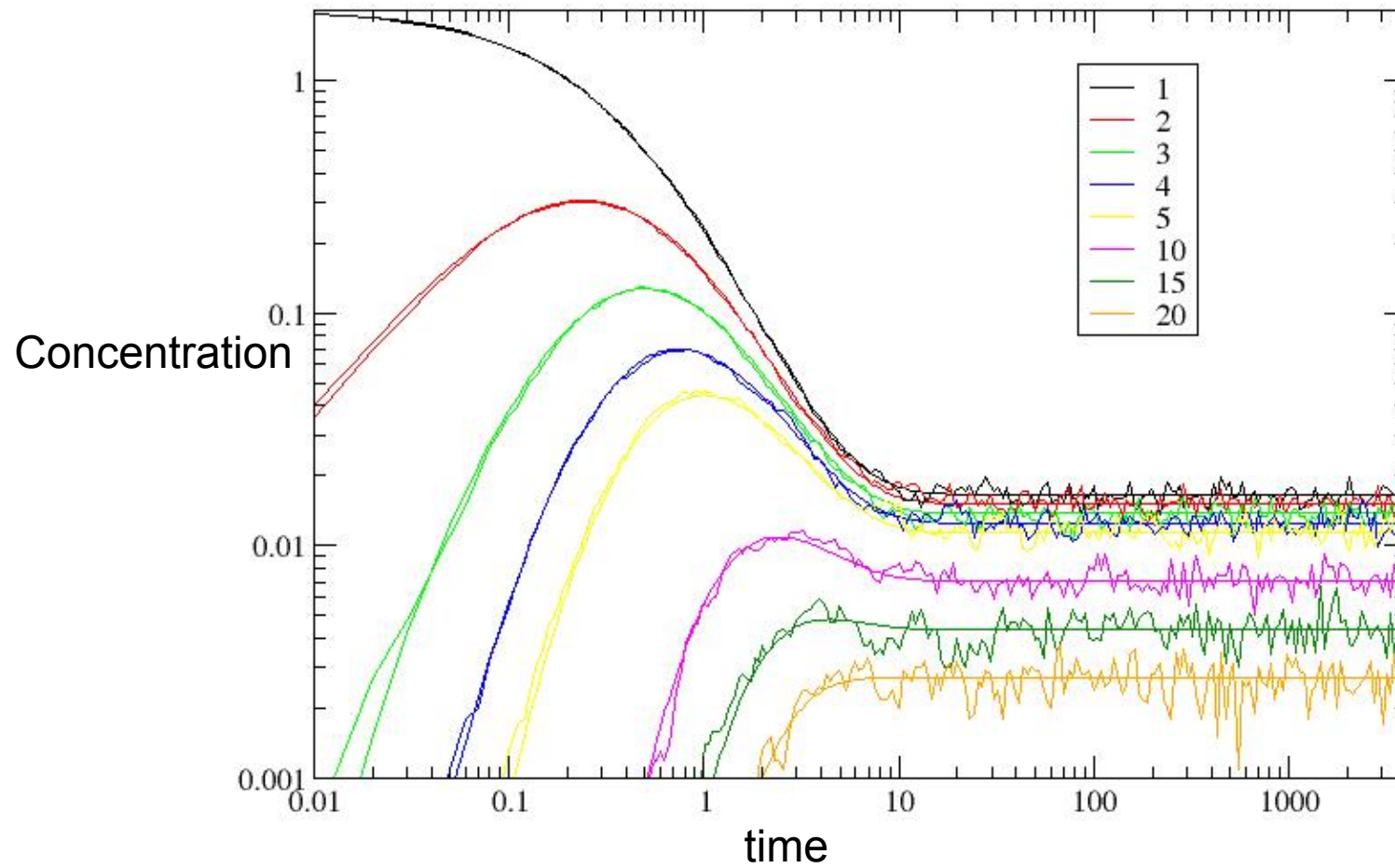
Can also solve this by Monte Carlo simulation

