

Complex organic molecules in low-mass protostars on solar system scales

I. Oxygen-bearing species

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ABSTRACT

Context. Complex organic molecules (COMs) are thought to form on icy dust grains in the earliest phase of star formation. The evolution of these COMs from the youngest Class 0/I protostellar phases toward the more evolved Class II phase is still not fully understood. Since planet formation seems to start early, and mature disks are too cold for characteristic COM emission lines, studying the inventory of COMs on solar system scales in the Class 0/I stage is relevant.

Aims. The aim is to determine the abundance ratios of oxygen-bearing COMs in Class 0 protostellar systems on scales of ~ 100 AU radius. These abundances will be inter-compared, and contrasted to other low-mass protostars such as IRAS16293-2422B and HH 212. Additionally, using both cold and hot COM lines, the gas-phase abundances can be tracked from a cold to a hot component, and ultimately be compared with those in ices to be measured with the *James Webb Space Telescope* (JWST). The abundance of deuterated methanol allows us to probe the ambient temperature during the formation of this species.

Methods. ALMA Band 3 (3 mm) and Band 6 (1 mm) observations are obtained of seven Class 0 protostars in the Perseus and Serpens star-forming regions. By modelling the inner protostellar region using 'LTE' models, the excitation temperature and column densities are determined for several O-bearing COMs including methanol (CH_3OH), acetaldehyde (CH_3CHO), methyl formate (CH_3OCHO), and dimethyl ether (CH_3OCH_3). Abundance ratios are taken with respect to CH_3OH .

Results. B1-c and B1-bS (both Perseus) and Serpens S68N (Serpens) show COM emission, i.e., three out of the seven of the observed sources. No clear correlation seems to exist between the occurrence of COMs and source luminosity. The abundances of several COMs such as CH_3OCHO , CH_3OCH_3 , acetone (CH_3COCH_3), and ethylene glycol ($(\text{CH}_2\text{OH})_2$) are remarkably similar for the three COM-rich sources, and to IRAS 16238-2422B and HH 212, even though these sources originate from four different star-forming regions (i.e., Perseus, Serpens, Ophiuchus, Orion). For other COMs like CH_3CHO , ethanol ($\text{CH}_3\text{CH}_2\text{OH}$), and glycolaldehyde (CH_2OHCHO), the abundances differ by up to an order of magnitude, indicating that local source conditions are case determining. B1-c hosts a cold ($T_{\text{ex}} \approx 60$ K), more extended component of COM emission with a column density of typically a few % of the warm/hot ($T_{\text{ex}} \sim 200$ K), central component. A D/H ratio of 1–3 % is derived for B1-c, S68N, and B1-bS based on the $\text{CH}_2\text{DOH}/\text{CH}_3\text{OH}$ ratio (taking into account statistical weighting) suggesting a temperature of ~ 15 K during the formation of methanol. This ratio is consistent with other low-mass protostars, but lower than for high-mass star-forming regions.

Conclusions. The abundance ratios of most O-bearing COMs are roughly fixed between different star-forming regions, and are presumably set at an earlier cold prestellar phase. For several COMs, local source properties become important. Future mid-infrared facilities such as JWST/MIRI will be essential to directly observe COM ices. Combining this with a larger sample of COM-rich sources with ALMA will allow for directly linking ice and gas-phase abundances in order to constrain the routes that produce and maintain chemical complexity during the star formation process.

Key words. Astrochemistry – stars: formation – stars: protostars – stars: low-mass – ISM: abundances – techniques: interferometric

1. Introduction

Complex organic molecules (COMs) are molecules with six or more atoms of which at least one atom is carbon (see review by Herbst & van Dishoeck 2009). They have been observed toward

high-mass and low-mass star-forming regions, both in young protostars (e.g., Caselli & Ceccarelli 2012; Jørgensen et al. 2016; McGuire 2018; Bøgelund et al. 2019), and in protostellar outflows (e.g., Arce et al. 2008; Öberg et al. 2010). As a protostellar

system develops from the embedded Class 0/I phase toward the more evolved Class II phase, the COMs may eventually either be destroyed or become incorporated in the ice mantles of dust grains or planetesimals in disks (Visser et al. 2009, 2011; Drozdovskaya et al. 2016). Since planet formation may start already in the embedded Class 0/I phase (e.g., Harsono et al. 2018; Manara et al. 2018), studying the inventory and abundances of COMs on solar system scales in the earliest stages of star formation is essential to probe the initial conditions of planet formation. With the launch of the *James Webb Space Telescope* (JWST) in the near future, the ice abundances of COMs can be directly linked to their gaseous counterpart.

In cold prestellar envelopes, formation of COMs happens on the ice mantles of dust grains through hydrogenation of smaller molecules such as CO (Charnley et al. 1992; Watanabe & Kouchi 2002; Fuchs et al. 2009). This leads to the formation of methanol (CH₃OH), but these COMs may even get as complex as, for example, glycolaldehyde (CH₂OHCHO), ethylene glycol ((CH₂OH)₂), and glycerol (HOCH₂CH(OH)CH₂OH) (Öberg et al. 2009; Chuang et al. 2016, 2017; Fedoseev et al. 2017). When the dust grains move inwards toward protostars, the ultraviolet (UV) radiation field increases (e.g., Visser et al. 2009; Drozdovskaya et al. 2016), leading to the dissociation of some COMs into smaller radicals (Gerakines et al. 1996; Öberg 2016). These radicals react with other COMs in the ice mantles to form larger species (Garrod & Herbst 2006; Garrod et al. 2008; Öberg et al. 2009), and at higher temperatures perhaps through thermal induced chemistry (Theulé et al. 2013). When the temperature of the dust grains reaches $T \approx 100 - 300$ K, all COMs desorb from the grains into the gas phase where they can be transformed by high temperature gas-phase chemistry and UV radiation (e.g., Charnley et al. 1992; Balucani et al. 2015; Skouteris et al. 2018). A fraction of the COMs may be incorporated into larger bodies in the cold mid-plane of an accretion disk.

Another reason for studying COMs in the earliest (Class 0) phase of star formation is that gaseous COM lines are hard to detect in the Class I phase (e.g., Artur de la Villarmois et al. 2019). An exception is when a (more evolved) source undergoes an accretion burst; the sudden increase in luminosity can release frozen COMs from the grains (van 't Hoff et al. 2018; Lee et al. 2019b). In the ice phase, they are difficult to observe, with only well constrained abundances for CH₃OH and some limits or tentative detections for formic acid (HCOOH), acetaldehyde (CH₃CHO), and ethanol (CH₃CH₂OH) (see review by Boogert et al. 2015; Terwisscha van Scheltinga et al. 2018).

Most studies of gaseous COMs have been carried out on high-mass hot cores in massive star-forming regions such as Sagittarius B2 and Orion KL, where COMs are typically present in high abundances (e.g., Belloche et al. 2013; Neill et al. 2014; Crockett et al. 2014; Pagani et al. 2017). In low-mass young stellar objects, single-dish observations with beams of a few thousand AU show that the abundances of oxygen-bearing COMs with respect to CH₃OH seem to be narrowly distributed (Bottinelli et al. 2004; Jørgensen et al. 2005; Bergner et al. 2017; Belloche et al. 2020).

With the Atacama Large Millimeter/submillimeter Array (ALMA), it is now possible to spatially resolve individual protostellar systems on scales of ~ 50 AU. Additionally, the spectral coverage and sensitivity of ALMA allow a large variety of COMs and corresponding isotopologues to be observed in optically thin lines, resulting in more accurate abundance determinations. Gaseous COMs have been studied mostly in the Class 0 IRAS 16293-2422 binary system (hereafter IRAS 16293) as part of the ALMA PILS survey (Jørgensen et al. 2016). This led

to many detections of new species in the interstellar medium (ISM) such as HONO (Coutens et al. 2019), CH₃Cl (Fayolle et al. 2017), and CHD₂OCHO (Manigand et al. 2019). Furthermore, IRAS 16293 has been studied extensively for both oxygen and nitrogen-bearing COMs (Jørgensen et al. 2018; Calcutt et al. 2018; Ligterink et al. 2018a,b; Manigand et al. 2020). Other low-mass sources studied interferometrically on $\sim 50 - 150$ AU scales include HH 212 (Lee et al. 2019a), NGC 1333-IRAS 2A and IRAS 4A (hereafter IRAS 2A and IRAS 4A; Taquet et al. 2015, 2019), L483 (Jacobsen et al. 2019), and three sources in the Serpens Main star forming region (Bergner et al. 2019).

Single dish observations show hints of a cold, more extended component of COM emission in Class 0 protostars where the excitation temperature may drop below 20–60 K. This component is present additional to the 'typical' warm/hot, more compact COM emission with $T \gtrsim 100 - 300$ K (Bisschop et al. 2007; Isokoski et al. 2013; Marcelino et al. 2018). Currently, the chemical relationship between the cold COMs on large scales and warm/hot COMs on small scales is still unknown. The evolution of COMs from a cold extended component to a hot central component can be traced by observing COMs in multiple frequency bands.

The Perseus and Serpens star-forming regions are known to host many embedded protostars (e.g., Enoch et al. 2009; Tobin et al. 2016; Karska et al. 2018). In this paper, four protostellar regions with strong ice features in infrared observations (Boogert et al. 2008) are targeted with ALMA in Band 3 and Band 6. These targets are also part of the JWST/MIRI guaranteed time observations (GTO) program (project ID 1290). JWST/MIRI will, for instance, observe the 7.2 μ m and 7.4 μ m features for which CH₃CH₂OH and CH₃CHO, respectively, are strong candidates based on laboratory spectroscopy (e.g., Schutte et al. 1999; Terwisscha van Scheltinga et al. 2018). Ultimately, this will allow for directly comparing gaseous and icy COMs. Some additional nearby sources falling within the ALMA field of view are also included here, leading to a total of seven sources: B1-b, B1-bS, B1-bN and B1-c in the Perseus Barnard 1 cloud, and Serpens S68N (hereafter S68N), Ser-emb 8 (N), and Serpens SMM3 (hereafter SMM3) in the Serpens Main region.

B1-b (sometimes also referred to as B1-bW) is a bright infrared source with strong ice features in CH₃OH and (potentially) CH₃CHO (Öberg et al. 2011). B1-bN is a suggested first hydrostatic core candidate (e.g., Pezzuto et al. 2012; Hirano & Liu 2014; Gerin et al. 2017), and B1-bS is proposed as a hot corino (Marcelino et al. 2018). B1-c is a young Class 0 object showing both strong CH₃OH ice and gas COM features in the hot corino (Boogert et al. 2008; Bergner et al. 2017), and a high velocity outflow (Jørgensen et al. 2006; Hatchell et al. 2007). Moreover, all Barnard 1 objects have been extensively studied in the VLA Nascent Disk and Multiplicity (VANDAM) survey (Tobin et al. 2016; Tychoniec et al. 2018b), showing compact disk-like structures at scales of several tens of AU. In Serpens, S68N and SMM3 are both Class 0 sources studied in detail by Kristensen et al. (2010). Hull et al. (2017) and Tychoniec et al. (2018a, 2019) observed several protostars in Serpens with ALMA, focusing on outflows. Interestingly, whereas Ser-emb 8 (N) hosts a high velocity outflow, S68N only has a slow velocity outflow. The characteristics of our targets are shown in Table 1. We focus here on the O-bearing COMs in the data; the N-bearing species will be part of a future paper.

Table 1. List of protostars discussed in this paper as well as their main astronomical properties.

Object	Cloud	RA (J2000) (hh:mm:ss.ss)	Dec (J2000) (dd:mm:ss.s)	d pc	L_{bol} L_{\odot}	T_{bol} K	M_{env} M_{\odot}	$\text{CH}_3\text{OH}/\text{H}_2\text{O}_{\text{ice}}$ (%)	COMs detected?	
									Band 3	Band 6
B1-b	Perseus	03:33:20.34	31:07:21.3	320 ¹	0.32 ²	151 ²	1.5 ²	11.2 ³	n	n
<i>B1-bN</i>	Perseus	03:33:21.21	31:07:43.6	320 ¹	0.28 ⁴	17 ⁴	0.4 ⁴	–	n	–
<i>B1-bS</i>	Perseus	03:33:21.36	31:07:26.3	320 ¹	0.57 ⁴	22 ⁴	0.4 ⁴	–	y	y
B1-c	Perseus	03:33:17.88	31:09:31.8	320 ¹	5.9 ⁵	46 ⁵	3.8 ²	<7.1 ³	y	y
S68N	Serpens	18:29:48.09	01:16:43.3	436 ⁶	5.4 ⁷	58 ⁶	9.4 ⁷	–	y	y
<i>Ser-emb 8 (N)</i>	Serpens	18:29:48.73	01:16:55.6	436 ⁶	–	–	–	–	n	y ⁸
SMM3	Serpens	18:29:57.75	01:14:06.7	436 ⁶	27.5 ⁵	37 ⁵	–	–	n	n ⁹

Notes. The objects in italic additionally fell within the field of view of the ALMA observations. In the last 2 columns, yes (y) indicates that COMs were detected in this work, no (n) indicates that the spectra lacked any COM detections, and – indicates that the Band was not observed for the corresponding target.

⁽¹⁾ Ortiz-León et al. (2018). ⁽²⁾ Enoch et al. (2009). ⁽³⁾ Boogert et al. (2008). ⁽⁴⁾ Hirano & Liu (2014). ⁽⁵⁾ Karska et al. (2018). ⁽⁶⁾ Ortiz-León et al. (2017). ⁽⁷⁾ Enoch et al. (2011). ⁽⁸⁾ Only CH₃OH emission related to the outflow detected. ⁽⁹⁾ Based on data from ALMA program 2017.1.01350.S

2. Observations

ALMA Cycle 5 observations (project 2017.1.01174.S; PI: E.F. van Dishoeck) were taken in Band 3 (3 mm) and Band 6 (1 mm). The key targeted COMs were CH₃OH, HCOOH, CH₃CHO, CH₃OCHO, CH₃CH₂OH, and NH₂CHO. We list the main technical properties of our ALMA images in Table E.1. For B1-bN and SMM3, we lack Band 6 observations in our program. All data were taken using the 12m array, with a synthesized beam of ~1.5–2.5" (~500–1000 AU at the distances of our sources; C43-2 & C43-3 configuration) and ~0.45" (~100–200 AU; C43-4 configuration) for Band 3 and Band 6, respectively. The largest recoverable scales (LAS) were ~20" (~6000–9000 AU) and ~6" (~2000–3000 AU) for Band 3 and Band 6, respectively. The spectral resolution was ~0.2 km s⁻¹ for most spectral windows, with a few windows in Band 3 having ~0.3–0.4 km s⁻¹. At this resolution, all lines are spectrally resolved. The line sensitivity is ~0.1 K. The absolute uncertainty on the flux calibration is ≤15% for all observations. For SMM3, we use archival Band 6 data (project 2017.1.01350.S; PI: Ł. Tychoniec) at similar spatial resolution (~0.3") but lower spectral resolution (0.7 km s⁻¹).

The images were primary beam corrected, with factors of ~1 for sources in the center of the images (i.e., B1-b, B1-c, S68N, SMM3), and ~3 and ~4 for B1-bS and Ser-emb 8 (N) in Band 6, respectively. In Band 3 the primary beam correction for B1-bS, B1-bN, and Ser-emb 8 (N) was ~1.2, ~2, and ~1.3, respectively.

Continuum images were made using line-free channels in all spectral windows. For the line-dense sources (e.g., B1-c) the line-free channels were carefully selected to exclude any line emission from the continuum images. Using the continuum solutions, all line data were continuum subtracted. From the pipeline product data, it was clear that all COM emission (except CH₃OH) was unresolved and centrally peaked on all sources. Therefore, line images were made using a mask size of a few times the size of the synthesized beam with the CASA 5.1.1 *tlean* task using a Briggs weighting of 0.5. Figure 1 shows the continuum overlaid on the spatial distribution of CH₃OH 2_{1,1}–1_{0,1} and CH₃OCHO 21_{7,14}–20_{7,13} emission in Band 6 from B1-c, S68N, and B1-bS. Figure 2 shows the emission of CH₃OCH₃ 7_{0,7}–6_{1,6} in Band 3. Spectra were extracted from the central beam of the sources.

3. Spectral modelling and results

In the Band 3 and Band 6 spectra of B1-b, B1-bN, Ser-emb 8 (N), and SMM3, COM emission features are absent at our detection sensitivity (see Table E.1), except for spatially extended CH₃OH emission in Ser-emb 8 (N) which is related to the outflow (Tychoniec et al. 2018a, 2019). These sources are therefore excluded in the further analysis and labelled as COM-poor sources. The other sources, B1-c, S68N, and B1-bS, are labelled as COM-rich sources, and analyzed further below.

3.1. Methodology

The spectral analysis tool CASSIS¹ was used to determine column densities and excitation temperatures of detected species. The line lists of each molecule were taken from the CDMS (Müller et al. 2001, 2005; Endres et al. 2016)² and JPL (Pickett et al. 1998)³ catalogs. The laboratory spectroscopy of molecules discussed in this paper is introduced in Appendix A and a full list of lines can be found in Tables E.2 and E.3 for Band 3 and Band 6, respectively. On average, the transition upper energies of the spectral lines are lower in Band 3 compared to Band 6. The Band 3 and Band 6 data are modelled separately. For a fixed V_{lsr} , a grid of column densities, excitation temperatures, and full width half maximum (FWHM) of the line was set with steps of 0.05 on a logarithmic scale, and 10 K and 0.1 km s⁻¹ on a linear scale, respectively. The total parameter space probed is 10¹³–10¹⁷ cm⁻², 50–550 K, and 0.5–9.0 km s⁻¹, respectively; for CH₂DOH the explored parameter space of the column density is 10¹⁴–5 × 10¹⁷ cm⁻². For species with a small number of transitions in Band 6, the excitation temperature was fixed to $T_{\text{ex}} = 200$ K, and only the column density and FWHM were varied. For most species in Band 3, we had to assume that T_{ex} is equal to the Band 6 temperature because the signal-to-noise ratio (S/N) is too low to detect multiple spectral lines. Since the FWHM of the lines in B1-bS was close to the spectral resolution limit, we fixed the FWHM to 1 km s⁻¹.