Any two strands on same site can bond at rate  $k^+$

A strand of length  $m$  hydrolyzes at rate  $(m-1)k^-$

Strands can hop to neighbouring sites at rate  $h$

Two methods are the same when diffusion is fast

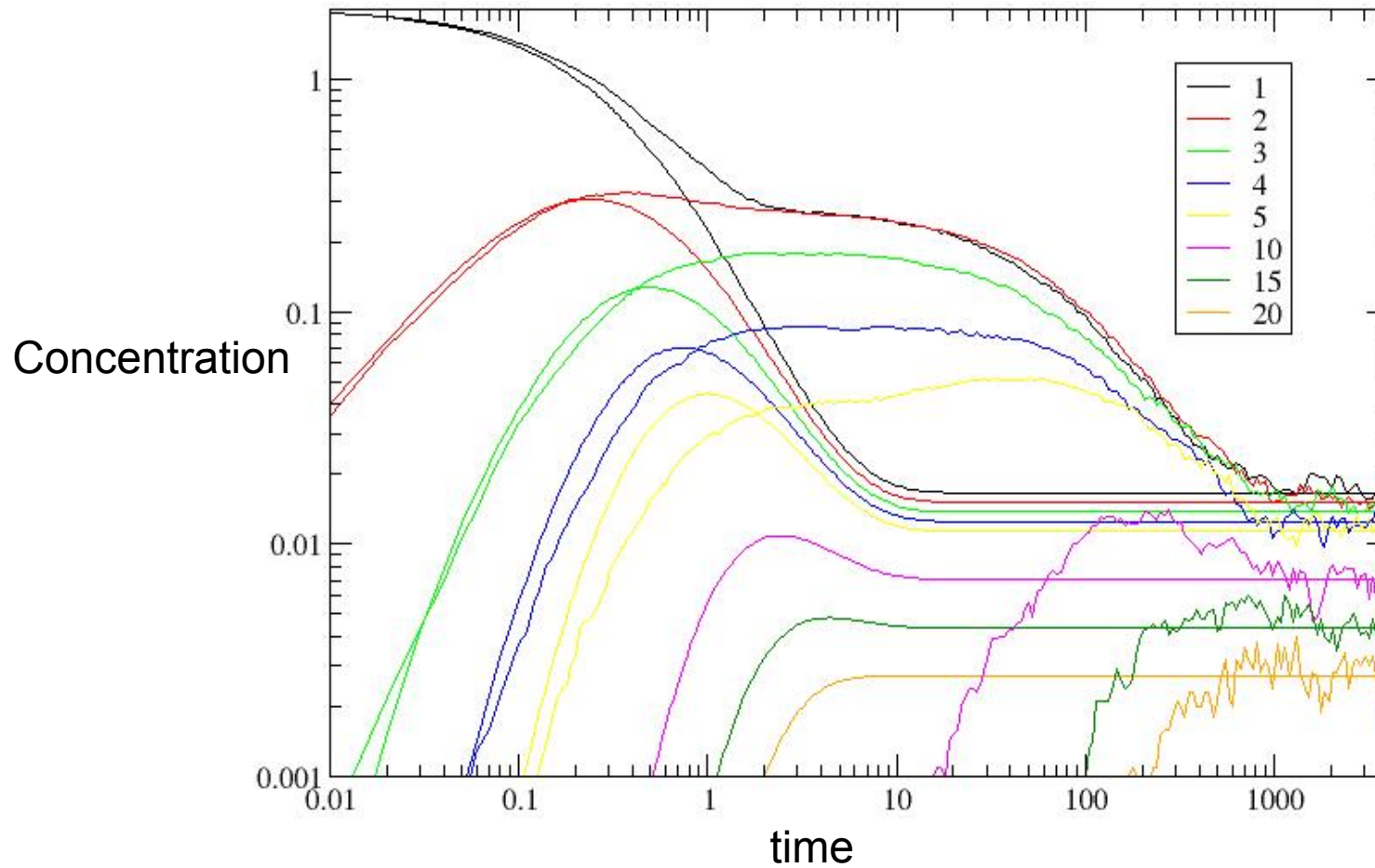


smooth = reaction kinetics

jagged = Monte Carlo with very fast diffusion



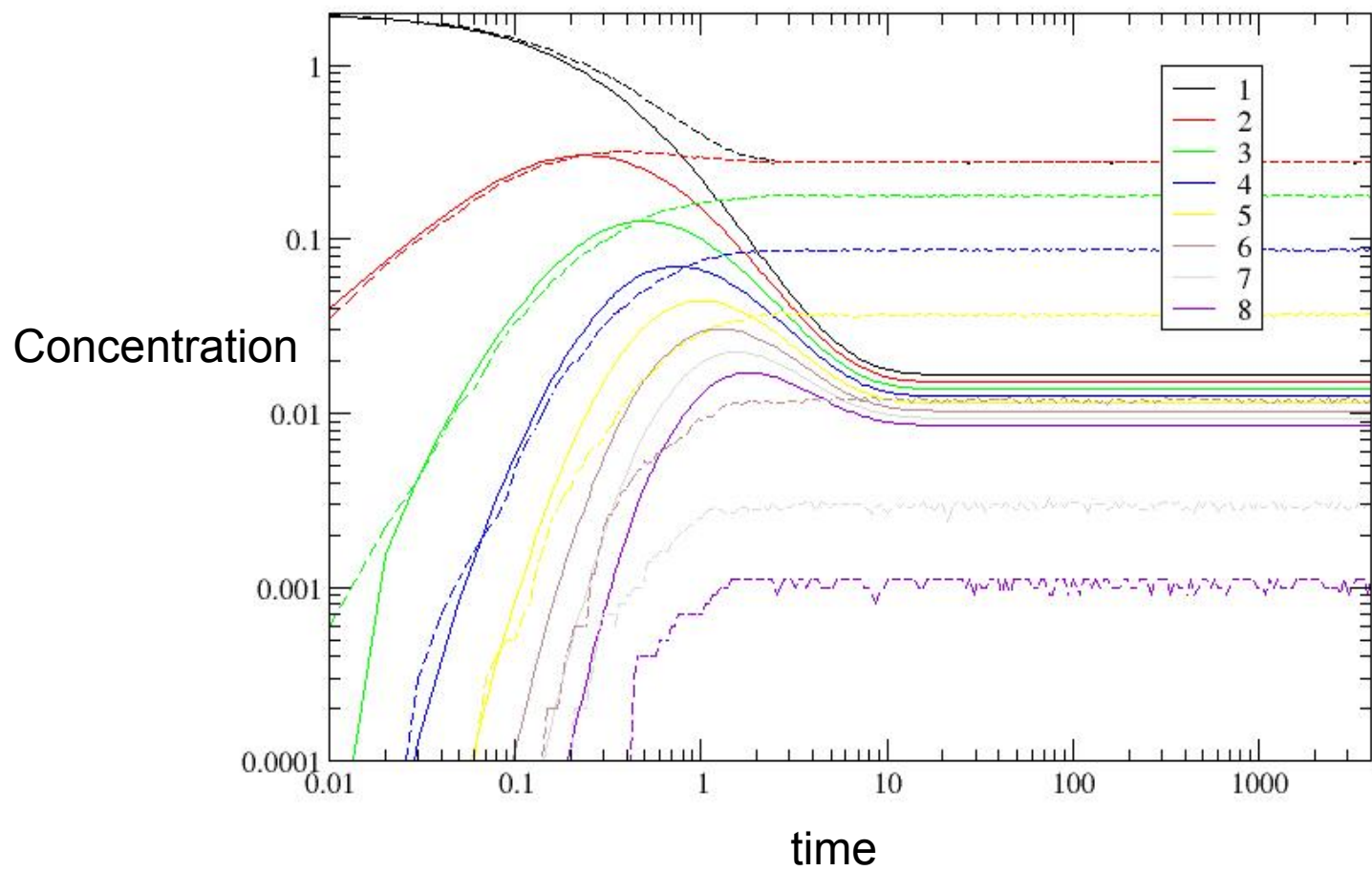
When diffusion is slow ( $h = 0.01$ ) the Monte Carlo simulation is much slower to reach equilibrium, but the same equilibrium is reached.