¹ <http://cassis.irap.omp.eu/>

² <https://cdms.astro.uni-koeln.de/>

³ <https://spec.jpl.nasa.gov/>

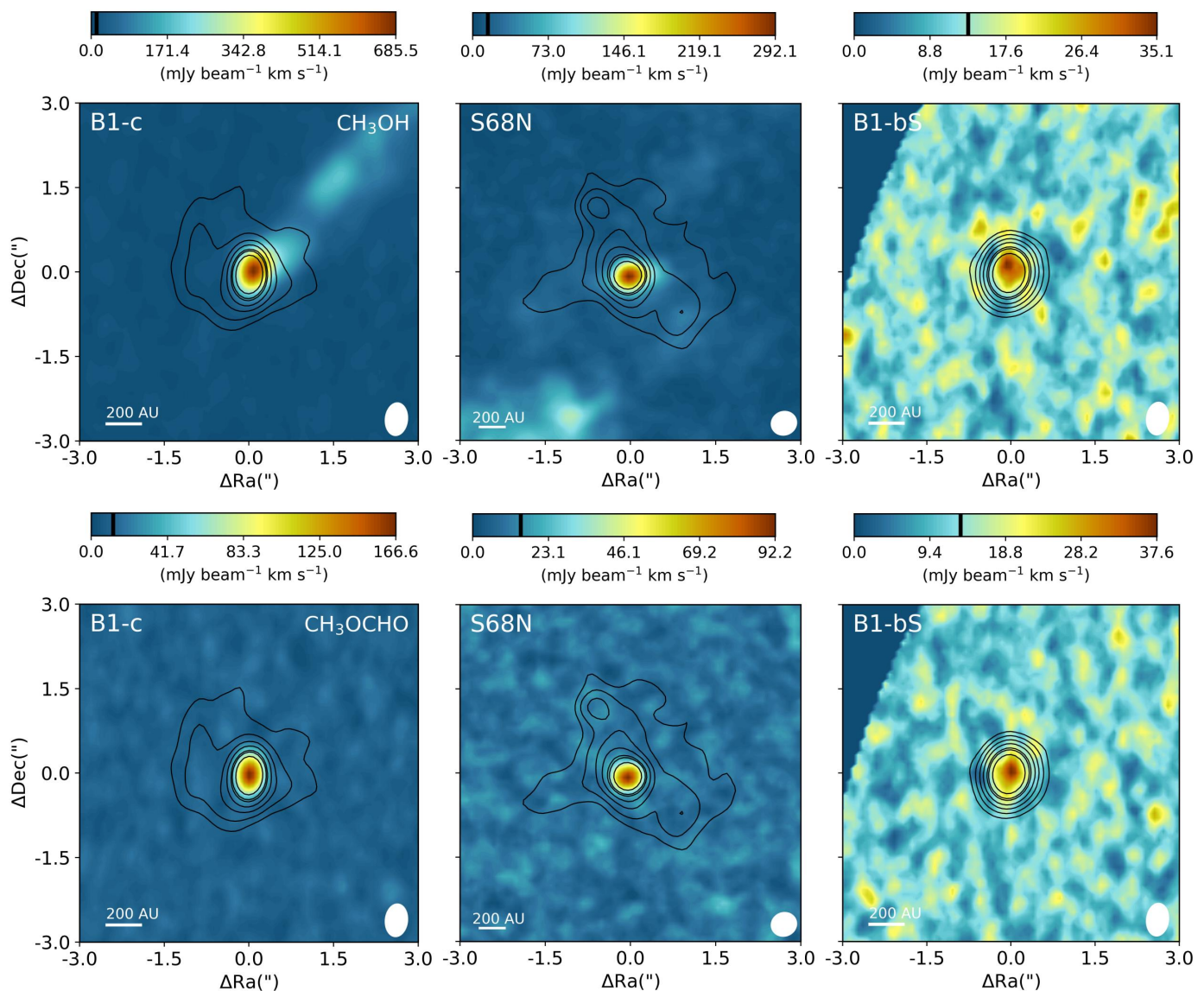


Fig. 1. ALMA Band 6 moment 0 images of the COM-rich sources analyzed in this study: B1-c (left), S68N (middle), and B1-bS (right). In color the spatial distribution of CH_3OH $2_{1,1}-1_{0,1}$ (top row, $E_{\text{up}} = 28$ K) and CH_3OCHO $2_{17,14}-2_{07,13}$ emission (bottom row, $E_{\text{up}} = 170$ K) is shown, with the color scale shown on top of each image. The images are integrated over $[-10, 10]$ km s^{-1} with respect to the V_{lsr} . In the colorbar, the $3\sigma_{\text{line}}$ level is indicated with the black bar, with $\sigma_{\text{line}} = 4.0, 4.9,$ and 4.4 $\text{mJy beam}^{-1} \text{km s}^{-1}$ for B1-c, S68N, and B1-bS, respectively. The continuum is shown with the black contours with increasingly $[3, 5, 8, 12, 18, 21, 30]$ σ_{cont} , where σ_{cont} is $1.5, 0.8,$ and 1.0 $\text{mJy beam}^{-1} \text{km s}^{-1}$ for B1-c, S68N, and B1-bS, respectively. The size of the beam is shown in the lower right of each image, and in the lower left a scale bar is displayed. The image of B1-bS has a lower S/N since the source is located on the edge of the primary beam.

A source size of $0.45''$ (equal to the Band 6 beam) is used for all sources. Given the equation for beam dilution:

$$\text{Dilutionfactor}^{-1} = \frac{\theta_s^2}{\theta_b^2 + \theta_s^2}, \quad (1)$$

with θ_s the (assumed) angular source size of $0.45''$ and θ_b the beam size, the beam dilution factor is ~ 20 and 2 for Band 3 and Band 6, respectively. In reality, the size of the emitting region is smaller than the assumed $0.45''$ and any derived column densities are thus lower limits. The effect of a smaller source size is discussed in Section 4.2

Each species is modelled separately, and optically thick lines and blended lines are excluded from the fitting procedure. Assuming that the populations of all levels can be described by a single excitation temperature T_{ex} , the χ^2 is computed for each

grid point, after which the best fitting model is determined by the minimum reduced χ^2 . This method is often called 'local thermodynamic equilibrium (LTE)' as the high-density limit of molecular excitation in which the excitation temperature approaches the kinetic temperature T_{kin} . Strictly speaking, thermodynamic equilibrium implies that not only the excitation but also the motions, ionization balance and radiation field are in equilibrium at the same temperature. The word 'local' refers to a specific region in an atmosphere with varying temperature and density, having its origin in stellar astrophysics. In radio astronomy of the interstellar medium, however, the word LTE often just implies a single excitation temperature to characterize the level populations of a molecule, which we will refer to here as 'LTE'.

The 2σ uncertainty on the column density, excitation temperature, and FWHM is computed from the grid. The main con-

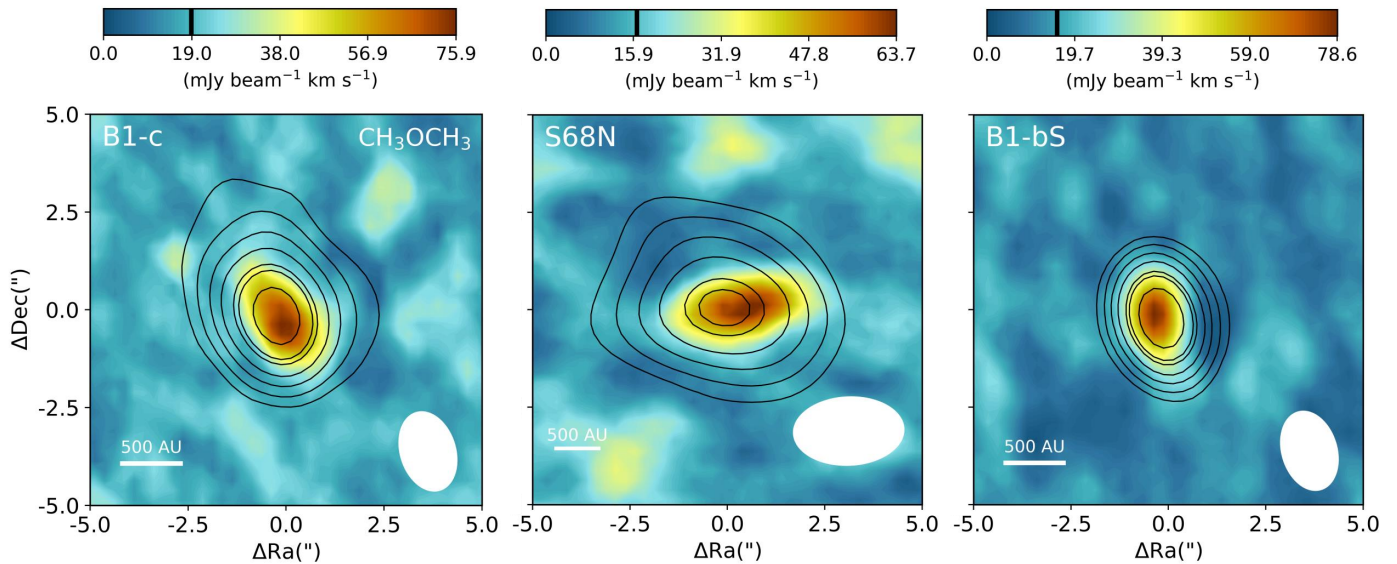


Fig. 2. Same as Fig. 1, but now for CH_3OCH_3 $7_{0,7}-6_{1,6}$ ($E_{\text{up}} = 25$ K) in Band 3. Here, $\sigma_{\text{line}} = 6.3, 5.5,$ and 5.2 $\text{mJy beam}^{-1} \text{km s}^{-1}$ for B1-c, S68N, and B1-bS, respectively. The continuum contours are $[3,5,8,12,18,21,30] \sigma_{\text{cont}}$, where σ_{cont} is $0.46, 0.54,$ and 0.71 $\text{mJy beam}^{-1} \text{km s}^{-1}$ for B1-c, S68N, and B1-bS, respectively.

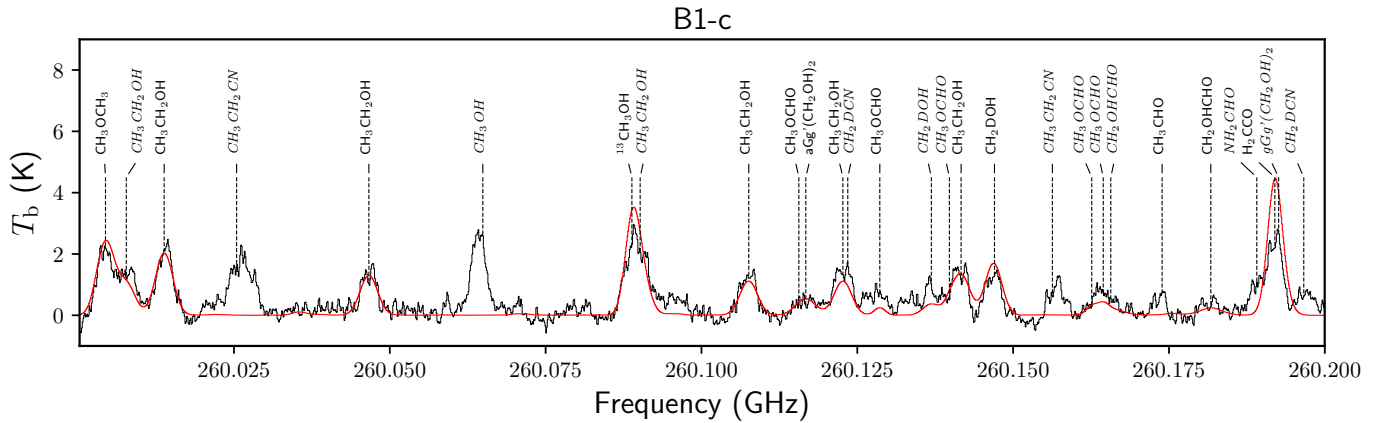


Fig. 3. Part of the Band 6 spectrum of B1-c shown in black with the best fitting model overplotted in red. The window is centred around a few CH_3CHO , CH_3OCHO , and $\text{CH}_3\text{CH}_2\text{OH}$ lines. The focus here is on the O-bearing COMs; for this reason several transitions clearly visible in the data (i.e., $\text{CH}_3\text{CH}_2\text{CN}$) do not show up in the fitting model. Lines in italics were excluded during the fitting.

tributors to the uncertainty are the assumption of 'LTE', the flux calibration error of ALMA, and the assumption of Gaussian line profiles. For sources with a high line density, such as B1-c, many lines had to be excluded from the fit due to line blending, therefore increasing the uncertainty on the column densities and excitation temperatures.

3.2. Column densities and excitation temperatures

The column densities and excitation temperatures of all COM-rich sources are presented in Appendix B. We consider a derived column density an upper limit if the error is more than 80% of the corresponding best fitting value. The number of lines used for fitting each species is listed in Table B.2. Upper limits (3σ) to the column density are provided if no unblended lines are detected. For B1-c in Band 6, unambiguous detections (i.e., > 5 unblended lines identified) are made for $^{13}\text{CH}_3\text{OH}$, $\text{CH}_3^{18}\text{OH}$, CH_2DOH , CH_3CHO , CH_3OCHO , CH_3OCH_3 , CH_3COCH_3 , and $\text{aGg}'(\text{CH}_2\text{OH})_2$. For S68N and B1-bS, line blending and sen-

sitivity hampered unambiguous detections for several of these COMs, but tentative detections can still be made for most species. H_2CCO and t-HCOOH show < 3 detected transitions, and their identifications are therefore considered tentative. $\text{gGg}'(\text{CH}_2\text{OH})_2$ and CH_2OHCHO are only detected toward B1-c in Band 6; for all other sources (and B1-c Band 3) we put upper limits on the column density for an excitation temperature of 200 K. Due to the absence of any optically thin CH_3OH lines in our data, the column density of CH_3OH is determined by scaling it from $^{13}\text{CH}_3\text{OH}$ for Band 3 and $\text{CH}_3^{18}\text{OH}$ for Band 6 (using $^{12}\text{C}/^{13}\text{C} \sim 70$ and $^{16}\text{O}/^{18}\text{O} \sim 560$ for the local ISM; [Milam et al. 2005](#); [Wilson & Rood 1994](#)). In Fig. 3, part of the Band 6 spectrum of B1-c is shown with the best fitting model overlaid. The full Band 6 spectra with best-fit model of all sources can be found in Appendix C. The full Band 3 spectrum of B1-c is presented in Appendix D.

B1-c is the most COM rich source of our sample. The measured column densities in Band 6 vary between $1.9 \times 10^{18} \text{cm}^{-2}$ for CH_3OH and $7.0 \times 10^{14} \text{cm}^{-2}$ for t-HCOOH in a ~ 150 AU

Table 2. Column densities of CH₃OH and abundance fractions (%) with respect to CH₃OH for all COM rich sources.

Species	Name	B1c		S68N		B1bS	
		Band 3	Band 6	Band 3	Band 6	Band 3	Band 6
CH ₃ OH	Methanol	18 ± 4	19 ± 6	3.6 ± 0.8	14 ± 6	2.4 ± 0.7	5.0 ± 0.6
¹³ CH ₃ OH		≡ 1.43	1.0 ± 0.3	≡ 1.43	0.7 ± 0.3	≡ 1.43	0.3 ± 0.1
CH ₃ ¹⁸ OH		< 1.0	≡ 0.18	< 2.8	≡ 0.18	< 4.2	≡ 0.18
CH ₂ DOH		5.7 ± 2.1	8.4 ± 2.8	< 13	4.3 ± 1.9	< 9.9	< 4.7
CH ₃ CH ₂ OH	Ethanol	1.6 ± 0.9	0.8 ± 0.3	< 5.6	0.22 ± 0.09	< 8.3	< 0.6
CH ₃ CHO	Acetaldehyde	1.3 ± 0.5	0.2 ± 0.1	< 1.4	0.07 ± 0.03	< 2.1	0.13 ± 0.02
CH ₃ OCHO	Methyl formate	2.3 ± 1.0	1.0 ± 0.3	0.29 ± 0.07	1.1 ± 0.5	6.1 ± 1.8	0.75 ± 0.10
CH ₃ OCH ₃	Dimethyl ether	1.0 ± 0.2	1.3 ± 0.4	1.8 ± 0.4	0.9 ± 0.4	4.3 ± 1.2	1.3 ± 0.2
CH ₃ COCH ₃	Acetone	< 2.8	0.4 ± 0.1	< 14	0.4 ± 0.2	< 21	0.10 ± 0.03
aGg'(CH ₂ OH) ₂	Ethylene glycol	< 2.4	0.3 ± 0.1	< 12	0.16 ± 0.07	< 18	< 0.43
gGg'(CH ₂ OH) ₂		< 1.2	0.4 ± 0.1	< 6	< 0.81	< 8.9	< 0.43
CH ₂ OHCHO	Glycolaldehyde	< 0.89	0.11 ± 0.04	< 2.2	< 0.06	< 3.3	< 0.064
H ₂ CCO	Ketene	–	0.07 ± 0.02	–	0.07 ± 0.03	–	0.03 ± 0.01
t-HCOOH	Formic acid	0.5 ± 0.1	0.04 ± 0.02	< 1.4	0.07 ± 0.03	< 4.2	< 0.1

Notes. The CH₃OH entry is the column density in 10¹⁷ cm⁻² in a 0.45" beam derived from ¹³CH₃OH for Band 3 and CH₃¹⁸OH for Band 6 using ¹²C/¹³C = 70 and ¹⁶O/¹⁸O = 560, respectively. All other entries are abundances with respect to CH₃OH (in %). The errors are derived from the 2σ (95%) uncertainties on the column densities. Abundance ratios which are set to their respective isotope ratio are indicated with a ≡ symbol.

diameter region. Our derived column density for CH₂OHCHO is 2.0 × 10¹⁵ cm⁻². Most COMs show column densities on the order of ~ 10¹⁵⁻¹⁶ cm⁻² in both Band 3 and Band 6.

The column densities toward S68N are a factor ~ 1.5 – 2 lower than toward B1-c. Similarly, the highest column density is derived for CH₃OH (1.4 × 10¹⁸ cm⁻²), the lowest for t-HCOOH (9.1 × 10¹⁴ cm⁻²) in Band 6, and ~ 10¹⁵⁻¹⁶ cm⁻² for most COMs.

B1-bS has the lowest column densities measured in this sample, up to an order of magnitude lower than toward B1-c with values of the order of ~ 10¹⁴⁻¹⁷ cm⁻². It was located in the edge of our ALMA field of view, resulting in a lower sensitivity and an increased level of noise. Our derived column densities (for a source size of 0.45") are up to a factor of a few lower than those derived by [Marcelino et al. \(2018\)](#) for a 0.35" source size, which is roughly the difference in beam dilution.

For molecules with multiple lines for which a T_{ex} is derived, the average excitation temperature in Band 3 is ~ 30 K lower than in Band 6. The average FWHM for B1-c and S68N in Band 6 is 3.3 ± 0.2 and 5.0 ± 0.2 km s⁻¹, respectively (the FWHM for B1-bS was fixed to 1 km s⁻¹). In Band 3 this is 3.1 ± 0.2 and 4.5 ± 0.5, respectively. The average FWHM in Band 3 is thus slightly lower than in Band 6 for B1-c and S68N, but within the uncertainties. This matches the expectation that the Band 3 observations trace the colder more extended regions and Band 6 being more sensitive to warmer compact regions since the transition upper energies in Band 3 are on average lower than in Band 6 (see [Tables E.2 and E.3](#)) and because the beam size and LAS are larger. When the Band 3 excitation temperature of all species for which T_{ex} was fixed is decreased by 30 K, no significant changes in the column densities and abundances arise. In [Table B.3](#), the results of a fit with $T_{\text{ex}} = 100$ K to the Band 3 data are presented. The derived column densities are typically about a factor ~ 2 lower than derived with T_{ex} as a free parameter or fixed to 200 K.

3.3. Relative abundances

Clear isotopologue detections are only made for methanol. The ¹³CH₃OH/CH₃¹⁸OH abundance ratio is < 6 for all targets, which is lower than the elemental value of ~ 8. This hints at the possibility that even ¹³CH₃OH is optically thick for these sources. The CH₂DOH/CH₃OH ratio is on the order of ~ 1 – 10%. Hints for features of isotopologues are seen for partially deuterated ethanol (CH₂DCH₂OH) and methyl formate (CH₃O¹³CHO), but these lines are too weak to allow for an unambiguous confirmation and deriving physical properties.

In order to compare our derived column densities, we determine the abundance ratios of each species with respect to CH₃OH. Because the usual tracer of H₂, CO, comes from more extended regions, CH₃OH is chosen as the reference species. In [Table 2](#), the derived abundances are presented for the three COM rich sources. The abundances between B1-c and S68N vary within a factor 2-3. This may simply be an effect of the source properties; the lines in S68N are, for example, a factor ~ 1.5 broader (~5.0 km s⁻¹ and ~3.3 km s⁻¹ in Band 6, respectively) resulting in more blended lines. In S68N, our abundances of CH₃OCHO, CH₃OCH₃, and H₂CCO are an order of magnitude lower than those derived by [Bergner et al. \(2019\)](#). The discrepancies are potentially due to different approaches to optical depth correction. Our observations of B1-bS have a low S/N. Nevertheless our derived abundance ratios coincide within a factor of a few with the hot (200 K) component modelled by [Marcelino et al. \(2018\)](#).

The abundances in Band 3 for B1-c deviate within a factor of ~2 from the abundances derived in Band 6. Only CH₃CHO and t-HCOOH are more abundant based on the Band 3 data. In S68N, CH₃OCHO is less abundant in Band 3 whereas CH₃OCH₃ is more abundant. In B1-bS, both CH₃OCHO and CH₃OCH₃ are significantly more abundant in Band 3, possibly because the column density of CH₃OH is underestimated.

4. Discussion

4.1. Occurrence of COMs in young protostars

The emission of COMs is assumed to originate from inside the $T_{\text{dust}} \sim 100 - 300$ K radius. Using the relation of Bisschop et al. (2007) for the 100 K radius in hot cores:

$$R_{T=100\text{K}} \approx 15.4 \sqrt{\frac{L}{L_{\odot}}} \text{ AU}, \quad (2)$$

with L the source luminosity, implies an $R_{T=100\text{K}}$ of 37.4 AU, 35.8 AU, and 11.6 AU for B1-c, S68N, and B1-bS, respectively. Our Band 6 beam size of $\sim 0.45''$ (corresponding to a radius of ~ 72 AU and ~ 98 AU for Perseus and Serpens, respectively) is larger than this radius and thus agrees with all COM emission being unresolved, except for any CH_3OH emission related to an outflow.

Previous studies have found that a higher luminosity implies more emission of COMs (e.g., Jørgensen et al. 2002; Young & Evans 2005; Visser et al. 2009). Following Eq. 2 we would indeed expect that the occurrence of COM emission is correlated to the source luminosity. Here, three of the seven sources studied show emission of COMs. However, our most luminous source, SMM3, does not exhibit any COM emission. On the other end, the two sources with the lowest luminosity, B1-bN and B1-b, do not show any COM emission either. Interestingly, B1-bS also has a rather low luminosity but does exhibit emission of COMs. This suggests that, for the targets that we observed, no clear correlation exists between the occurrence of gaseous COMs and source luminosity.

Not finding such a correlation is surprising. The relation for $R_{T=100\text{K}}$ in Eq. (2), however, assumes a spherically symmetric infalling envelope (Bisschop et al. 2007). In reality, our Class 0 sources may be surrounded by an accretion disk (e.g., Tobin et al. 2012; Murillo et al. 2013). The presence of a disk or disk-like structure may shift the $R_{T=100\text{K}}$ inwards as the temperature in the mid-plane of a disk-like structure is lower than in the surface layers (e.g., Jørgensen et al. 2005; Visser et al. 2009). However, even for a fixed $R_{T=100\text{K}}$, the amount of material with $T_{\text{dust}} > 100\text{K}$ (e.g., high enough to release the COMs from the dust grains) can vary between different sources due to, for example, differences in the disk mass and radius (Persson et al. 2016).

The absence of COM emission can also originate from other source properties. B1-bN has been suggested as a first hydrostatic core candidate (e.g., Pezzuto et al. 2012; Hirano & Liu 2014; Gerin et al. 2017); the temperature in the inner regions may not have reached the threshold for ice COM sublimation. B1-b has a high bolometric temperature, and could therefore be more evolved than the other sources. Here only a weak continuum is detected, indicating that B1-b may have a cleared inner region; all gaseous COMs have either already been destroyed or are incorporated as ices in the cold outer part. The strong ice absorption observed toward this source might originate in a region that is more extended than the inner B1-b envelope (Öberg et al. 2011). SMM3 shows strong continuum, which may imply that increased dust opacity hides any COM emission.

Despite the absence of a clear correlation between the occurrence of COM emission and source luminosity, an increase in luminosity should result in higher temperatures and an increased UV radiation field (Visser et al. 2009; Drozdovskaya et al. 2016). The abundance of COMs with respect to CH_3OH may therefore be related to the source luminosity. However, Taquet et al. (2015) and Belloche et al. (2020) have shown that for most COM abundances no clear correlation exists with luminosity, similar

to what is found here; see Fig. 4. Our range of source luminosities is significantly smaller (up to $\sim 20 L_{\odot}$ compared to $10^6 L_{\odot}$; Taquet et al. 2015), and a similar scatter of abundances can be seen for their lower luminosities.

4.2. Dependence on source size

In modelling the COM emission, a source size equal to the beam size ($0.45''$) was assumed for all sources. However, since the emission is spatially unresolved, the source size in reality will be smaller. A smaller emitting region will increase the derived column densities, and may result in spectral lines actually being optically thick. We can test this scenario by assuming the 100 K diameter as a source size. This gives an emitting region of $0.23''$, $0.16''$ and $0.07''$ for B1-c, S68N, and B1-bS, respectively. Assuming the lines remain optically thin, the column densities derived for a $0.45''$ source size can be scaled to a smaller size by computing the difference in beam dilution (see Eq. (1)):

$$N_1 \frac{\theta_{s_1}^2}{\theta_b^2 + \theta_{s_1}^2} = N_2 \frac{\theta_{s_2}^2}{\theta_b^2 + \theta_{s_2}^2}, \quad (3)$$

where $N_{1,2}$ are the column densities for respective source sizes $\theta_{s_{1,2}}$, and θ_b is the beam size.

For B1-c and S68N, the resulting column densities still give optically thin ($\tau < 0.1$) lines for all species except CH_3OH . However, the lines of CH_3OH are already optically thick for a $0.45''$ source size and the column density is derived from the ^{13}C and ^{18}O isotopologues, whose lines remain optically thin. This means that the abundances derived previously in this work for these two sources remain the same. For B1-bS most lines do become optically thick (i.e., $\tau > 0.1$), indicating that the abundances derived above in Table 2 for B1-bS should be used with care.

4.3. From cold (Band 3) to hot (Band 6) COMs

The temperature in the envelopes of Class 0 sources is expected to have an onion shell layered structure, with increasing temperature when moving closer to the star (e.g., Jørgensen et al. 2002). In warmer layers the COMs could be further processed to form more complex molecules (on the dust grains) and, more importantly, will all be released from the dust grains when the temperature reaches $T_{\text{dust}} \approx 100 - 300$ K. With our multi-Band observations, we can check for hints of this onion-layered temperature structure by comparing the excitation temperature of COMs in Band 3, which is more sensitive to extended emission, to the COMs in Band 6, which is more sensitive to compact emission. Indeed, the average excitation temperature for the few species for which an excitation temperature could be derived in Band 3 (i.e., $\text{CH}_3\text{CH}_2\text{OH}$, CH_3OCHO , and CH_3OCH_3) is about 30 K lower than in Band 6. However, as mentioned in Section 3.2, no significant changes in the column densities and abundances arise if the Band 3 excitation temperature of all species for which T_{ex} was fixed is decreased by 30 K. The outcome of the Band 6 model does not change when all lines with a low $E_{\text{up}} < 100$ K are excluded.

Jørgensen et al. (2018) modelled the abundances of O-bearing COMs in IRAS 16293B and found that they can be divided in two classes with temperatures of ~ 125 K (warm) and ~ 300 K (hot). Most COMs in our sample have excitation temperatures of ~ 200 K, which falls in the middle between these two classes. Only CH_3OCH_3 consistently shows $T_{\text{ex}} \sim 100$ K, both in the Band 3 and Band 6 observations for all sources. This

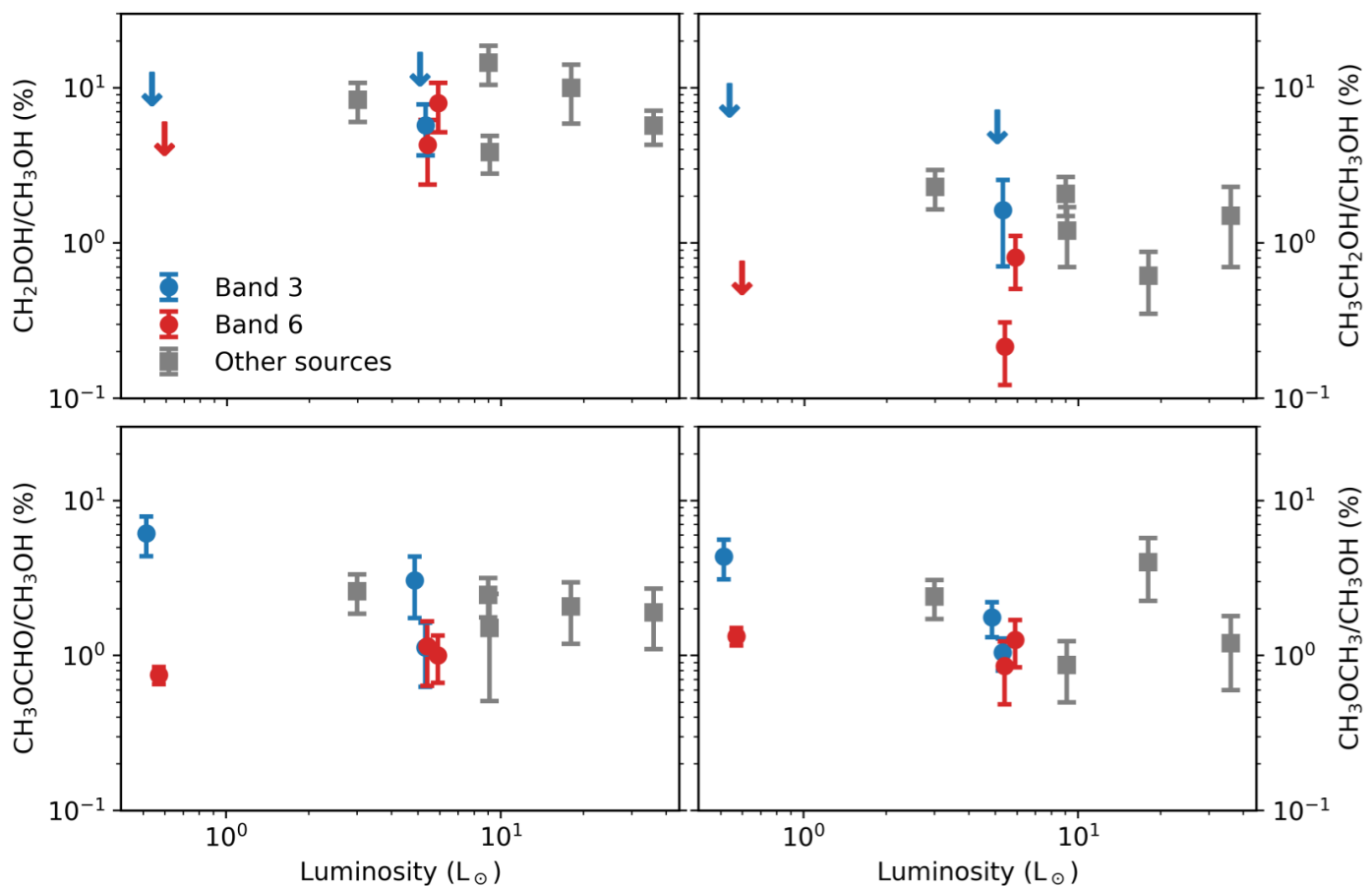


Fig. 4. Abundance of CH₂DOH (top left), CH₃CH₂OH (top right), CH₃OCHO (bottom left), and CH₃OCH₃ (bottom right) with respect to CH₃OH as function of bolometric source luminosity. Both the abundances in Band 3 (blue) and Band 6 (red) are plotted, where the Band 3 data-points are slightly shifted in luminosity for clarity. Arrows denote upper limits. In gray, the abundances of IRAS 2A and IRAS 4A (Taquet et al. 2015, 2019), HH 212 (Lee et al. 2019a), IRAS 16293A (Manigand et al. 2020), and IRAS 16293B (Jørgensen et al. 2018) are presented.

agrees with Jørgensen et al. (2018), who classified CH₃OCH₃ as a warm species. However, some ambiguity remains, since CH₃OCH₃ is also observed to have hot components in, for instance, higher mass cores (Belloche et al. 2013; Isokoski et al. 2013).

4.3.1. Multi-component model

The emission of COMs can be a superposition of a warm/hot inner component, and a second colder and more extended component. Earlier observational studies have shown that indeed a cold ($T < 100$ K, i.e., below the desorption temperature), more extended component may be present (e.g., Bisschop et al. 2007; Isokoski et al. 2013; Fayolle et al. 2015; Marcelino et al. 2018). Using our data, a possible cold component can be probed by fitting a multi-component model to the Band 3 data. This is only done for B1-c and S68N since the S/N of B1-bs in Band 3 is insufficient to carry out such analysis. H₂CCO, CH₃COCH₃, (CH₂OH)₂, and CH₂OHCHO are excluded from this analysis due to the lack of clear detections in our Band 3 data.

For the warm/hot component, the best-fit model of Band 6 is adopted. This component is kept fixed. As a second component, a colder model is introduced. Here, an excitation temperature of 60 K was assumed and only the column density was fitted for. The FWHM was initially set as a free parameter, but fixed to that of the warm/hot component in the cases that the fit results were unconstrained. A source size of 2.0" (equal to Band 3 beam) is

assumed for the cold component, compared with the source size of the warm component of 0.45".

The results are presented in Table 3. In Fig. 5 part of the Band 3 spectrum of B1-c centered around several CH₃OCHO lines is shown. The abundances of Table 3 are presented in a bar plot in Fig. 6. In B1-c, a cold ($T_{\text{ex}} = 60$ K) and more extended component is detected for most species. The column density of this cold component is typically a few % but can rise up to 10% of the warm/hot inner component. The abundances of more complex species in the cold component are higher than those in the warm/hot component, which contradicts the idea that COMs are more effectively processed and released from icy grains in warmer layers. However, one should also note that the column density of ¹³CH₃OH, and therefore CH₃OH, is likely underestimated.

Only for CH₃CH₂OH and CH₃OCHO, a FWHM of 3.3 ± 0.8 could be derived for the cold component which agrees with the average FWHM of the warm/hot component of 3.3 ± 0.2 . The line width of the cold component can be expected to be smaller since the emitting material is located further out in the envelope. However, Murillo et al. (2018) find that lower resolution data can also result in a larger line width, especially if the molecule is associated with the outflow or its cavity.

The presence of a cold component of COM emission indicates that some gas-phase formation routes are already efficient in the cold envelope as most COMs formed through grain-surface chemistry are still frozen out at these temperatures. Or

Table 3. Column densities resulting from the multi-component analysis.

Species	S68N			
	Component 1: warm/hot		Component 2: cold	
	N (cm^{-2})	$X/\text{CH}_3\text{OH}$ (%)	N (cm^{-2})	$X/\text{CH}_3\text{OH}$ (%)
CH_3OH	$(1.9 \pm 0.6) \times 10^{18}$	≈ 100	$(1.2 \pm 0.7) \times 10^{16}$	≈ 100
$^{13}\text{CH}_3\text{OH}$	$(1.8 \pm 0.2) \times 10^{16}$	1.0 ± 0.3	$(1.7 \pm 1.0) \times 10^{14}$	≈ 1.43
$\text{CH}_3^{18}\text{OH}$	$(3.4 \pm 1.1) \times 10^{15}$	≈ 0.18	$< 7.0 \times 10^{15}$	< 62
CH_2DOH	$(1.6 \pm 0.1) \times 10^{17}$	8.4 ± 2.8	$< 1.6 \times 10^{14}$	< 15
$\text{CH}_3\text{CH}_2\text{OH}$	$(1.5 \pm 0.3) \times 10^{16}$	0.8 ± 0.3	$(1.2 \pm 0.2) \times 10^{15}$	9.8 ± 6.1
CH_3CHO	$(4.6 \pm 1.0) \times 10^{15}$	0.24 ± 0.10	$(1.8 \pm 0.2) \times 10^{14}$	1.5 ± 0.9
CH_3OCHO	$(1.9 \pm 0.1) \times 10^{16}$	1.0 ± 0.3	$(4.7 \pm 0.3) \times 10^{14}$	3.8 ± 2.3
CH_3OCH_3	$(2.4 \pm 0.1) \times 10^{16}$	1.3 ± 0.4	$< 1.0 \times 10^{14}$	< 0.22
t-HCOOH	$(7.0 \pm 2.0) \times 10^{14}$	0.04 ± 0.02	$(2.2 \pm 0.2) \times 10^{14}$	1.8 ± 1.1

Notes. The values of the warm/hot Component 1 are the same as derived for Band 6 in a $0.45''$ beam and are fixed during the fitting (typical $T_{\text{ex}} \sim 200$ K). Component 2 represents cold (typical $T_{\text{ex}} = 60$ K), more extended material in a $2.0''$ beam. The column density of CH_3OH is determined by scaling from $^{13}\text{CH}_3\text{OH}$ for Band 3 and $\text{CH}_3^{18}\text{OH}$ for Band 6 using $^{12}\text{C}/^{13}\text{C} = 70$ and $^{16}\text{O}/^{18}\text{O} = 560$. Abundance ratio's which are set to their respective isotope ratio are indicated with a \approx symbol.

(¹) No abundance is computed since the column densities of both this species and CH_3OH are upper limits. (²) Lower limit to the abundance due to an upper limit of the CH_3OH column density.

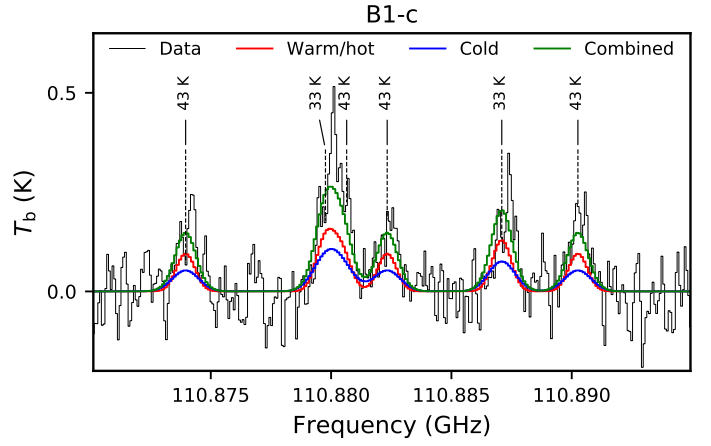


Fig. 5. Part of the Band 3 spectrum of the warm/hot ($T_{\text{ex}} \sim 100\text{--}300$ K) compact (red), cold ($T_{\text{ex}} = 60$ K) extended (blue), and combined (green) model overlaid on the data (black). The spectral features are from CH_3OCHO , with the upper energy indicated on top.

alternatively, that other mechanisms are capable of transferring the COMs from the solid state to the gas phase. Possible mechanisms include photodesorption (Fayolle et al. 2011; Muñoz Caro et al. 2016), desorption of fragments following photo-dissociation (Bertin et al. 2016), low temperature co-desorption (Fayolle et al. 2013; Ligterink et al. 2018c), or reactive desorption via a thermal hot-spot (Minissale et al. 2016; Chuang et al. 2018). Only for CH_3OCH_3 and the deuterated and ^{18}O isotopologues of CH_3OH , an extended component seems to be absent.