smooth = reaction kinetics

jagged = Monte Carlo with slow diffusion

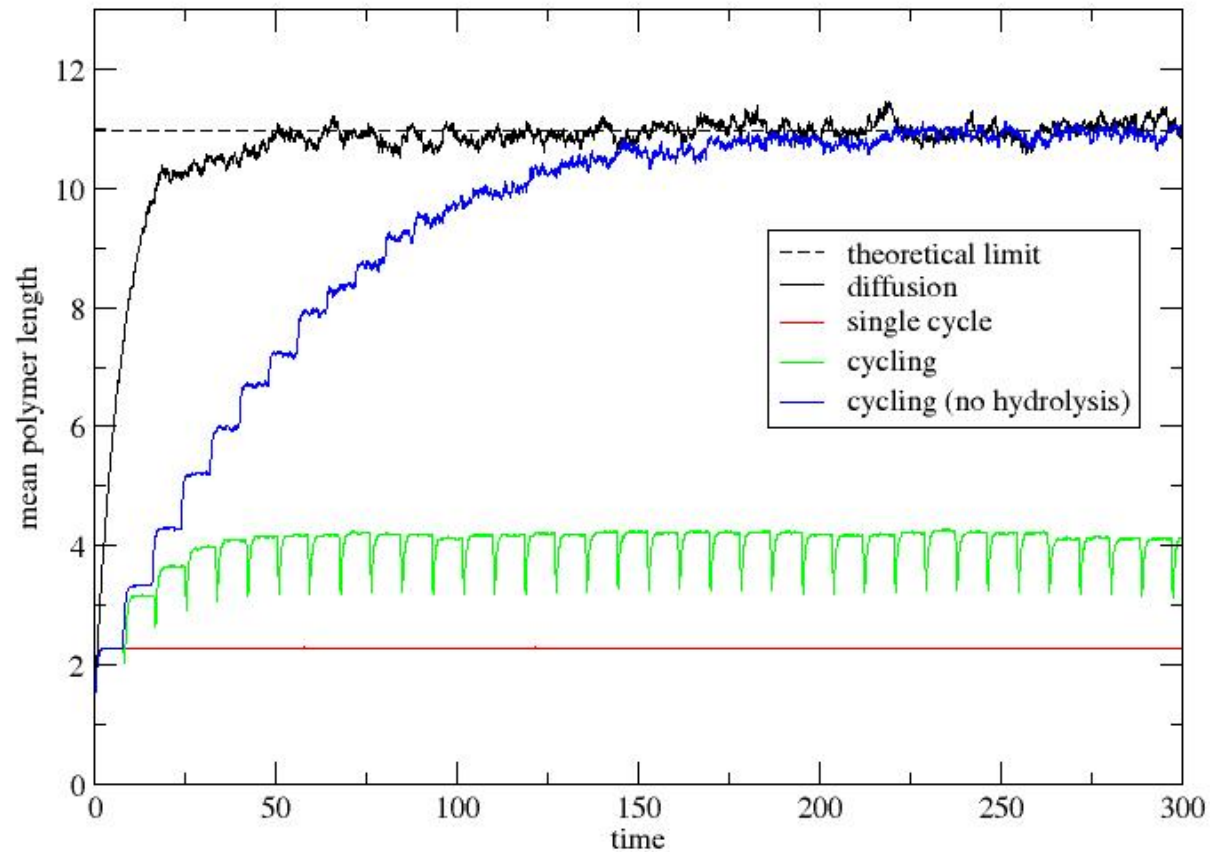
When diffusion is zero, the Monte Carlo simulation cannot reach the equilibrium distribution. Only monomers initially on the same site can react.



smooth = reaction kinetics

jagged = Monte Carlo with no diffusion

## The effects of wet-dry cycling on polymerization



- Dry phase gives long polymers when diffusion is possible
- Single dry phase without diffusion gives short mean length
- Cycling improves on single dry phase
- Cycling can approach the optimum case if  $wt_{wet} \ll 1$

# An Evolutionary Theory for the RNA World



Art: Julie Newdoll  
"A brush with science"

Emergence from Prebiotic Chemistry

What were the first kind of ribozymes – a Polymerase or something else?

Stochastic – small number of molecules

Cooperation at the molecular level

Parasitic templates

Spatial clustering

Was there a complex RNA metabolism?

Evolution of cells

Evolution of chromosomes (linkage)

# Prebiotic Chemistry

Precursors



Monomers



Random RNA strands



Ribozymes

rare  
stochastic  
not reproducible

Chemistry to make monomers and random strands must be reproducible:

- thermodynamics
- reaction kinetics

Does not need to be autocatalytic

Does not need to evolve

Not really a metabolism (?)

➔ Dead (non-living)

# RNA World

Precursors



Nucleotide  
synthetase

Monomers



Polymerase

Ribozymes

Making ribozymes is now  
reproducible because of sequence  
replication.