Formic acid is very abundant in the cold, extended component. This disagrees with the hot identification in IRAS 16293B (Jørgensen et al. 2018), but is more consistent with the cold identification in high-mass star-forming regions (Bisschop et al. 2007; Belloche et al. 2013) and it likely being an abundant ice species (Öberg et al. 2011; Boogert et al. 2015). However, both in Band 3 and Band 6, only 1–3 lines could be fitted, hence the identification of t-HCOOH should be considered rather tentative.

In S68N, a cold component is less evident. Only for CH_3OCHO a cold column density is derived; for all other species only upper limits are found. These results suggest that a possible cold component would have a column density of $\lesssim 1\%$ of the warm component.

4.4. Comparison to other sources

Our Band 6 abundances are compared to the ALMA Band 7 abundances of IRAS 16293B (Jørgensen et al. 2018) and HH 212 (Lee et al. 2019a), see Fig. 7. The abundances of several COMs (i.e., CH_2DOH , CH_3OCHO , CH_3OCH_3 , and CH_3COCH_3) from this work are very similar to IRAS 16293B and HH 212; the distribution of abundances seems to span only a factor of a few. The abundances are also within a factor 3 of IRAS 2A and IRAS 4A (Taquet et al. 2015). This similarity in abundances is rather interesting given that the source properties (e.g., L_{bol} , T_{bol} , etc.) are different for all sources. For example, IRAS 2A has a luminosity of $36 L_{\odot}$, whereas the luminosity of B1-bS is two orders of magnitude lower ($0.57 L_{\odot}$). A similarly large range is seen for the bolometric temperature and envelope mass. Moreover, these sources originate from four different star-forming regions: Perseus (B1-bS, B1-c, IRAS 2A, IRAS 4A), Serpens (S68N), Ophiuchus (IRAS 16293B), and Orion (HH 212). Apparently, despite the differences in source properties and local environment, the internal distribution of these COM abundances

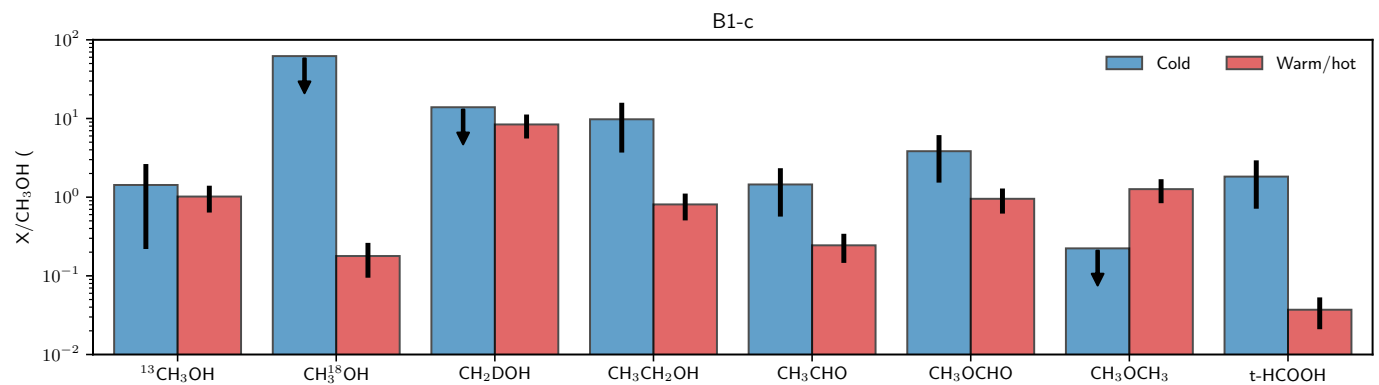


Fig. 6. Abundance of several species with respect to CH_3OH for our multi-component analysis. The warm ($T_{\text{ex}} \sim 200$ K) component is fixed to the Band 6 model with a $0.45''$ source size, whereas the cold ($T_{\text{ex}} = 60$ K) component is fitted with a more extended $2.0''$ source size. The 2σ (95%) errors are shown in black, with arrows denoting upper limits. The CH_3OH column density is determined by scaling from $^{13}\text{CH}_3\text{OH}$ using $^{12}\text{C}/^{13}\text{C} = 70$ and $^{16}\text{O}/^{18}\text{O} = 560$, respectively. The higher abundances in the cold component are likely due to an underestimate of the column density of $^{13}\text{CH}_3\text{OH}$.

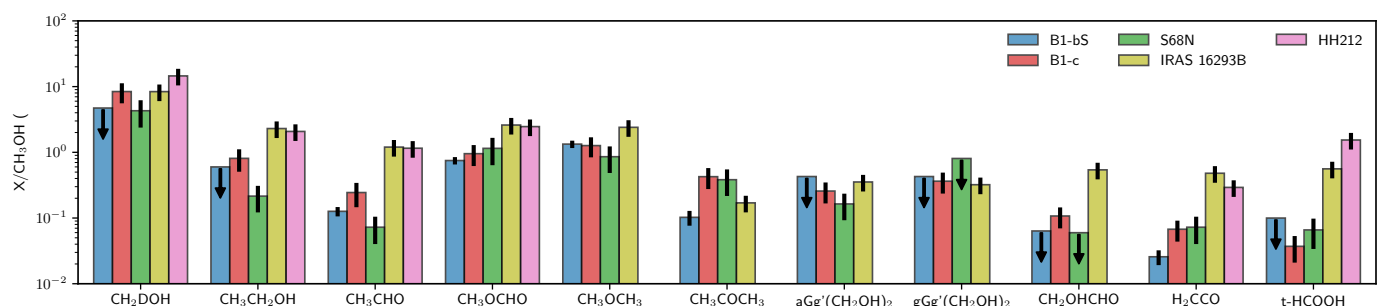


Fig. 7. Bar plot comparing our derived Band 6 abundances to other sources. The IRAS 16293B abundances for all species except $(\text{CH}_2\text{OH})_2$, CH_2OHCHO , and CH_3COCH_3 are from Jørgensen et al. (2018); the $(\text{CH}_2\text{OH})_2$ and CH_2OHCHO abundances are from Jørgensen et al. (2016), and the CH_3COCH_3 abundances from Lykke et al. (2017). The HH 212 abundances are from Lee et al. (2019a). The 2σ (95%) errors are shown in black, with arrows denoting upper limits. We note that the abundances of our targets are remarkably similar and show comparable values to those derived for IRAS 16293B and HH 212.

is roughly similar. This indicates that their formation schemes must follow a rather universal recipe, for instance following the atom addition and abstraction reaction networks proposed to govern the surface chemistry in cold interstellar clouds (e.g., Chuang et al. 2016). This implies that solid-state processes are already enriching the complex organic inventory of the interstellar medium in the prestellar phase.

Interestingly, our sources are under-abundant up to more than one order of magnitude in $\text{CH}_3\text{CH}_2\text{OH}$, CH_3CHO , and CH_2OHCHO compared with IRAS 16293B, HH 212, IRAS 2A and IRAS 4A. Our abundances of these COMs do fall within a factor of a few with IRAS 16293A (Manigand et al. 2020). This suggests that for some COMs local source conditions do become important for the formation or the destruction. The column density ratio of CH_3OCHO to CH_2OHCHO of B1-c is consistent with the lower part of the bimodal distribution found by El-Abd et al. (2019). For H_2CCO and t-HCOOH , our sources also have lower abundances. The abundance of t-HCOOH in the cold component of the multi component analysis (see Section 4.3.1) is consistent with IRAS 16293B and HH 212. However, the detections of these species are rather tentative since only < 3 clear unblended lines (not taking into account (hyper)fine structured lines) are detected.

In more massive star-forming regions, a larger scatter in COM abundances seems to be present. AFGL 4176 has abundances very similar to ours for most COMs (Bøgelund et al.

2019), whereas those in Orion KL show deviations: the abundance of $\text{CH}_3\text{CH}_2\text{OH}$ is similar (Tercero et al. 2018), but CH_3OCHO , CH_3OCH_3 have lower abundances in our sources, and CH_3COCH_3 higher abundances (see Table 6 in Bøgelund et al. 2019). Even larger deviations are found toward Sagittarius B2, where the abundances of all COMs are higher by up to an order of magnitude (Belloche et al. 2016).

The abundances derived here are higher than those in chemical models of a collapsing cloud for most COMs (e.g., Garrod 2013). Especially for CH_3COCH_3 and $(\text{CH}_2\text{OH})_2$, these models underestimate the abundances by up to an order of magnitude. This may be a direct result for a still not fully characterized solid-state and gas-phase chemistry and underlying gas-grain interactions (see discussion by Bøgelund et al. 2019).

4.5. Comparison with ices

No clear COM ice features have (yet) been observed toward any of our COM-rich sources. However, toward B1-b (from the same cloud as B1-c) hints for CH_3CHO , $\text{CH}_3\text{CH}_2\text{OH}$, and t-HCOOH ice have been detected with an abundance of $\sim 10\%$ with respect to CH_3OH (Öberg et al. 2011; Boogert et al. 2015). These are higher than those derived for the gaseous warm/hot component in this work ($\sim 0.1 - 1\%$). A possible explanation for this is that, besides from typical grain-surface chemistry, additional gas-phase chemistry in the cold envelope can contribute

in transforming CH_3OH into more complex species such as CH_3CHO and $\text{CH}_3\text{CH}_2\text{OH}$ and thus lowering the abundance of these species with respect to CH_3OH (e.g., Balucani et al. 2015). Another explanation is that other non-thermal desorption mechanisms, such as desorption of fragments following photodissociation (Bertin et al. 2016), are active.

To better compare the abundances of COMs in the ice and gas phase, it is essential to directly observe the ices. However, the above mentioned detections are still very uncertain, with only CH_3OH being securely identified in interstellar ices. Measuring the column densities of interstellar COM ices remains difficult since current infrared telescopes lack the sensitivity or wavelength coverage, and because the spectral features originating from ices are broad and depend on the ice composition. Moreover, the number of COMs for which extensive high resolution infrared laboratory spectra have become available is limited (Hudson et al. 2018; Terwisscha van Scheltinga et al. 2018). Future mid-infrared facilities such as JWST/MIRI will allow for directly observing COMs in the ice, which will be essential for studying their formation and evolution from the cold envelope toward the warm inner regions.

4.6. Temperature dependence of deuterated methanol

The deuteration fraction of methanol is an indication of the temperature during the solid-state formation of this species. At low temperatures, deuterium is more efficient than regular hydrogen in the hydrogenation process on interstellar dust grains (e.g., Nagaoka et al. 2005). This leads to a higher deuteration fraction than the elemental ratio ($\text{D}/\text{H} \sim 10^{-5}$; Caselli & Ceccarelli 2012). Bøgelund et al. (2018) investigated the effect of temperature during the formation of methanol with the astrochemical GRAINOBLE model (Taquet et al. 2012). In order to compare our methanol D/H ratios to their model results, our gas-phase CH_2DOH abundances have to be corrected for statistical weights; a D atom has a 3 times higher probability to stick into the CH_3 group than into the OH group. In Band 3, this gives a D/H ratio of $1.9 \pm 0.7\%$, $< 4.3\%$, and $< 3.3\%$ for B1-c, S68N, and B1-bS respectively. In Band 6, this is $2.7 \pm 0.9\%$, $1.4 \pm 0.6\%$, and $< 1.6\%$. These deuteration fractions correspond to a temperature during the formation of methanol of ~ 15 K, where a density of $n_{\text{H}} = 10^{5-6} \text{ cm}^{-3}$ was assumed (cf. Fig. 8 of Bøgelund et al. 2018).

Our derived methanol D/H fractions agree well with other well-studied low-mass protostars such as IRAS16293B ($\sim 2\%$; Jørgensen et al. 2018), IRAS 2A and IRAS 4A ($\sim 2\%$ and $\sim 1\%$, respectively; Taquet et al. 2019), and HH 212 ($\sim 1\%$ after taking into account statistical weights; Bianchi et al. 2017; Taquet et al. 2019). Our fractions are up to an order of magnitude higher than for high-mass star-forming regions such as NGC 6334I ($\sim 0.1\%$ on average; Bøgelund et al. 2018), and Sagittarius B2 (0.04% after statistical correction; Belloche et al. 2016). The lower D/H ratios in more massive star-forming regions hint at a higher temperature (i.e., > 20 K) already during the earliest phase of high-mass star formation.

5. Summary

In this paper, the content of oxygen bearing COMs is studied in seven young protostellar systems. High resolution ALMA observations in both Band 3 and Band 6 were used to determine the column densities and excitation temperatures. The main conclusions are as follows:

- Three out of the seven sources exhibit warm COM emission features with typical $T_{\text{ex}} \sim 200$ K. There seems to be no correlation between the bolometric luminosity (and temperature) and occurrence of COMs of these Class 0 sources. Other effects, such as disk formation and clearing of an inner region, may play a role as well.
- A multi-component model was introduced to check whether a cold ($T_{\text{ex}} = 60$ K), more extended component of COM emission is present. Only B1-c shows clear signs of such a cold component, with typically a few % of the column density of the warm component. This indicates that in the cold envelope already gas-phase formation may take place and that this could be a direct consequence of the grain-gas transitions of COMs or (radical) COM fragments. t-HCOOH is significantly more abundant in the cold component, hinting at the cold origin of this species.
- The abundance of several COMs (i.e., CH_2DOH , CH_3OCHO , CH_3OCH_3 , CH_3COCH_3 , and $(\text{CH}_2\text{OH})_2$) are remarkably similar in comparison to other young Class 0 protostars in different star-forming regions such as IRAS 16293-2422B and HH 212. This similarity suggests that the distribution of these COM abundances is roughly fixed at an early cold, i.e. prestellar stage. However, the abundances of some COMs (i.e., $\text{CH}_3\text{CH}_2\text{OH}$, CH_3CHO , and CH_2OHCHO) in our sources differ significantly compared to IRAS 16293B and HH 212, possibly because local source conditions become important and affect formation and destruction pathways in different ways. Astrochemical modeling will be needed to extend on these findings.
- The D/H ratio of deuterated CH_3OH is around a few % for all COM-rich sources, suggesting a dust temperature of ~ 15 K during its solid-state formation. The derived formation temperature is similar to other low-mass protostars, but lower than in high-mass star-forming regions (> 20 K).

Our sample of COM-rich protostars studied on solar system scales remains small. It is still unknown how much of the prestellar cloud material is reprocessed during the evolution toward protostellar systems and more evolved sources. The abundance of most COMs in the ice phase is still highly uncertain. Future mid-infrared facilities, most notably the *James Webb Space Telescope*, will provide vital information on COM ices. A larger sample of COM-rich sources with ALMA, both in the Class 0 phase and in the more evolved Class I phase where most COM emission has already disappeared, would provide better constraints on the chemical complexity during these earliest phases of star formation. The present work already shows that such studies will be valuable, given the rich distribution in COMs discussed here.

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Appendix A: Laboratory spectroscopic data

All laboratory spectroscopic data are acquired using the CDMS (Müller et al. 2001, 2005; Endres et al. 2016) and JPL (Pickett et al. 1998) catalogs. In some of the entries, vibrational or torsional states are not taken into account in the calculation of the partition function. At low temperatures, the contribution of these states are negligible. At higher temperatures (i.e., $T > 100$ K), however, a so-called vibrational correction factor must be applied if the torsional and vibrational states are not accounted for in the partition functions. Below, we detail the laboratory spectroscopy of each individual species discussed in this paper and indicate if a vibrational correction factor was applied.

The data of CH₃OH are taken from the CDMS database and are based on the work of Xu et al. (2008). The ground state and first three torsional states are taken into account in the partition function. The entry of the ¹³C isotopologue takes the first two torsional states into account (Xu & Lovas 1997). They assumed that the dipole moment is the same as the main isotopologue and calculated the partition function taking only the permanent dipole moment into account. The CH₃¹⁸O entry is based mostly on the data of Fisher et al. (2007). The spectroscopic data of CH₂DOH are taken from the JPL catalog, where the entry is based on the work of Pearson et al. (2012). The vibrational correction factor at 200 K is calculated to be 1.181 by summing up over torsional substates extrapolated from Lauvergnat et al. (2009)

CH₃CH₂OH exists as two conformers: *trans* and *gauche*, which originates from the torsion in the OH group. The spectroscopic data are taken from the CDMS catalog, where the entry is based on Pearson et al. (2008) and updated by Müller et al. (2016), who noticed that the old catalog entry did not predict the line intensities correctly around 3 mm in Sagittarius B2. This applies also to other frequency regions, but not to all transitions. Both conformers are included in the entry.

The data of CH₃CHO are taken from the JPL catalog. It is based on laboratory spectroscopy of Kleiner et al. (1996) and includes the first and second torsional state in its partition function.

The spectroscopic data of CH₃OCHO are taken from the JPL database. The entry is based on the work of Ilyushin et al. (2009) and Brown et al. (1975). The partition function includes states up to the first torsional state.

CH₃OCH₃ data are taken from the CDMS entry, which includes mostly the work of Endres et al. (2009). The partition function includes the ground vibrational state and excited states up to 500 cm⁻¹. However, for the excitation temperatures derived in this work (~100 K), the contribution of higher order vibrational states to the partition function is negligible.

The CH₃COCH₃ data are taken from the JPL catalog. The rotational transitions were calculated by Groner et al. (2002) and include only the ground state. Here, we use an updated version of the entry. It is based on the study of Ordu et al. (2019) with the majority of the lines from Groner et al. (2002).

The (CH₂OH)₂ structure consists of 3 coupled rotors, which give rise to ten different stable configurations (Christen et al. 2001). Here, only the two lowest energy conformers of (CH₂OH)₂ are studied. The aGg' state is the lowest energy state whereas the gGg' state lies about 290 K above that (Müller & Christen 2004). The spectroscopic data are taken from the CDMS entries, which are based on the work of Christen et al. (1995) and Christen & Müller (2003) for the aGg' conformer and Christen et al. (2001) and Müller & Christen (2004) for the gGg' conformer. For gGg'(CH₂OH)₂, the partition function and line intensities were calculated assuming as if it is the low-

est energy state conformer. Since for these calculations only the ground state is considered, we approximate the effect of vibrational states by using a vibrational correction factor of 2.143 for both conformers at 200 K.

The CH₂OHCHO data are taken from the CDMS catalog. The entry is based on the work of Butler et al. (2001) and only includes data on the ground vibrational state. A vibrational correction factor of 1.5901 is calculated at 200 K in the harmonic approximation.

The spectroscopic data of H₂CCO are acquired from the CDMS catalog. The entry is calculated mostly by Johnson & Strandberg (1952), Fabricant et al. (1977), and Brown et al. (1990). The contribution of vibrational states is assumed to be negligible.

Formic acid, HCOOH, exists both in a *trans* and *cis* state, where the lowest energy state of *c*-HCOOH is about 2000 K higher than for *t*-HCOOH. In this study we only detect *t*-HCOOH, for which we use the entry of the JPL catalog. This is based on the work of Bellet et al. (1971). The vibrational correction is < 1.1 at 200 K and is therefore neglected.

Appendix B: CASSIS modelling results

Table B.1. Computed Band 3 and Band 6 column densities and excitation temperatures of our sources.