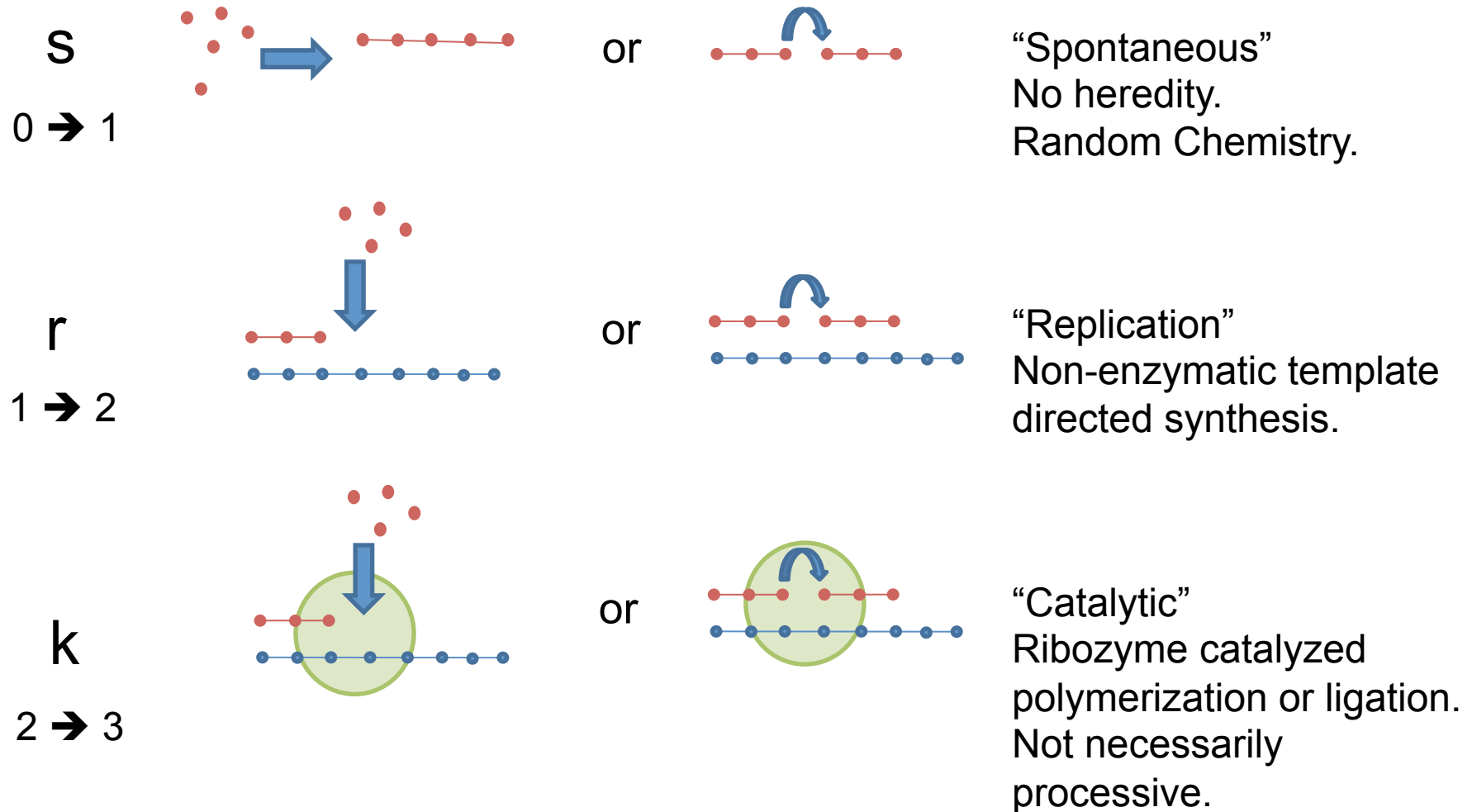
- heredity
- autocatalysis

➔ Living





“Life is a self-sustained chemical  
system capable of evolution”

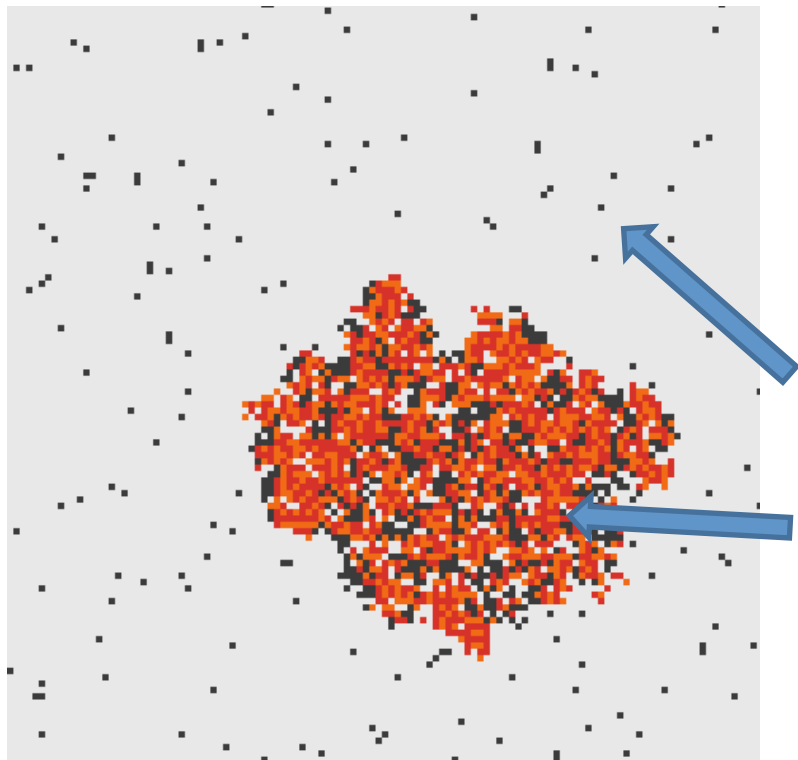
It counts as life if there are  
molecules that can only be  
made reproducibly because  
of replication/heredity

## Three ways of making an RNA strand: s, r and k reactions



## Spatial Model for Origin of Replication in the RNA World

-  0 = Empty site
-  1 = Polymerase
-  2 = Complementary to Polymerase
-  3 = Random strand or Mutant



The model in brief:

s reaction creates random strands

r reaction copies all strands

k reaction only performed by a polymerase.

Polymerases replicate strands on neighbouring sites.

New strands are placed on neighbouring vacancies.

Strands can hop to neighbouring sites.

Strands breakdown to monomers again.

Successful replication creates a complementary strand.

Mutations create parasites.

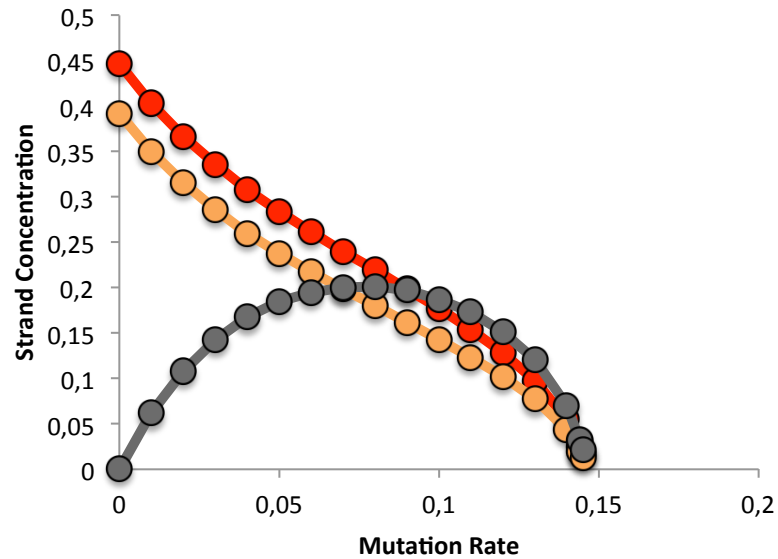
Dead state = low conc of random strands

Living state = high concentration of polymerases, complements, and mutants

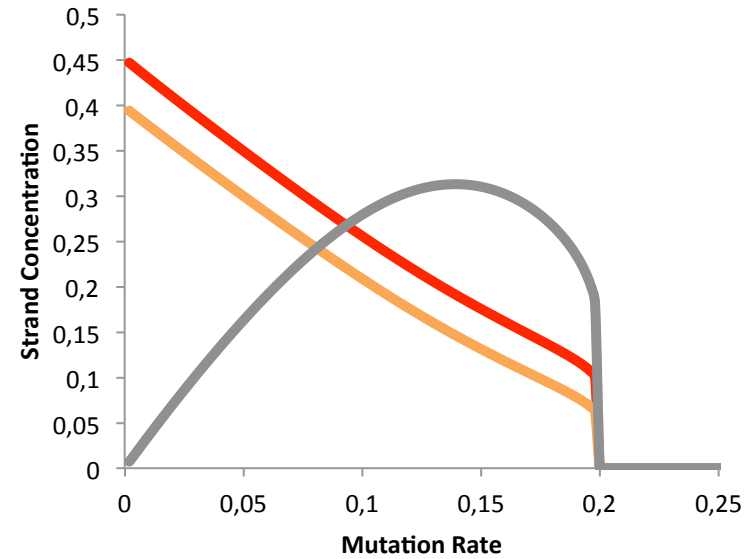
The living patch has grown from the introduction of a single polymerase



## Concentration as a function of mutation rate M



Simulations



Approximate theory based on paired site concentrations

The Error Threshold is the max mutation rate for which the system survives.

Spatial clustering allows for a fairly high error threshold.

## Polymerases first

The importance of cooperation.  
Spatial clustering allows survival against parasites

Shay, Huynh & Higgs (2015) J Theor Biol  
Higgs & Lehman (2015) Nature Rev Genetics

## Nucleotide synthases second








The synthase helps the polymerase by making more monomers.  
The polymerase helps the synthase by replicating it.

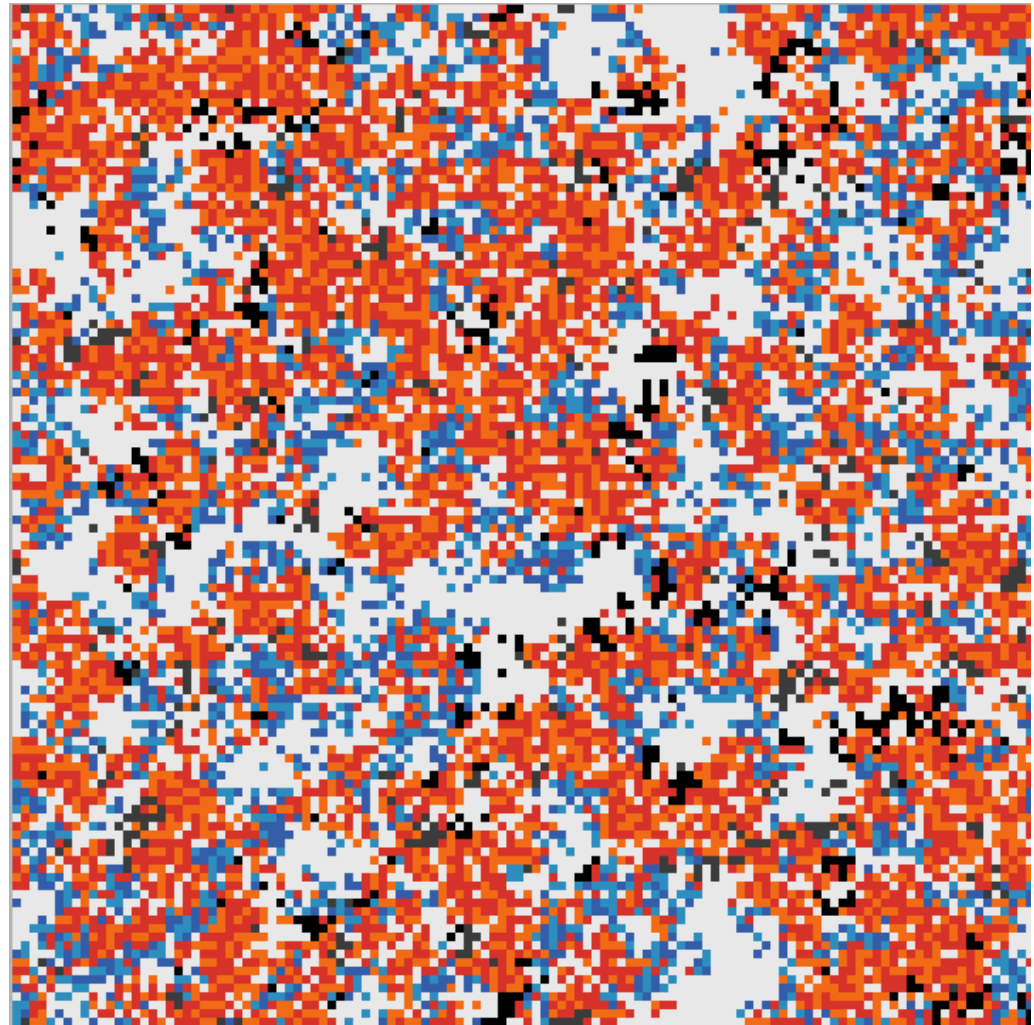
When can the two coexist?

Can we find conditions where the two can survive together but the polymerase could not survive on its own?