B1-c							
Species	Catalog	Band 3			Band 6		
		T_{ex} (K)	N^l (cm^{-2})	X/CH ₃ OH (%)	T_{ex} (K)	N^l (cm^{-2})	X/CH ₃ OH (%)
CH ₃ OH	CDMS	–	$(1.8 \pm 0.4) \times 10^{18}$	$\equiv 100$	–	$(1.9 \pm 0.6) \times 10^{18}$	$\equiv 100$
¹³ CH ₃ OH	CDMS	[190]	$(2.6 \pm 0.6) \times 10^{16}$	$\equiv 1.43$	190 ± 30	$(1.8 \pm 0.2) \times 10^{16}$	1.0 ± 0.3
CH ₃ ¹⁸ OH	CDMS	[140]	$< 1.8 \times 10^{16}$	< 1.0	140 ± 60	$(3.4 \pm 1.1) \times 10^{15}$	$\equiv 0.18$
CH ₂ DOH	JPL	210 ± 60	$(1.0 \pm 0.3) \times 10^{17}$	5.7 ± 2.1	180 ± 20	$(1.6 \pm 0.1) \times 10^{17}$	8.4 ± 2.8
CH ₃ CH ₂ OH	CDMS	160 ± 60	$(2.9 \pm 1.5) \times 10^{16}$	1.6 ± 0.9	250 ± 60	$(1.5 \pm 0.3) \times 10^{16}$	0.8 ± 0.3
CH ₃ CHO	JPL	260 ± 50	$(2.4 \pm 0.8) \times 10^{16}$	1.3 ± 0.5	260 ± 60	$(4.6 \pm 1.0) \times 10^{15}$	0.24 ± 0.10
CH ₃ OCHO	JPL	120 ± 40	$(2.0 \pm 0.8) \times 10^{16}$	1.1 ± 0.5	180 ± 20	$(1.9 \pm 0.1) \times 10^{16}$	1.0 ± 0.3
CH ₃ OCH ₃	CDMS	100 ± 10	$(1.9 \pm 0.1) \times 10^{16}$	1.0 ± 0.2	120 ± 10	$(2.4 \pm 0.1) \times 10^{16}$	1.3 ± 0.4
CH ₃ COCH ₃	JPL	[200]	$< 5.0 \times 10^{16}$	< 2.8	[200]	$(8.0 \pm 0.9) \times 10^{15}$	0.4 ± 0.1
aGg'(CH ₂ OH) ₂	CDMS	[180]	$< 4.2 \times 10^{16}$	< 2.4	180 ± 100	$(4.8 \pm 0.6) \times 10^{15}$	0.3 ± 0.1
gGg'(CH ₂ OH) ₂	CDMS	[200]	$< 2.2 \times 10^{16}$	< 1.2	[200]	$(6.8 \pm 0.8) \times 10^{15}$	0.4 ± 0.1
CH ₂ OHCHO	CDMS	[200]	$< 1.6 \times 10^{16}$	< 0.89	[200]	$(2.0 \pm 0.2) \times 10^{15}$	0.11 ± 0.04
H ₂ CCO	CDMS	–	–	–	[200]	$(1.3 \pm 0.1) \times 10^{15}$	0.07 ± 0.02
t-HCOOH	JPL	[200]	$(9.0 \pm 1.0) \times 10^{15}$	0.5 ± 0.1	[200]	$(7.0 \pm 2.0) \times 10^{14}$	0.04 ± 0.02

S68N							
Species	Catalog	Band 3			Band 6		
		T_{ex} (K)	N^l (cm^{-2})	X/CH ₃ OH (%)	T_{ex} (K)	N^l (cm^{-2})	X/CH ₃ OH (%)
CH ₃ OH	CDMS	–	$(3.6 \pm 0.8) \times 10^{17}$	$\equiv 100$	–	$(1.4 \pm 0.6) \times 10^{18}$	$\equiv 100$
¹³ CH ₃ OH	CDMS	[160]	$(5.1 \pm 1.2) \times 10^{15}$	$\equiv 1.43$	160 ± 20	$(1.0 \pm 0.1) \times 10^{16}$	0.7 ± 0.3
CH ₃ ¹⁸ OH	CDMS	[200]	$< 1.0 \times 10^{16}$	< 2.8	200 ± 80	$(2.5 \pm 1.1) \times 10^{15}$	$\equiv 0.18$
CH ₂ DOH	JPL	[200]	$< 4.7 \times 10^{16}$	< 13	[200]	$(6.0 \pm 0.7) \times 10^{16}$	4.3 ± 1.9
CH ₃ CH ₂ OH	CDMS	[200]	$< 2.0 \times 10^{16}$	< 5.6	[200]	$(3.0 \pm 0.2) \times 10^{15}$	0.22 ± 0.09
CH ₃ CHO	JPL	[200]	$< 5.0 \times 10^{15}$	< 1.4	[200]	$(1.0 \pm 0.1) \times 10^{15}$	0.07 ± 0.03
CH ₃ OCHO	JPL	220 ± 80	$(1.6 \pm 0.9) \times 10^{16}$	3.9 ± 2.5	290 ± 30	$(1.6 \pm 0.2) \times 10^{16}$	1.1 ± 0.5
CH ₃ OCH ₃	CDMS	110 ± 10	$(6.4 \pm 0.7) \times 10^{15}$	1.8 ± 0.4	90 ± 10	$(1.2 \pm 0.1) \times 10^{16}$	0.9 ± 0.4
CH ₃ COCH ₃	JPL	[200]	$< 5.0 \times 10^{16}$	< 14	[200]	$(5.3 \pm 0.3) \times 10^{15}$	0.4 ± 0.2
aGg'(CH ₂ OH) ₂	CDMS	[200]	$< 4.2 \times 10^{16}$	< 12	[200]	$(2.3 \pm 0.1) \times 10^{15}$	0.16 ± 0.07
gGg'(CH ₂ OH) ₂	CDMS	[200]	$< 2.2 \times 10^{16}$	< 6	[200]	$< 1.1 \times 10^{16}$	< 0.81
CH ₂ OHCHO	CDMS	[200]	$< 8.0 \times 10^{15}$	< 2.2	[200]	$< 8.0 \times 10^{14}$	< 0.06
H ₂ CCO	CDMS	–	–	–	[200]	$(1.0 \pm 0.1) \times 10^{15}$	0.07 ± 0.03
t-HCOOH	JPL	[200]	$< 5.0 \times 10^{15}$	< 1.4	[200]	$(9.1 \pm 2.1) \times 10^{14}$	0.07 ± 0.03

B1-bS							
Species	Catalog	Band 3			Band 6		
		T_{ex} (K)	N^l (cm^{-2})	X/CH ₃ OH (%)	T_{ex} (K)	N^l (cm^{-2})	X/CH ₃ OH (%)
CH ₃ OH	CDMS	–	$(2.4 \pm 0.7) \times 10^{17}$	$\equiv 100$	–	$(5.0 \pm 0.6) \times 10^{17}$	$\equiv 100$
¹³ CH ₃ OH	CDMS	[160]	$(3.5 \pm 1.0) \times 10^{15}$	$\equiv 1.43$	160 ± 40	$(1.5 \pm 0.5) \times 10^{15}$	0.3 ± 0.1
CH ₃ ¹⁸ OH	CDMS	[200]	$< 1.0 \times 10^{16}$	< 4.2	[200]	$(9.0 \pm 1.0) \times 10^{14}$	$\equiv 0.18$
CH ₂ DOH	JPL	[200]	$< 2.4 \times 10^{16}$	< 9.9	[200]	$< 2.4 \times 10^{16}$	< 4.7
CH ₃ CH ₂ OH	CDMS	[200]	$< 2.0 \times 10^{16}$	< 8.3	[200]	$< 3.0 \times 10^{15}$	< 0.6
CH ₃ CHO	JPL	[200]	$< 5.0 \times 10^{15}$	< 2.1	[200]	$(6.4 \pm 0.7) \times 10^{14}$	0.13 ± 0.02
CH ₃ OCHO	JPL	[200]	$(1.5 \pm 0.1) \times 10^{16}$	6.1 ± 1.8	[200]	$(3.8 \pm 0.2) \times 10^{15}$	0.75 ± 0.10
CH ₃ OCH ₃	CDMS	90 ± 10	$(1.1 \pm 0.1) \times 10^{16}$	4.3 ± 1.2	110 ± 10	$(6.7 \pm 0.4) \times 10^{15}$	1.3 ± 0.2
CH ₃ COCH ₃	JPL	[200]	$< 5.0 \times 10^{16}$	< 21	[200]	$(5.1 \pm 1.2) \times 10^{14}$	0.10 ± 0.03
aGg'(CH ₂ OH) ₂	CDMS	[200]	$< 4.2 \times 10^{16}$	< 18	[200]	$< 2.2 \times 10^{15}$	< 0.43
gGg'(CH ₂ OH) ₂	CDMS	[200]	$< 2.2 \times 10^{16}$	< 8.9	[200]	$< 2.2 \times 10^{15}$	< 0.43
CH ₂ OHCHO	CDMS	[200]	$< 8.0 \times 10^{15}$	< 3.3	[200]	$< 3.2 \times 10^{14}$	< 0.064
H ₂ CCO	CDMS	–	–	–	[200]	$(1.3 \pm 0.3) \times 10^{14}$	0.03 ± 0.01
t-HCOOH	JPL	[200]	$< 1.0 \times 10^{16}$	< 4.2	[200]	$< 5.0 \times 10^{14}$	< 0.1

Notes. The column density of CH₃OH is determined by scaling ¹³CH₃OH for Band 3 and CH₃¹⁸OH for Band 6 using ¹²C/¹³C = 70 and ¹⁶O/¹⁸O = 560, respectively. A value between squared brackets means the parameter was fixed to that value during the computation. Abundance ratio's which are set to their respective isotope ratio are indicated with a \equiv symbol.

⁽¹⁾ The presented column densities are derived for a 0.45" source size. However, in reality the source sizes will be smaller, and the presented column densities thus represent lower limits.

Table B.2. Number of transitions and lines fitted for all sources.

B1-c						
Species	Band 3			Band 6		
	T	L	F	T	L	F
CH ₃ OH	1	1	0	3	3	0
¹³ CH ₃ OH	2	1	1	8	7	4
CH ₃ ¹⁸ OH	4	0	0	20	16	8
CH ₂ DOH	5	2	2	16	9	6
CH ₃ CH ₂ OH	12	3	3	30	14	8
CH ₃ CHO	17	2	2	27	6	5
CH ₃ OCHO	34	16	13	111	12	7
CH ₃ OCH ₃	16	8	8	112	27	20
CH ₃ COCH ₃	59	0	0	128	6	5
aGg'(CH ₂ OH) ₂	25	0	0	105	18	8
gGg'(CH ₂ OH) ₂	43	0	0	115	4	2
CH ₂ OHCHO	15	0	0	27	5	3
H ₂ CCO	0	0	0	6	2	2
t-HCOOH	6	3	3	3	1	1

S68N						
Species	Band 3			Band 6		
	T	L	F	T	L	F
CH ₃ OH	1	1	0	3	3	0
¹³ CH ₃ OH	2	1	1	8	5	5
CH ₃ ¹⁸ OH	4	0	0	20	2	2
CH ₂ DOH	5	0	0	16	2	2
CH ₃ CH ₂ OH	12	0	0	30	3	2
CH ₃ CHO	17	0	0	27	2	2
CH ₃ OCHO	34	7	7	111	10	7
CH ₃ OCH ₃	16	8	8	112	27	20
CH ₃ COCH ₃	59	0	0	128	2	1
aGg'(CH ₂ OH) ₂	25	0	0	105	1	1
gGg'(CH ₂ OH) ₂	43	0	0	115	0	0
CH ₂ OHCHO	15	0	0	27	0	0
H ₂ CCO	0	0	0	6	2	1
t-HCOOH	6	0	0	3	1	1

B1-bS						
Species	Band 3			Band 6		
	T	L	F	T	L	F
CH ₃ OH	1	1	0	3	3	0
¹³ CH ₃ OH	2	1	1	8	4	3
CH ₃ ¹⁸ OH	4	0	0	20	13	5
CH ₂ DOH	3	0	0	16	0	0
CH ₃ CH ₂ OH	12	0	0	30	0	0
CH ₃ CHO	17	2	2	27	3	2
CH ₃ OCHO	34	8	8	111	10	7
CH ₃ OCH ₃	16	6	6	112	63	60
CH ₃ COCH ₃	59	0	0	128	1	1
aGg'(CH ₂ OH) ₂	25	0	0	0	0	0
gGg'(CH ₂ OH) ₂	43	0	0	0	0	0
CH ₂ OHCHO	15	0	0	0	0	0
H ₂ CCO	0	0	0	6	1	1
t-HCOOH	6	0	0	3	0	0

Notes. T is the number of transitions, L is the number of lines in the model with a 3σ detection, and F is the number of unblended lines (i.e., no other line within one FWHM) that are included in the fit.

Table B.3. Computed column densities and abundance ratios in Band 3, assuming $T_{\text{ex}} = 100$ K.

B1-c			
Species	Band 3		
	T_{ex} (K)	N (cm ⁻²)	X/CH ₃ OH (%)
CH ₃ OH	–	$(6.7 \pm 1.1) \times 10^{17}$	$\equiv 100$
¹³ CH ₃ OH	[100]	$(9.6 \pm 1.6) \times 10^{15}$	$\equiv 1.43$
CH ₃ ¹⁸ OH	[100]	$< 2.2 \times 10^{16}$	< 3.4
CH ₂ DOH	[100]	$(5.7 \pm 0.7) \times 10^{16}$	8.4 ± 1.7
CH ₃ CH ₂ OH	[100]	$(1.8 \pm 0.2) \times 10^{16}$	2.7 ± 0.5
CH ₃ CHO	[100]	$(4.7 \pm 1.6) \times 10^{15}$	0.7 ± 0.3
CH ₃ OCHO	[100]	$(1.6 \pm 0.2) \times 10^{16}$	2.4 ± 0.5
CH ₃ OCH ₃	[100]	$(1.9 \pm 0.1) \times 10^{16}$	2.8 ± 0.5
CH ₃ COCH ₃	[100]	$< 5.0 \times 10^{16}$	< 7.5
aGg'(CH ₂ OH) ₂	[100]	$< 2.1 \times 10^{16}$	< 3.2
gGg'(CH ₂ OH) ₂	[100]	$< 2.1 \times 10^{16}$	< 3.2
CH ₂ OHCHO	[100]	$< 8.0 \times 10^{15}$	< 1.2
t-HCOOH	[100]	$(4.3 \pm 0.7) \times 10^{15}$	0.6 ± 0.2

S68N			
Species	Band 3		
	T_{ex} (K)	N (cm ⁻²)	X/CH ₃ OH (%)
CH ₃ OH	–	$(2.1 \pm 0.7) \times 10^{17}$	$\equiv 100$
¹³ CH ₃ OH	[100]	$(3.0 \pm 1.0) \times 10^{15}$	$\equiv 1.43$
CH ₃ ¹⁸ OH	[100]	$< 2.0 \times 10^{16}$	< 9.8
CH ₂ DOH	[100]	$< 1.0 \times 10^{17}$	< 49
CH ₃ CH ₂ OH	[100]	$< 2.0 \times 10^{16}$	< 9.8
CH ₃ CHO	[100]	$< 2.9 \times 10^{14}$	< 0.14
CH ₃ OCHO	[100]	$(4.7 \pm 0.3) \times 10^{15}$	2.3 ± 0.8
CH ₃ OCH ₃	[100]	$(6.0 \pm 0.3) \times 10^{15}$	2.9 ± 1.0
CH ₃ COCH ₃	[100]	$< 5.0 \times 10^{16}$	< 25
aGg'(CH ₂ OH) ₂	[100]	$< 4.3 \times 10^{16}$	< 21
gGg'(CH ₂ OH) ₂	[100]	$< 4.3 \times 10^{16}$	< 21
CH ₂ OHCHO	[100]	$< 1.6 \times 10^{16}$	< 7.8
t-HCOOH	[100]	$< 5.0 \times 10^{15}$	< 2.5

B1-bS			
Species	Band 3		
	T_{ex} (K)	N (cm ⁻²)	X/CH ₃ OH (%)
CH ₃ OH	–	$(1.1 \pm 0.3) \times 10^{17}$	$\equiv 100$
¹³ CH ₃ OH	[100]	$(1.6 \pm 0.4) \times 10^{15}$	$\equiv 1.43$
CH ₃ ¹⁸ OH	[100]	$< 1.0 \times 10^{16}$	< 8.9
CH ₂ DOH	[100]	$< 2.0 \times 10^{16}$	< 18
CH ₃ CH ₂ OH	[100]	$< 2.0 \times 10^{16}$	< 18
CH ₃ CHO	[100]	$(1.4 \pm 0.2) \times 10^{15}$	1.2 ± 0.3
CH ₃ OCHO	[100]	$(5.7 \pm 0.7) \times 10^{15}$	5.0 ± 1.3
CH ₃ OCH ₃	[100]	$(1.2 \pm 0.1) \times 10^{16}$	11 ± 3
CH ₃ COCH ₃	[100]	$< 1.0 \times 10^{16}$	< 8.9
aGg'(CH ₂ OH) ₂	[100]	$< 1.1 \times 10^{16}$	< 9.6
gGg'(CH ₂ OH) ₂	[100]	$< 2.1 \times 10^{16}$	< 19
CH ₂ OHCHO	[100]	$< 8.0 \times 10^{15}$	< 7.2
t-HCOOH	[100]	$< 1.0 \times 10^{16}$	< 8.9

Notes. The column density of CH₃OH is determined by scaling ¹³CH₃OH using ¹²C/¹³C = 70. A value between squared brackets means the parameter was fixed to that value during the computation. Abundance ratios which are set to their respective isotope ratio are indicated with a \equiv symbol.

Appendix C: Full ALMA Band 6 spectra

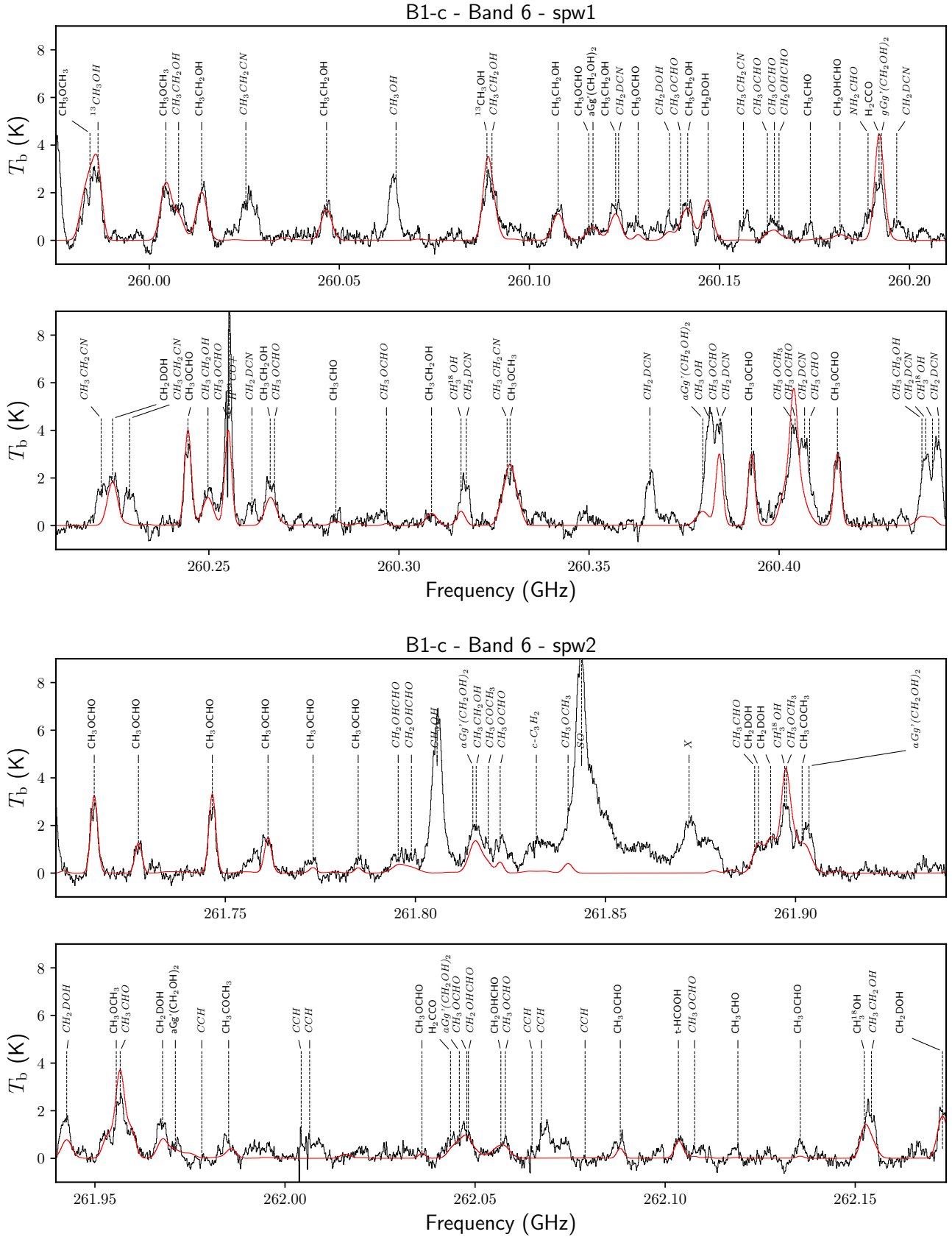


Fig. C.1. Full Band 6 spectrum of B1-c (black) with best fitting CASSIS model overplotted (red). We indicate the positions of species, where lines in *italic* are excluded in the fitting. Lines annotated with an 'X' are unidentified.

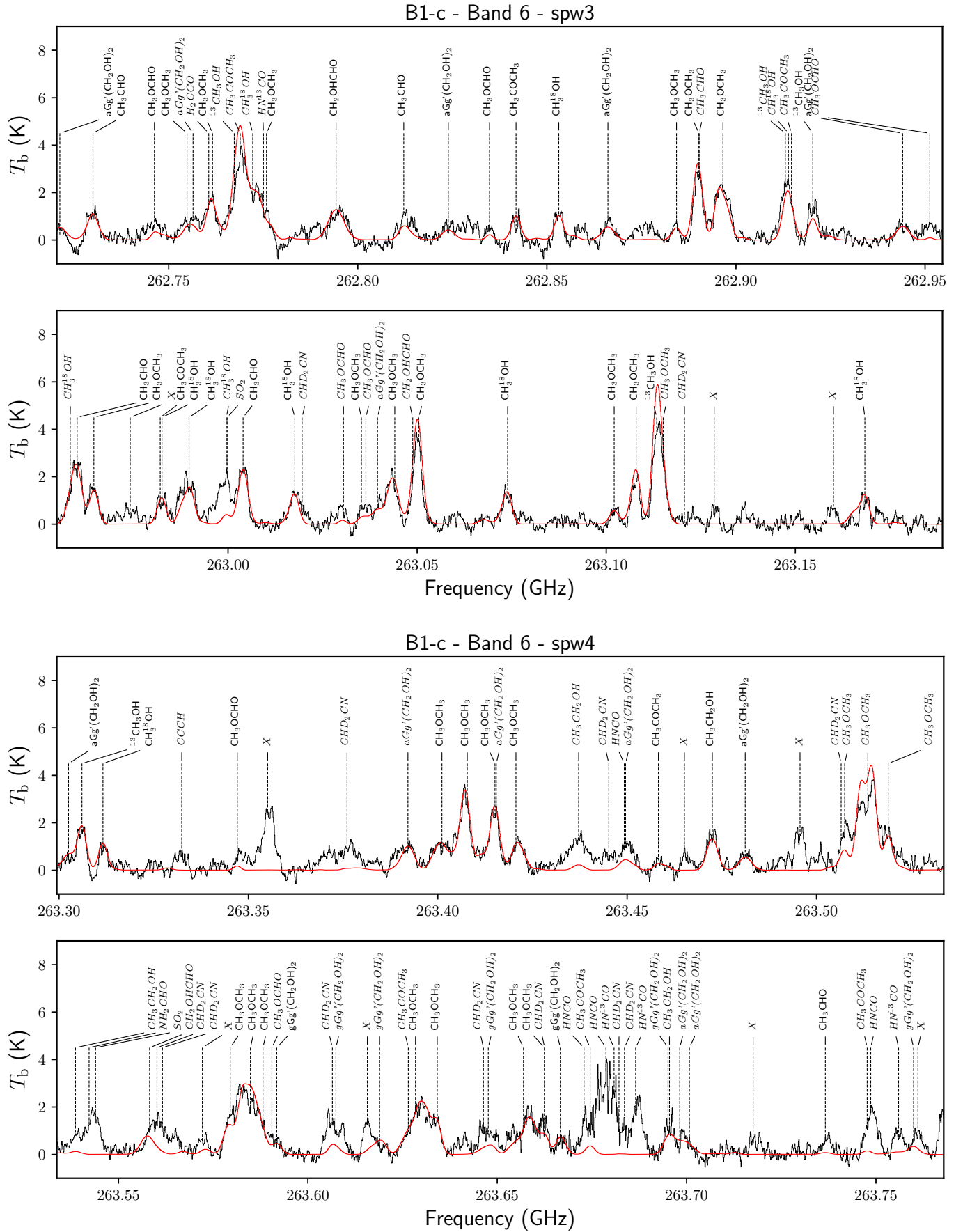


Fig. C.2. (Continued)

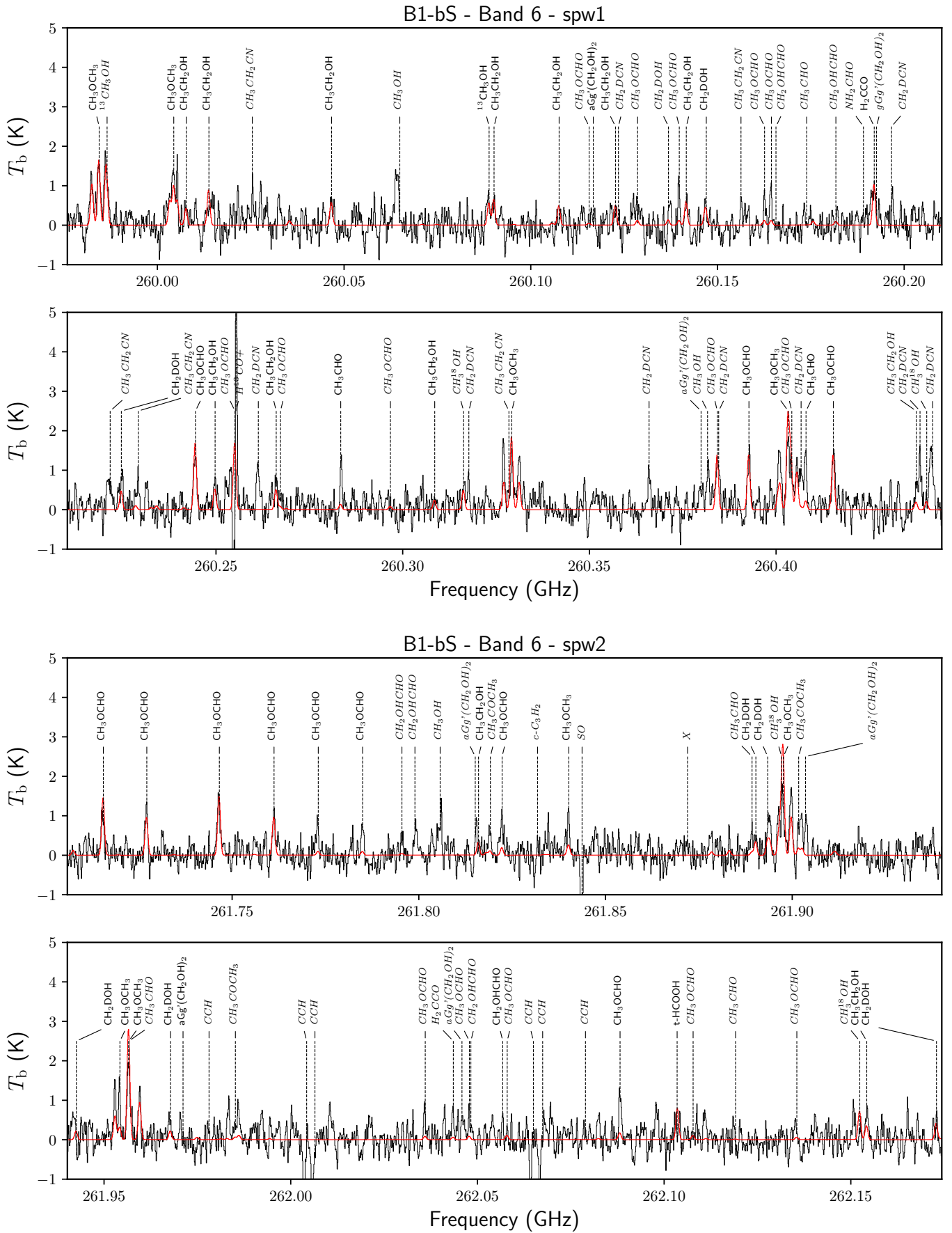


Fig. C.5. Same as Fig. C.6 but now for B1-bS.

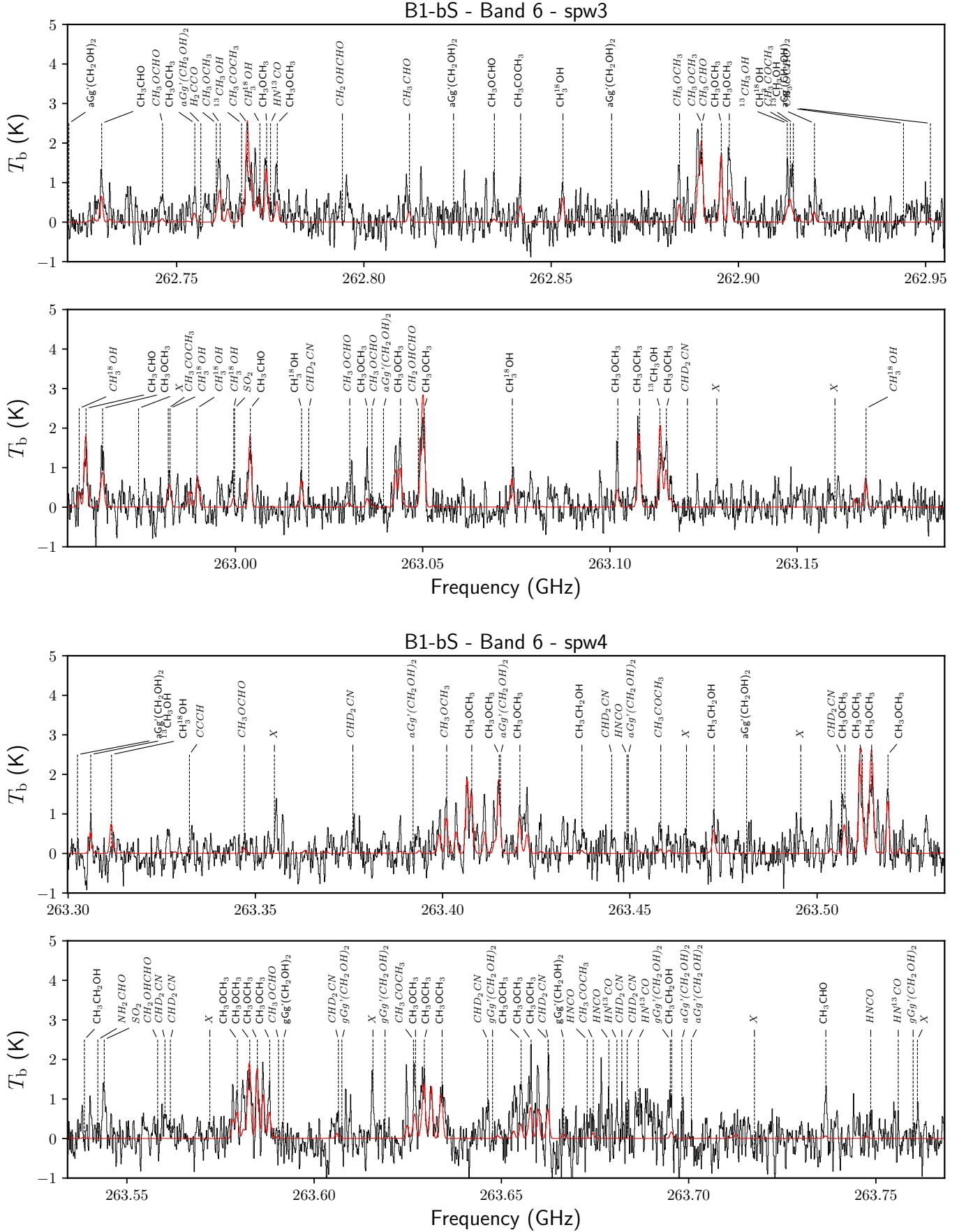


Fig. C.6. (Continued)

Appendix D: B1-c Band 3 spectrum

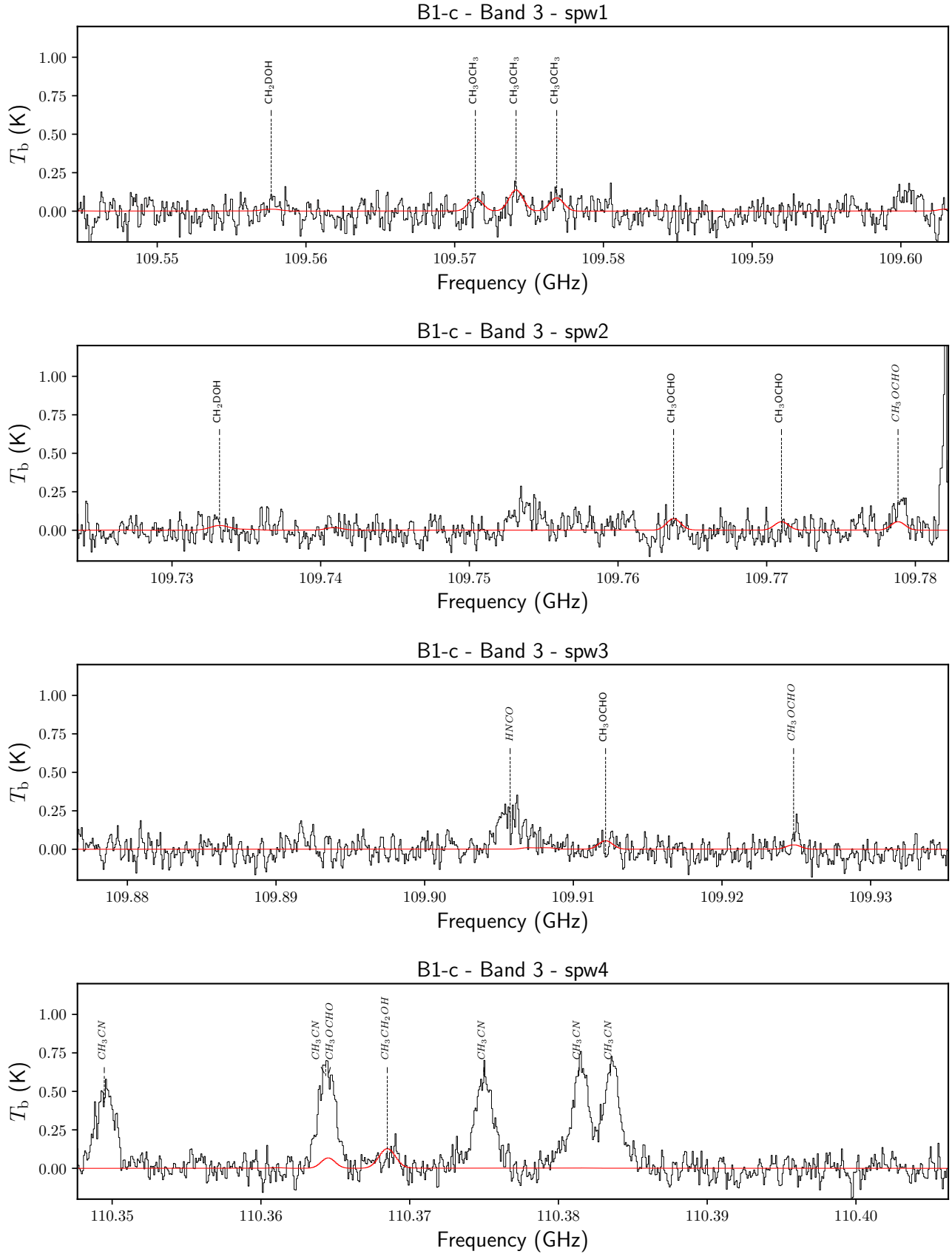


Fig. D.1. Full Band 3 spectrum of B1-c (black) with best fitting CASSIS model overplotted (red). We indicate the positions of species, where lines in italic are excluded in the fitting.