Current work – Ye Eun Kim, Andrew Tupper

## When can a Nucleotide Synthetase Cooperate with a Polymerase?

-  0 = Empty site
-  1 = Polymerase
-  2 = Complementary strand to Polymerase
-  3 = Mutant polymerase (non-functional)
-  4 = Synthetase
-  5 = Complementary strand to Synthetase
-  6 = Mutant synthetase (non-functional)

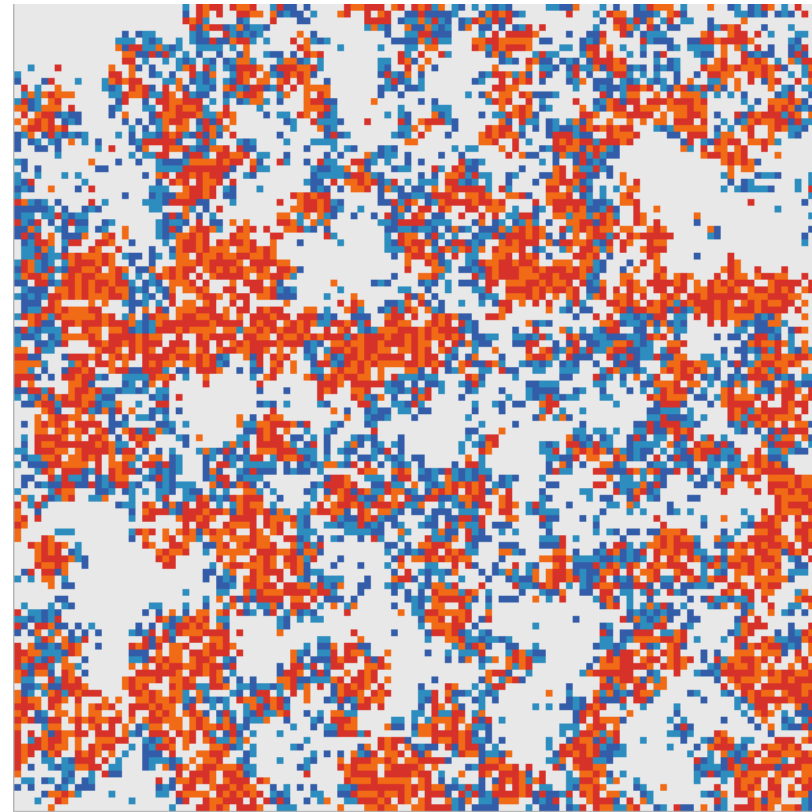


## Polymerase + Independent short strands

- 0 = Empty site
- 1 = Polymerase ( $L_1 = 100$ )
- 2 = Complementary to Polymerase ( $L_1 = 100$ )
- 4 = Short strand ( $L_2 = 50$ )
- 5 = Complementary short strand ( $L_2 = 50$ )

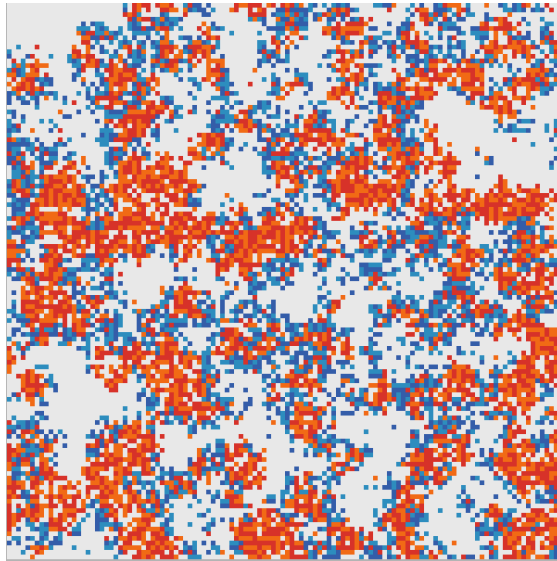
$$\text{replication rate} \rightarrow k = \frac{v_{pol} A}{L}$$

polymerization rate constant
monomer concentration
template length

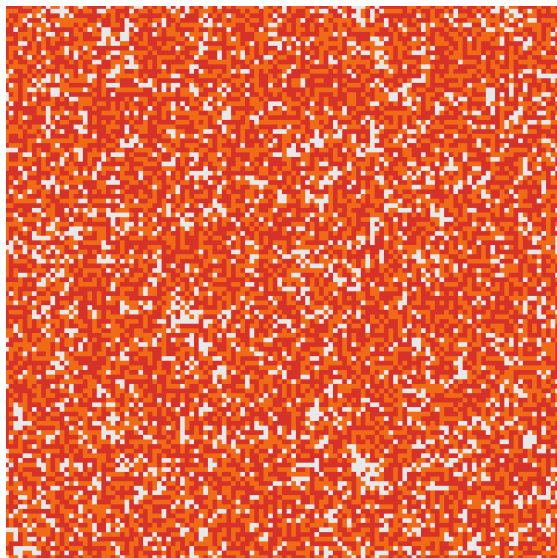
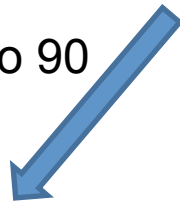


The short strands are non-functional parasites that multiply twice as fast as the polymerase. The polymerase still survives because of spatial clustering.

Polymerase +  
Independent short strands  
 $L_2 = 50$

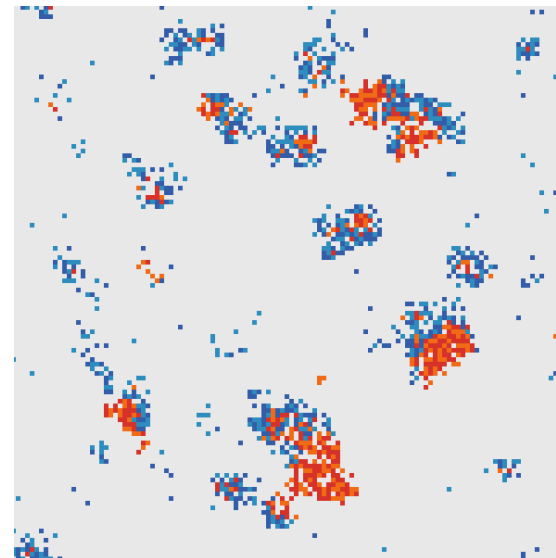
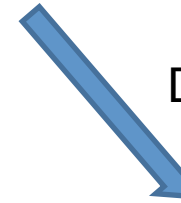