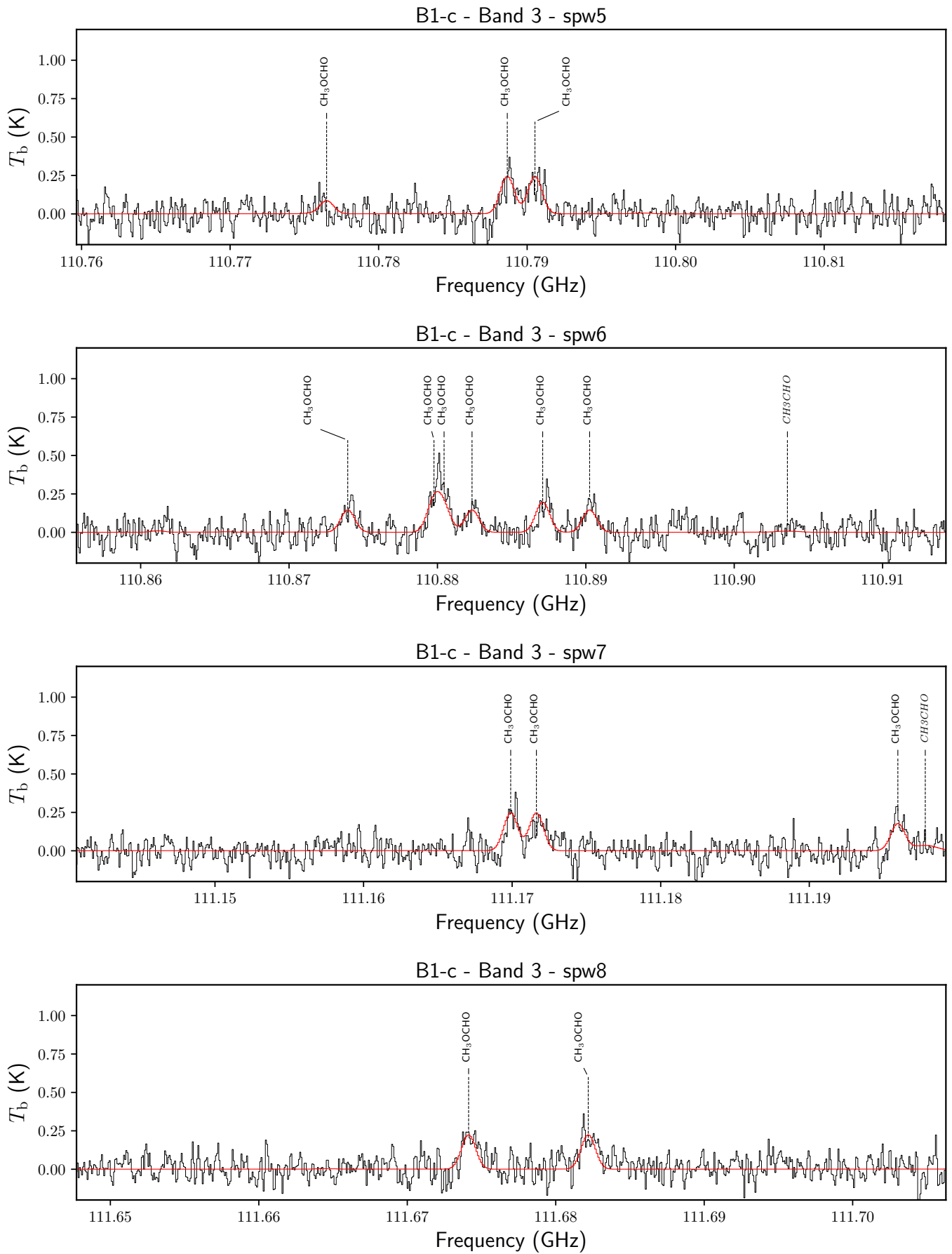


Fig. D.2. (Continued)

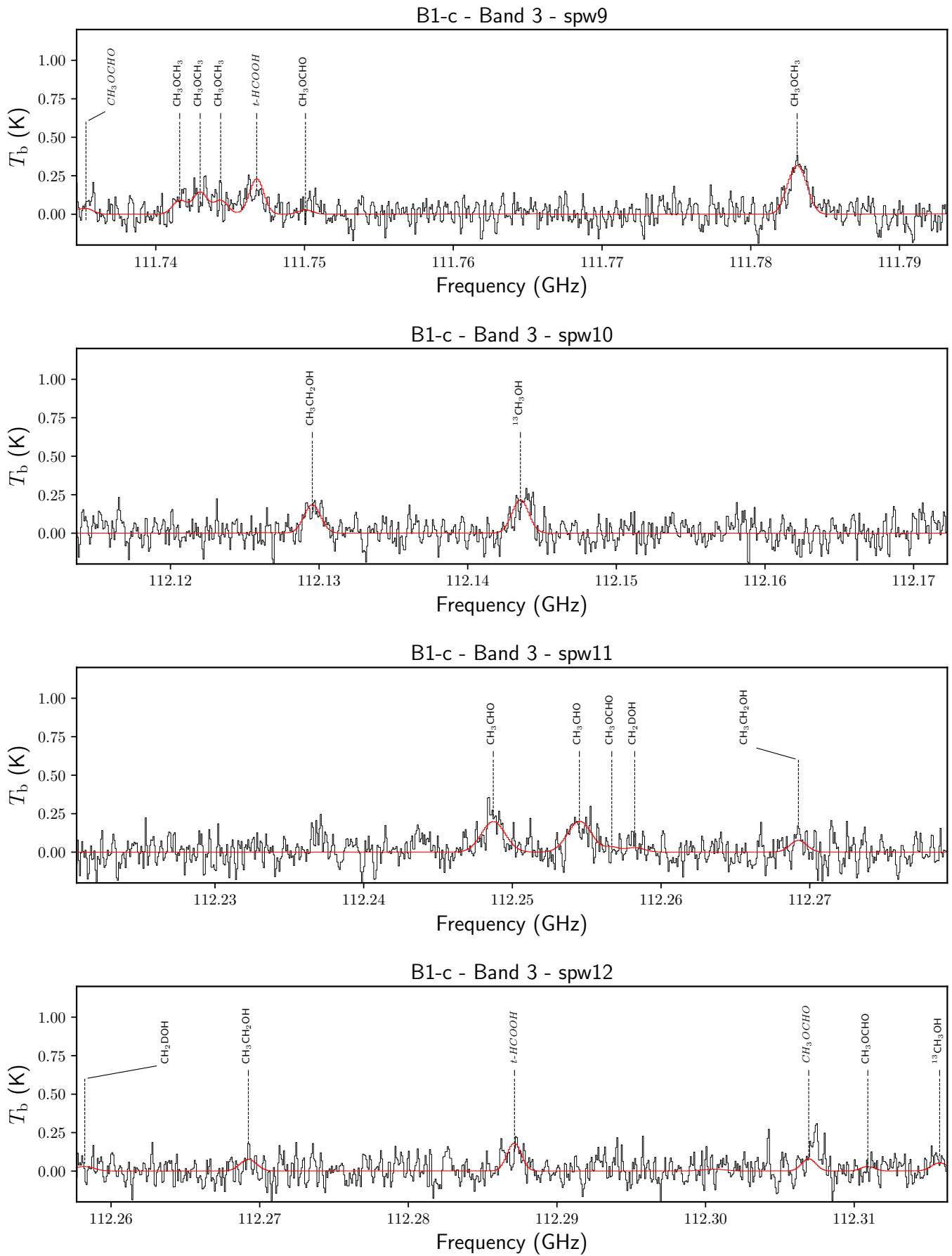


Fig. D.3. (Continued)

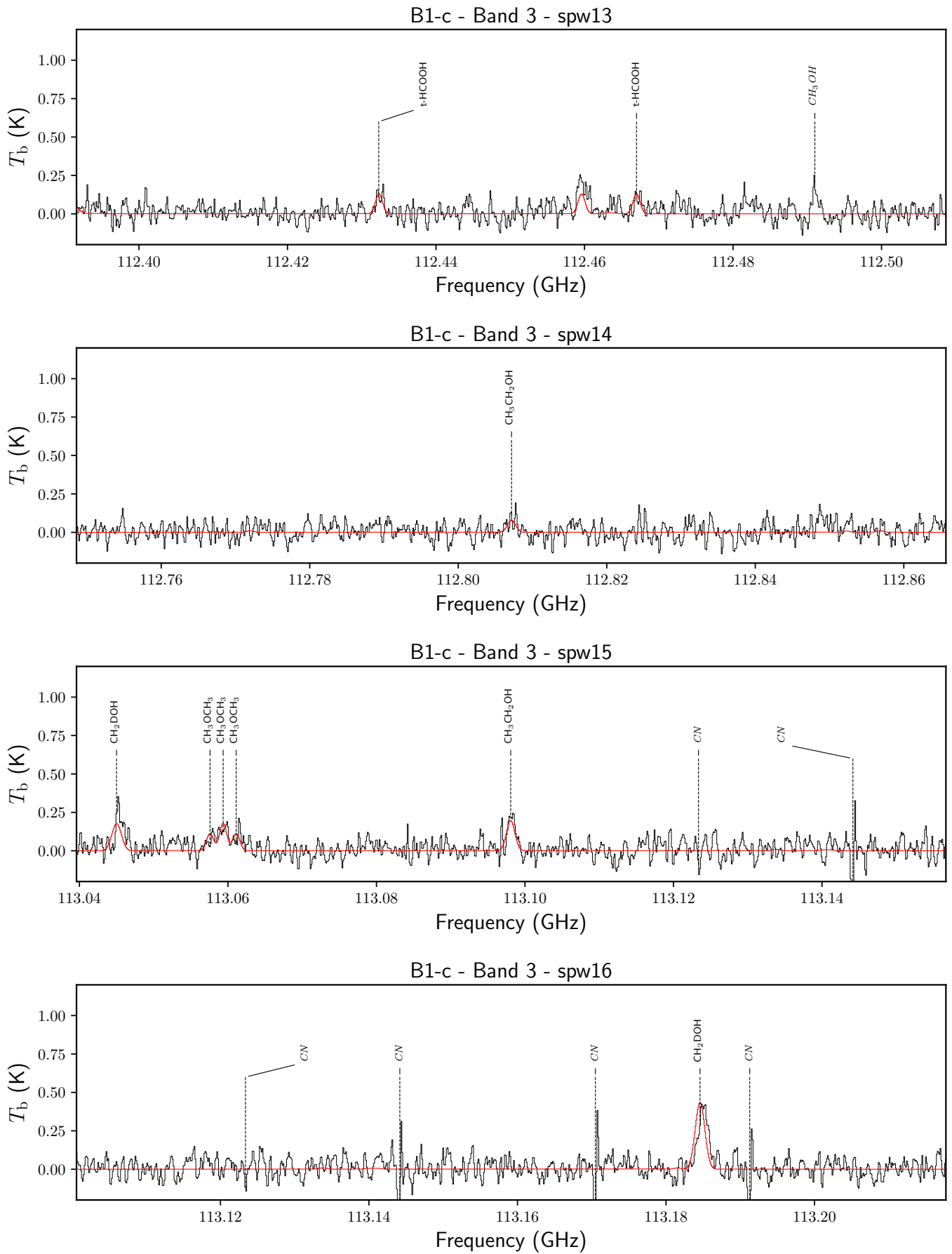


Fig. D.4. (Continued)

Appendix E: Additional tables**Table E.1.** List of ALMA image properties from program 2017.1.01774.S.

	B1-b ¹		B1-c		S68N ²		SMM3	
	Band 3	Band 6	Band 3	Band 6	Band 3	Band 6	Band 3	Band 6
Configuration	C43-3	C43-4	C43-3	C43-4	C43-2	C43-4	C43-2	–
Beam (")	2.1×1.4	0.58×0.39	2.1×1.4	0.58×0.39	2.8×1.8	0.46×0.42	2.8×1.8	–
LAS (")	16	6	16	6	22	6	22	–
ΔV (km s ⁻¹)	0.2 / 0.4	0.14	0.2 / 0.4	0.14	0.17 / 0.32	0.14	0.17 / 0.32	–
Continuum rms (mJy)	8.5	5.4	14.4	24.4	14.6	9.2	5.1	–
Line rms (K)	0.05	0.16	0.06	0.14	0.04	0.15	0.06	–
Flux uncertainty (%)	3	12	3	12	5	9	5	–

Notes. ⁽¹⁾ Including B1-bN and B1-bS. ⁽²⁾ Including Ser-emb 8 (N).

Table E.2. Lines per species used for fitting the Band 3 data.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
$^{13}\text{CH}_3\text{OH}$	3 1 3 +0	- 4 0 4 +0	112.1435	7.1×10^{-6}	28.0
	17 -2 16 0	- 17 1 16 0	112.3157	2.1×10^{-6}	373.0
$\text{CH}_3^{18}\text{OH}$	21 3 18 4	- 20 2 18 4	110.9133	5.0×10^{-6}	834.9
	18 1 17 4	- 19 2 17 4	111.6514	1.1×10^{-6}	790.6
	13 3 11 1	- 14 2 13 1	112.3039	2.6×10^{-6}	265.0
	25 0 25 1	- 25 1 25 1	113.0484	1.2×10^{-6}	730.9
CH_2DOH	28 0 28 2	- 28 1 28 1	109.7332	5.3×10^{-6}	878.3
	12 1 11 1	- 12 0 12 1	113.0450	2.5×10^{-6}	185.1
	10 1 9 2	- 10 0 10 2	113.1846	6.2×10^{-6}	141.6
$\text{CH}_3\text{CH}_2\text{OH}$	20 3 17 0	- 19 4 15 1	109.7409	1.5×10^{-6}	245.0
	8 1 8 1	- 8 0 8 0	110.3685	9.5×10^{-6}	91.7
	28 9 19 1	- 27 10 17 0	110.7978	1.0×10^{-6}	498.5
	28 9 20 1	- 27 10 18 0	110.7978	1.0×10^{-6}	498.5
	12 3 9 2	- 12 2 10 2	112.1295	8.4×10^{-6}	77.1
	9 1 8 0	- 8 0 8 1	112.2692	5.2×10^{-6}	96.5
	11 3 9 1	- 10 4 7 0	112.7463	1.1×10^{-6}	127.5
	2 2 1 2	- 1 1 0 2	112.8072	1.1×10^{-5}	7.5
	45 5 41 1	- 45 4 41 0	112.8196	5.0×10^{-6}	954.0
	28 5 24 0	- 27 6 22 1	113.0453	1.6×10^{-6}	426.0
	10 2 9 2	- 10 1 10 2	113.0981	8.8×10^{-6}	51.0
	31 1 31 0	- 30 2 29 1	113.1338	2.3×10^{-6}	449.9
CH_3CHO	17 4 13 6	- 18 3 16 6	109.5448	1.5×10^{-6}	570.8
	20 6 14 2	- 21 5 16 2	109.7350	1.2×10^{-6}	275.5
	34 9 25 3	- 35 8 28 3	109.7646	1.2×10^{-6}	937.4
	34 9 26 3	- 35 8 27 3	109.7657	1.2×10^{-6}	937.4
	23 3 21 1	- 22 4 18 2	110.9036	1.3×10^{-6}	275.9
	13 1 12 0	- 13 0 13 0	111.1978	5.0×10^{-6}	88.4
	15 1 15 7	- 14 1 13 8	111.7924	2.2×10^{-6}	489.6
	18 2 16 8	- 18 2 17 7	112.1263	2.4×10^{-6}	555.7
	6 1 6 0	- 5 1 5 0	112.2487	4.5×10^{-5}	21.1
	6 1 6 1	- 5 1 5 1	112.2545	4.5×10^{-5}	21.2
	6 1 5 6	- 6 0 6 6	112.3004	9.9×10^{-6}	392.2
	6 1 6 3	- 5 1 5 3	112.3908	4.6×10^{-5}	227.6
	14 5 10 4	- 15 4 11 5	112.3939	1.1×10^{-6}	358.6
	15 5 11 0	- 16 4 12 0	112.4634	1.2×10^{-6}	167.5
	27 3 24 8	- 27 3 25 7	112.8593	1.7×10^{-6}	755.2
	6 1 6 6	- 5 1 5 6	113.0365	4.4×10^{-5}	391.0
	15 5 10 2	- 16 4 13 1	113.1407	1.2×10^{-6}	167.4
CH_3OCHO	10 0 10 5	- 9 1 9 4	109.6030	2.7×10^{-6}	217.2
	9 6 3 3	- 8 6 2 3	109.6085	1.1×10^{-5}	238.2
	9 6 4 3	- 8 6 3 3	109.6085	1.1×10^{-5}	238.2
	9 3 7 3	- 8 3 6 3	109.7637	1.7×10^{-5}	220.2
	9 5 5 3	- 8 5 4 3	109.7710	1.3×10^{-5}	230.8
	9 5 4 3	- 8 5 3 3	109.7788	1.3×10^{-5}	230.8
	15 4 12 3	- 15 3 13 3	109.9070	1.9×10^{-6}	269.0
	28 6 22 5	- 28 5 23 5	109.9082	2.7×10^{-6}	455.2
	9 5 4 5	- 8 5 3 5	109.9122	1.3×10^{-5}	230.6
	9 7 3 4	- 8 7 2 4	109.9248	7.7×10^{-6}	246.1
	9 4 5 5	- 8 4 4 5	110.3645	1.6×10^{-5}	224.6
	9 1 8 3	- 8 1 7 3	110.7765	1.9×10^{-5}	215.7
	10 1 10 1	- 9 1 9 1	110.7887	2.0×10^{-5}	30.3
	10 1 10 0	- 9 1 9 0	110.7905	2.0×10^{-5}	30.3
	15 4 12 4	- 15 3 13 4	110.8612	1.8×10^{-6}	268.6
	9 5 4 2	- 8 5 3 2	110.8740	1.4×10^{-5}	43.2
	9 3 7 1	- 8 3 6 1	110.8798	1.8×10^{-5}	32.6
9 5 5 0	- 8 5 4 0	110.8804	1.4×10^{-5}	43.2	

Table E.2. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	9 5 5 1	- 8 5 4 1	110.8823	1.4×10^{-5}	43.2
	9 3 7 0	- 8 3 6 0	110.8871	1.8×10^{-5}	32.6
	9 5 4 0	- 8 5 3 0	110.8903	1.4×10^{-5}	43.2
	10 0 10 2	- 9 0 9 2	111.1699	2.0×10^{-5}	30.2
	10 0 10 0	- 9 0 9 0	111.1716	2.0×10^{-5}	30.2
	9 4 6 0	- 8 4 5 0	111.1960	1.6×10^{-5}	37.2
	9 1 8 2	- 8 1 7 2	111.6741	2.0×10^{-5}	28.1
	9 1 8 0	- 8 1 7 0	111.6822	2.0×10^{-5}	28.1
	23 6 17 3	- 23 5 18 3	111.7071	2.6×10^{-6}	375.2
	10 1 10 1	- 9 0 9 2	111.7340	3.0×10^{-6}	30.3
	10 1 10 0	- 9 0 9 0	111.7353	3.0×10^{-6}	30.3
	28 6 22 0	- 28 5 23 0	111.7501	2.8×10^{-6}	269.8
	25 5 20 2	- 25 4 21 2	112.2567	2.5×10^{-6}	214.4
	9 3 6 3	- 8 3 5 3	112.3069	1.8×10^{-5}	220.5
	25 5 20 0	- 25 4 21 0	112.3109	2.5×10^{-6}	214.4
	16 2 14 3	- 16 1 15 3	113.1815	1.5×10^{-6}	272.4
CH ₃ OCH ₃	8 2 7 3	- 8 1 8 3	109.5714	5.4×10^{-6}	38.3
	8 2 7 5	- 8 1 8 5	109.5714	5.4×10^{-6}	38.3
	8 2 7 1	- 8 1 8 1	109.5741	5.4×10^{-6}	38.3
	8 2 7 0	- 8 1 8 0	109.5769	5.4×10^{-6}	38.3
	19 3 16 3	- 19 2 17 3	111.7416	1.1×10^{-5}	187.5
	19 3 16 5	- 19 2 17 5	111.7416	1.1×10^{-5}	187.5
	19 3 16 1	- 19 2 17 1	111.7430	1.1×10^{-5}	187.5
	19 3 16 0	- 19 2 17 0	111.7444	1.1×10^{-5}	187.5
	7 0 7 0	- 6 1 6 0	111.7826	7.4×10^{-6}	25.2
	7 0 7 1	- 6 1 6 1	111.7831	7.4×10^{-6}	25.2
	7 0 7 5	- 6 1 6 5	111.7836	7.4×10^{-6}	25.3
	7 0 7 3	- 6 1 6 3	111.7836	7.4×10^{-6}	25.3
	17 3 14 5	- 17 2 15 5	113.0576	1.1×10^{-5}	153.1
	17 3 14 3	- 17 2 15 3	113.0576	1.1×10^{-5}	153.1
	17 3 14 1	- 17 2 15 1	113.0594	1.1×10^{-5}	153.1
	17 3 14 0	- 17 2 15 0	113.0611	1.1×10^{-5}	153.1
CH ₃ COCH ₃	32 23 10 1	- 32 22 11 1	109.6011	3.5×10^{-5}	444.0
	32 23 10 1	- 32 22 11 2	109.6030	3.4×10^{-5}	444.0
	9 6 3 1	- 8 7 2 2	109.9047	1.4×10^{-6}	37.3
	26 21 6 0	- 26 20 7 1	110.3754	2.6×10^{-5}	306.6
	29 22 7 0	- 29 21 8 1	110.3788	2.8×10^{-5}	371.8
	29 22 7 0	- 29 21 8 0	110.3828	3.1×10^{-5}	371.8
	26 21 5 0	- 26 20 6 1	110.3933	2.6×10^{-5}	306.6
	6 5 2 0	- 5 4 1 0	110.4013	3.4×10^{-5}	18.2
	8 4 5 0	- 7 3 4 1	110.4094	3.1×10^{-5}	26.8
	22 20 3 0	- 22 19 4 1	110.7627	1.6×10^{-5}	232.1
	42 25 17 0	- 42 24 18 1	110.7636	5.9×10^{-5}	732.4
	47 31 16 1	- 47 30 17 1	110.7727	4.4×10^{-5}	928.6
	38 25 14 0	- 38 24 15 1	110.8606	5.2×10^{-5}	609.5
	8 4 4 0	- 7 5 3 0	110.9171	1.6×10^{-5}	28.4
	8 4 4 1	- 7 5 3 2	111.1413	1.6×10^{-5}	28.6
	37 25 13 0	- 37 24 14 1	111.6404	4.9×10^{-5}	581.8
	37 25 13 1	- 37 24 14 2	111.6834	4.1×10^{-5}	581.7
	14 10 4 0	- 14 7 7 1	111.6931	1.8×10^{-6}	88.0
	24 11 13 1	- 24 10 14 1	111.7369	3.2×10^{-5}	230.4
	24 11 13 1	- 24 10 14 2	111.7372	3.2×10^{-5}	230.4
	14 10 4 0	- 14 7 7 0	111.7505	1.9×10^{-6}	87.9
	24 12 13 1	- 24 11 14 1	111.7591	3.2×10^{-5}	230.4
	24 12 13 1	- 24 11 14 2	111.7593	3.2×10^{-5}	230.4
	24 11 13 0	- 24 10 14 1	111.7973	3.2×10^{-5}	230.4