Increase  $L_2$  to 90



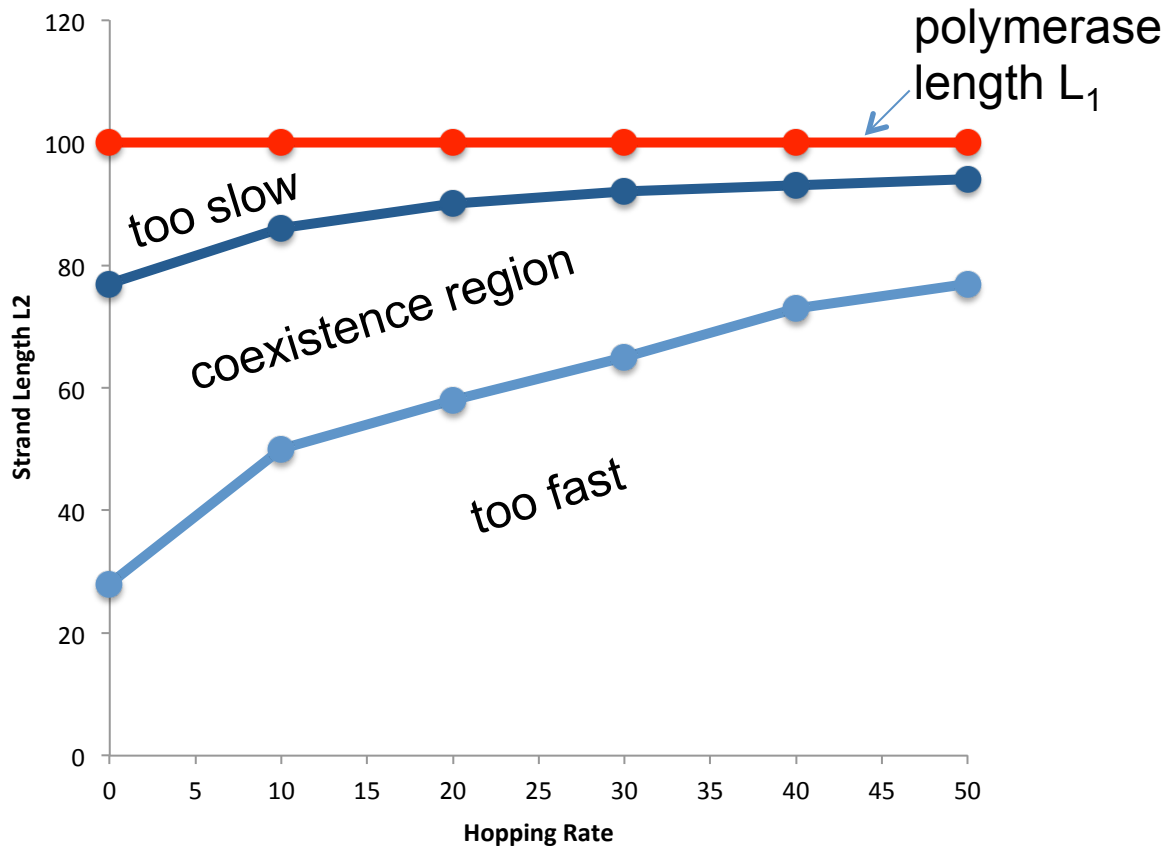
Blue strands multiply too slowly.  
Blue strands die and  
Polymerases survive alone

Decrease  $L_2$  to 30



Blue strands multiply too fast.  
Blue strands overrun the  
Polymerases. Everything dies.

Coexistence region between polymerases and independent short strands



In the well-mixed case, coexistence is only possible if the two lengths are exactly equal.

## Monomer synthesis and diffusion

Simple assumption that monomers diffuse rapidly everywhere does not work because synthetases are overrun by their own mutations.

*Need a finite rate of monomer diffusion*

$$\frac{\partial A}{\partial t} = a + bX_{syn} - A + D\nabla^2 A \quad - \text{Polymerization} + \text{Breakdown}$$

Polymerization - When a new strand is made,  $L$  monomers are used up from the site of the template.







Breakdown - When a strand is destroyed,  $L$  monomers are released to the site where the strand was.

Local concentration  $A_i$  on each site  $i$

$S_i = 1$  if there is a synthetase on site  $i$ , or 0 otherwise

$$\delta A_i = \delta t (a + bS_i - A_i - D(A_i - \frac{1}{8} \sum_j A_j)) \quad - \text{Polymerization} + \text{Breakdown}$$

## Polymerase + Synthetase + Local monomer diffusion

-  0 = Empty site
-  1 = Polymerase ( $L_1 = 100$ )
-  2 = Complementary to Polymerase ( $L_1 = 100$ )
-  4 = Synthetase ( $L_2 = 60$ )
-  5 = Complementary to Synthase ( $L_2 = 60$ )
-  6 = Mutant Synthetase ( $L_2 = 60$ )

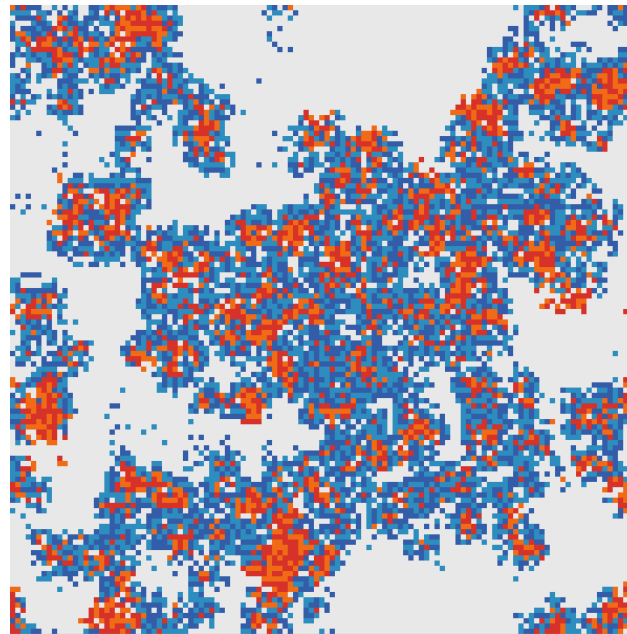
Monomer concentration  
controlled by diffusion

$$a = 0$$

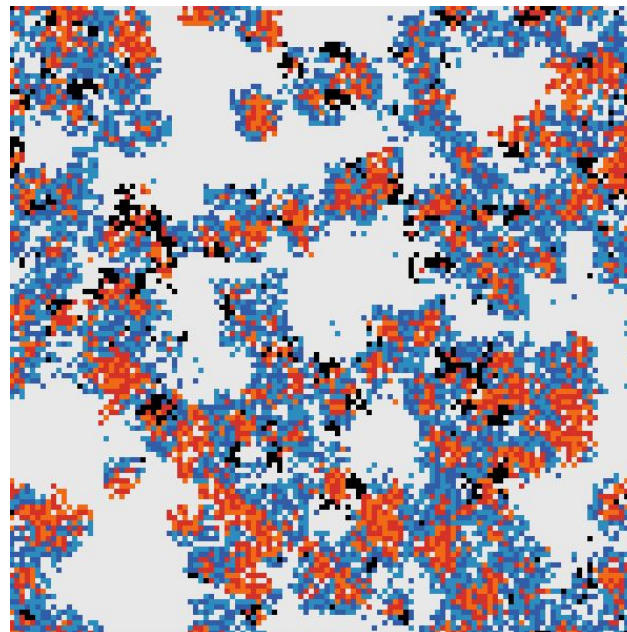
$$b = 5000$$

$$D = 30$$

$$v_{\text{pol}} = 5$$



Two ribozymes  
survive in cooperation  
even when  $a = 0$



Add mutation  $M_{\text{syn}} = 0.01$

System remains stable



Spatial clustering is sufficient to allow cooperation of two unlinked ribozymes without membrane compartments

Synthetases can coexist with Polymerases....

... but only if conditions are right.

- Strand diffusion must be slow enough to allow joint clusters to form
- Synthases must replicate sufficiently fast, otherwise they die out
- Synthases must not replicate too fast, otherwise they overrun the polymerases
- Monomer diffusion must be fast enough to benefit the polymerases
- Monomer diffusion must not be too fast, otherwise the synthases are overrun by their own mutations

## Is a multi-component ribozyme system possible?

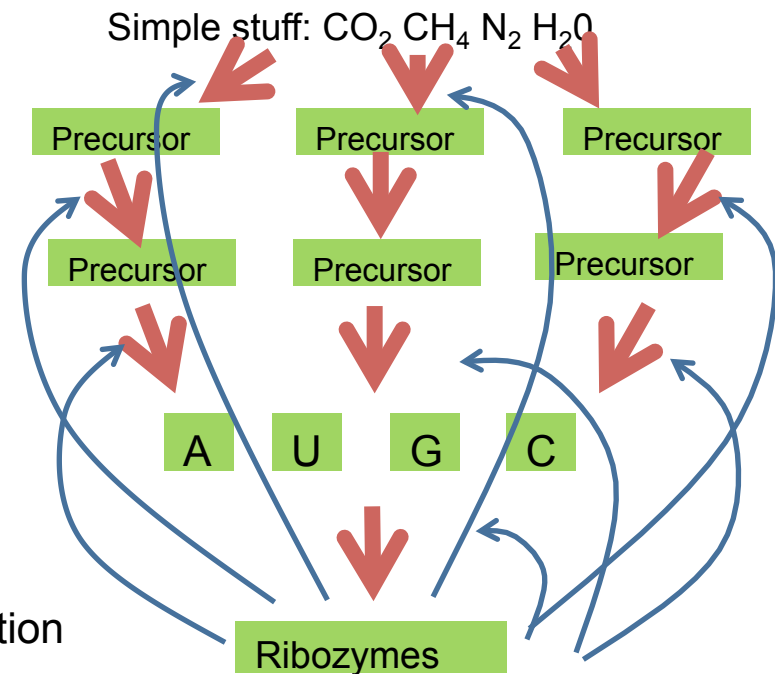
Multiple independent replicators in a surface based model will probably be difficult

*From surface-bound replicators to cells*

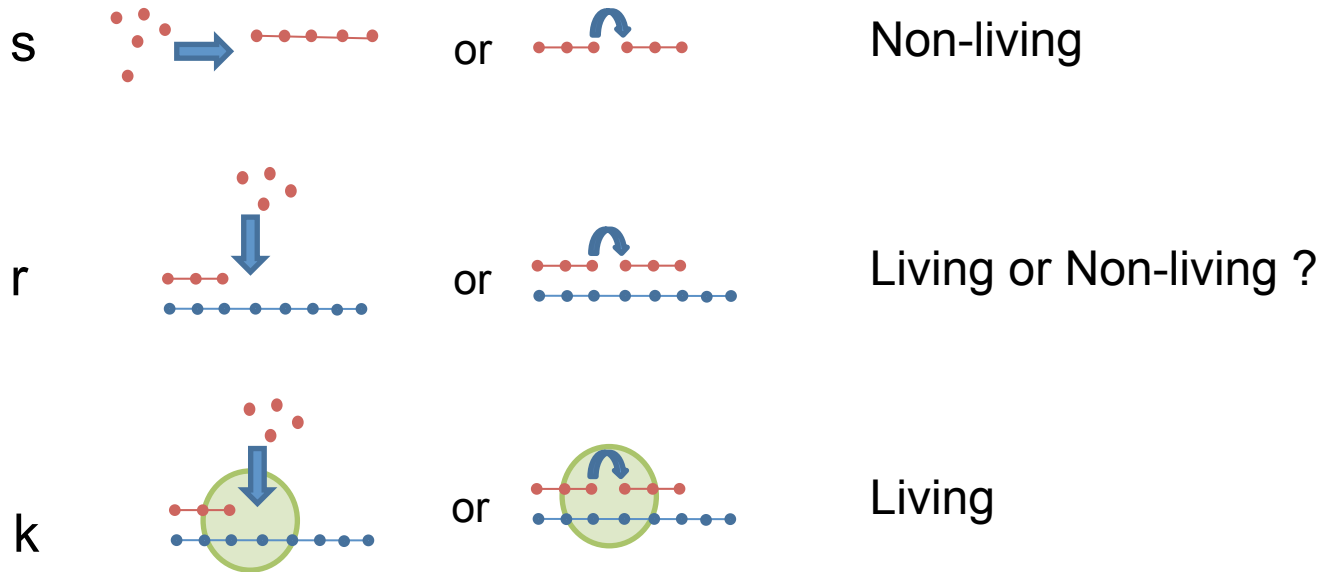
- + group selection
- + diffusion products kept within cell
- need to make lipids
- need to transport things across membranes
- need to control cell division

*From independent genes to chromosomes*

- + linked genes replicate together
- error threshold for long strands
- distinguish between replication and transcription



## The importance of r



The r reaction could support life if it was fast and accurate

- It has heredity
- Possibility of template evolution

If r is too slow, it cannot support life even if replication is 100% accurate.

If r is fast - maybe you don't need a polymerase at all ....

Another ribozyme could be first