Table E.2. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	37 25 13 1	- 37 24 14 1	111.7996	4.2×10^{-5}	581.7
	34 24 11 0	- 34 23 12 1	112.1478	4.1×10^{-5}	497.9
	31 18 14 0	- 31 17 15 0	112.1507	3.9×10^{-5}	396.7
	34 24 11 0	- 34 23 12 0	112.1546	3.9×10^{-5}	497.9
	31 18 14 0	- 31 17 15 1	112.1592	4.0×10^{-5}	396.6
	31 18 14 1	- 31 17 15 2	112.1677	3.9×10^{-5}	396.6
	31 18 14 1	- 31 17 15 1	112.1684	3.9×10^{-5}	396.6
	45 27 18 1	- 45 26 19 1	112.1796	4.6×10^{-5}	840.2
	21 20 1 0	- 21 19 2 1	112.2312	1.1×10^{-5}	216.8
	21 20 2 0	- 21 19 3 1	112.3046	1.2×10^{-5}	216.7
	45 30 15 0	- 45 29 16 0	112.4518	4.5×10^{-5}	854.0
	6 5 1 1	- 5 4 1 2	112.4951	1.4×10^{-5}	18.5
	38 24 15 0	- 38 23 16 0	112.5018	4.4×10^{-5}	604.2
	23 10 13 1	- 23 9 14 1	112.7549	3.1×10^{-5}	209.3
	23 10 13 1	- 23 9 14 2	112.7551	3.1×10^{-5}	209.3
	23 11 13 1	- 23 10 14 1	112.7621	3.1×10^{-5}	209.3
	23 11 13 1	- 23 10 14 2	112.7622	3.1×10^{-5}	209.3
	25 21 4 0	- 25 20 5 1	112.8283	2.4×10^{-5}	288.2
	23 10 13 0	- 23 9 14 1	112.8310	3.2×10^{-5}	209.3
	31 23 8 1	- 31 22 9 2	112.8320	2.7×10^{-5}	421.0
	23 11 13 0	- 23 10 14 1	112.8382	3.2×10^{-5}	209.3
	25 21 5 0	- 25 20 6 1	112.8397	2.4×10^{-5}	288.2
	31 23 8 0	- 31 22 9 1	112.8422	2.7×10^{-5}	421.0
	45 27 18 0	- 45 26 19 0	112.8627	4.7×10^{-5}	840.3
	31 23 8 0	- 31 22 9 0	112.8694	3.5×10^{-5}	421.1
	37 23 15 0	- 37 22 16 1	113.0468	5.3×10^{-5}	571.2
	25 21 4 0	- 25 20 5 0	113.0648	2.4×10^{-5}	288.3
	25 21 5 0	- 25 20 6 0	113.0707	2.4×10^{-5}	288.3
	39 25 15 0	- 39 24 16 0	113.0911	4.5×10^{-5}	638.0
	37 23 15 1	- 37 22 16 2	113.1143	4.4×10^{-5}	571.2
	37 23 15 1	- 37 22 16 1	113.1226	4.4×10^{-5}	571.2
	20 20 0 1	- 20 19 1 2	113.1584	6.2×10^{-6}	202.2
	45 30 15 0	- 45 29 16 1	113.2046	6.6×10^{-5}	854.0
	10 7 3 0	- 9 8 1 1	113.2057	1.2×10^{-6}	45.9
	31 23 9 1	- 31 22 10 2	113.2179	2.7×10^{-5}	421.0
aGg'(CH ₂ OH) ₂	18 6 13 0	- 18 5 14 0	109.5468	3.6×10^{-6}	102.0
	45 9 36 1	- 45 8 37 1	109.7212	5.9×10^{-6}	555.5
	25 7 19 1	- 24 8 16 1	109.7295	1.0×10^{-6}	184.8
	27 7 20 1	- 27 6 21 1	109.7364	4.3×10^{-6}	211.3
	23 6 18 1	- 22 7 15 1	109.7545	1.2×10^{-6}	154.4
	10 1 9 1	- 9 1 8 0	109.7605	2.4×10^{-5}	28.5
	16 6 11 0	- 16 5 12 0	110.4055	3.5×10^{-6}	84.7
	15 6 10 1	- 15 5 11 1	110.8540	3.8×10^{-6}	77.1
	23 4 20 1	- 23 3 21 1	111.6959	2.7×10^{-6}	144.1
	11 6 5 1	- 11 5 6 1	112.1232	3.4×10^{-6}	50.5
	11 6 6 0	- 11 5 7 0	112.1340	3.1×10^{-6}	50.1
	11 6 6 1	- 11 5 7 1	112.1506	3.4×10^{-6}	50.5
	46 9 37 0	- 46 8 38 0	112.2790	6.0×10^{-6}	578.7
	10 6 4 0	- 10 5 5 0	112.3116	2.9×10^{-6}	44.7
	9 6 3 0	- 9 5 4 0	112.4602	2.7×10^{-6}	39.8
	9 6 4 0	- 9 5 5 0	112.4637	2.7×10^{-6}	39.8
	9 6 3 1	- 9 5 4 1	112.4783	2.9×10^{-6}	40.1
	9 6 4 1	- 9 5 5 1	112.4818	2.9×10^{-6}	40.1
	25 6 20 0	- 25 5 21 0	112.8157	4.5×10^{-6}	178.4
	10 2 9 1	- 9 1 8 1	112.8355	1.2×10^{-6}	29.0
	10 2 9 0	- 9 1 8 0	113.0314	1.1×10^{-5}	28.7

Table E.2. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	27 7 21 1	- 27 6 21 0	113.0501	2.6×10^{-6}	211.2
	19 1 18 1	- 19 0 19 1	113.1029	1.3×10^{-6}	92.8
	42 9 33 1	- 42 8 34 1	113.1433	6.0×10^{-6}	488.4
	19 1 18 0	- 19 0 19 0	113.1609	1.3×10^{-6}	92.4
gGg'(CH ₂ OH) ₂	46 9 37 0	- 46 8 38 0	109.5531	1.1×10^{-5}	574.1
	39 7 32 1	- 39 6 33 1	109.5776	9.8×10^{-6}	412.0
	46 9 37 1	- 46 8 38 1	109.6094	1.1×10^{-5}	574.1
	35 8 27 0	- 35 7 28 0	109.7473	9.8×10^{-6}	341.1
	24 6 19 0	- 24 5 20 0	109.7560	8.3×10^{-6}	164.8
	24 6 19 1	- 24 5 20 1	109.7601	8.0×10^{-6}	164.8
	27 8 20 1	- 26 9 18 0	109.8720	1.9×10^{-6}	216.3
	27 7 20 0	- 27 6 21 0	109.8996	8.2×10^{-6}	209.4
	15 6 10 1	- 15 5 10 0	110.3408	7.0×10^{-6}	76.2
	13 6 8 1	- 13 5 9 1	110.3980	6.9×10^{-6}	62.0
	25 7 18 1	- 24 8 17 1	110.4062	2.0×10^{-6}	183.2
	19 2 18 1	- 19 1 19 1	110.7957	2.4×10^{-6}	92.0
	19 2 18 0	- 19 1 19 0	110.7993	2.6×10^{-6}	91.9
	11 6 5 1	- 11 5 6 1	110.8835	6.5×10^{-6}	49.8
	11 6 5 0	- 11 5 6 0	110.8971	6.4×10^{-6}	49.7
	11 5 7 0	- 10 5 6 1	110.9012	9.7×10^{-6}	44.4
	11 6 6 1	- 11 5 7 1	110.9086	6.5×10^{-6}	49.8
	15 6 9 1	- 15 5 11 0	110.9138	5.7×10^{-6}	76.2
	11 5 6 0	- 10 5 5 1	110.9170	9.7×10^{-6}	44.4
	6 3 3 1	- 5 2 3 0	111.7594	6.9×10^{-6}	14.7
	11 6 6 1	- 11 5 6 0	112.1434	5.8×10^{-6}	49.8
	12 0 12 0	- 11 0 11 1	112.1530	9.4×10^{-6}	36.1
	40 9 32 0	- 40 8 32 1	112.1613	5.8×10^{-6}	442.2
	11 6 5 1	- 11 5 7 0	112.1696	5.8×10^{-6}	49.8
	20 4 17 1	- 19 5 15 0	112.2534	2.4×10^{-6}	110.7
	7 2 6 1	- 6 1 6 0	112.4572	5.4×10^{-6}	15.6
	9 6 4 1	- 9 5 4 0	112.5023	5.2×10^{-6}	39.5
	9 6 3 1	- 9 5 5 0	112.5055	5.2×10^{-6}	39.5
	6 6 1 1	- 6 5 1 0	112.7433	2.3×10^{-6}	27.8
	6 6 0 1	- 6 5 2 0	112.7434	2.3×10^{-6}	27.8
	18 4 14 1	- 17 5 13 1	112.7800	1.5×10^{-6}	92.6
	8 1 7 1	- 7 0 7 0	112.8301	5.7×10^{-6}	18.7
	11 8 3 1	- 10 8 2 0	113.0460	6.1×10^{-6}	63.4
	11 8 4 1	- 10 8 3 0	113.0460	6.1×10^{-6}	63.4
	23 6 17 0	- 22 7 16 0	113.0973	2.4×10^{-6}	153.1
	12 0 12 1	- 11 1 11 1	113.1070	9.2×10^{-6}	36.2
	23 6 17 1	- 22 7 16 1	113.1093	2.4×10^{-6}	153.1
	12 0 12 0	- 11 1 11 0	113.1156	1.6×10^{-5}	36.1
	11 7 4 1	- 10 7 3 0	113.1237	7.7×10^{-6}	56.1
	11 7 5 1	- 10 7 4 0	113.1237	7.7×10^{-6}	56.1
	26 6 21 0	- 26 5 22 0	113.1609	9.0×10^{-6}	189.9
	26 6 21 1	- 26 5 22 1	113.1851	8.9×10^{-6}	190.0
	47 9 38 0	- 47 8 39 0	113.2009	1.2×10^{-5}	597.8
CH ₂ OHCHO	4 3 1	- 3 2 2	109.8771	2.3×10^{-5}	11.0
	15 3 13	- 15 2 14	110.4025	1.6×10^{-5}	71.8
	45 26 19	- 46 25 22	110.4059	2.2×10^{-6}	984.4
	45 26 20	- 46 25 21	110.4059	2.2×10^{-6}	984.4
	49 18 32	- 48 19 29	111.6495	4.3×10^{-6}	878.4
	49 18 31	- 48 19 30	111.6495	4.3×10^{-6}	878.4
	10 5 5	- 10 4 6	111.7349	2.0×10^{-5}	45.8
	34 21 13	- 35 20 16	112.1106	2.0×10^{-6}	597.9
	34 21 14	- 35 20 15	112.1106	2.0×10^{-6}	597.9

Table E.2. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	13 5 9	- 13 4 10	112.1777	2.3×10^{-5}	65.9
	20 6 14	- 20 5 15	112.2480	2.8×10^{-5}	139.6
	11 5 7	- 11 4 8	112.4443	2.2×10^{-5}	51.9
	14 5 10	- 14 4 11	112.4728	2.4×10^{-5}	73.8
	10 5 6	- 10 4 7	112.7734	2.1×10^{-5}	45.8
	9 5 5	- 9 4 6	113.1188	2.0×10^{-5}	40.2
t-HCOOH	5 0 5	- 4 0 4	111.7468	1.4×10^{-5}	16.1
	5 2 4	- 4 2 3	112.2871	1.2×10^{-5}	28.9
	5 4 2	- 4 4 1	112.4323	5.3×10^{-6}	67.1
	5 4 1	- 4 4 0	112.4323	5.3×10^{-6}	67.1
	5 3 3	- 4 3 2	112.4596	9.4×10^{-6}	44.8
	5 3 2	- 4 3 1	112.4670	9.4×10^{-6}	44.8

Table E.3. Lines per species used for fitting the Band 6 data.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)	
	(J K L M)	(J K L M)				
CH ₃ OH	20 8 13 2	- 21 7 14 2	260.0643	2.1×10^{-5}	808.2	
	20 3 18 0	- 20 2 19 0	260.3815	9.1×10^{-5}	536.9	
	2 1 1 1	- 1 0 1 1	261.8057	5.6×10^{-5}	28.0	
¹³ CH ₃ OH	2 1 1 0	- 1 0 1 0	259.9865	5.5×10^{-5}	27.9	
	18 3 16 +0	- 18 2 17 -0	260.0888	9.2×10^{-5}	437.2	
	20 3 18 +0	- 20 2 19 -0	262.7673	9.4×10^{-5}	525.4	
	7 4 4 -0	- 8 3 5 -0	262.9130	1.7×10^{-5}	144.2	
	7 4 3 +0	- 8 3 6 +0	262.9203	1.7×10^{-5}	144.2	
	5 2 3 0	- 4 1 3 0	263.1133	7.4×10^{-5}	56.3	
	11 2 10 -0	- 10 3 7 -0	263.3060	2.9×10^{-5}	187.3	
	25 -7 18 0	- 26 -6 20 0	263.5779	2.8×10^{-5}	996.5	
CH ₃ ¹⁸ OH	12 2 10 4	- 13 3 10 4	260.0355	7.1×10^{-5}	537.2	
	16 3 13 0	- 16 2 14 0	260.3163	9.3×10^{-5}	352.6	
	15 1 15 1	- 14 2 13 1	260.4403	2.8×10^{-5}	272.2	
	12 3 9 0	- 12 2 10 0	261.8972	9.2×10^{-5}	223.6	
	11 3 8 0	- 11 2 9 0	262.1524	9.2×10^{-5}	196.9	
	7 3 4 0	- 7 2 5 0	262.7722	8.6×10^{-5}	112.4	
	6 3 3 0	- 6 2 4 0	262.8532	8.3×10^{-5}	96.8	
	5 3 2 0	- 5 2 3 0	262.9139	7.7×10^{-5}	83.5	
	4 3 1 0	- 4 2 2 0	262.9583	6.8×10^{-5}	72.3	
	5 3 3 0	- 5 2 4 0	262.9825	7.7×10^{-5}	83.5	
	4 3 2 0	- 4 2 3 0	262.9878	6.8×10^{-5}	72.3	
	3 3 0 0	- 3 2 1 0	262.9897	4.8×10^{-5}	63.4	
	6 3 4 0	- 6 2 5 0	262.9900	8.3×10^{-5}	96.8	
	3 3 1 0	- 3 2 2 0	262.9995	4.8×10^{-5}	63.4	
	7 3 5 0	- 7 2 6 0	263.0177	8.6×10^{-5}	112.4	
	8 3 6 0	- 8 2 7 0	263.0739	8.9×10^{-5}	130.2	
	9 3 7 0	- 9 2 8 0	263.1684	9.0×10^{-5}	150.2	
	10 3 8 0	- 10 2 9 0	263.3116	9.2×10^{-5}	172.5	
	16 5 12 4	- 17 6 12 4	263.4603	5.8×10^{-5}	846.9	
	11 3 9 0	- 11 2 10 0	263.5151	9.2×10^{-5}	196.9	
CH ₂ DOH	25 5 20 2	- 24 6 19 2	260.0710	3.7×10^{-6}	811.2	
	25 5 21 2	- 24 6 18 2	260.0956	3.7×10^{-6}	811.2	
	28 2 26 1	- 28 1 27 1	260.1369	3.5×10^{-5}	902.1	
	16 6 10 2	- 17 5 12 1	260.1467	1.1×10^{-5}	448.1	
	16 6 11 2	- 17 5 13 1	260.1470	1.1×10^{-5}	448.1	
	10 6 5 1	- 11 5 7 2	260.2246	7.1×10^{-6}	269.6	
	10 6 4 1	- 11 5 6 2	260.2247	7.1×10^{-6}	269.6	
	24 6 19 1	- 23 7 16 0	260.4056	1.2×10^{-5}	793.9	
	24 6 18 1	- 23 7 17 0	260.4067	1.2×10^{-5}	793.9	
	22 2 20 1	- 22 0 22 2	261.7345	1.0×10^{-6}	571.6	
	21 5 17 1	- 20 6 14 0	261.8903	2.7×10^{-5}	602.5	
	21 5 16 1	- 20 6 15 0	261.8934	2.7×10^{-5}	602.5	
	18 4 14 2	- 17 5 12 1	261.9426	9.3×10^{-6}	448.2	
	18 4 15 2	- 17 5 13 1	261.9678	9.4×10^{-6}	448.2	
	10 1 9 1	- 9 2 8 1	262.1729	6.5×10^{-6}	135.5	
	30 0 30 0	- 29 2 28 2	262.7597	1.3×10^{-6}	978.8	
	CH ₃ CH ₂ OH	36 7 29 2	- 35 8 28 2	259.9564	3.4×10^{-5}	623.2
		25 2 24 2	- 25 1 25 2	260.0077	1.1×10^{-4}	272.8
15 6 10 0		- 14 6 9 0	260.0138	1.3×10^{-4}	200.5	
15 6 9 0		- 14 6 8 0	260.0138	1.3×10^{-4}	200.5	
15 3 13 1		- 14 3 12 1	260.0466	1.5×10^{-4}	172.5	
15 1 14 0		- 14 1 13 0	260.0902	1.6×10^{-4}	159.6	
15 5 11 1		- 14 5 10 1	260.1076	1.4×10^{-4}	192.1	
15 5 10 1		- 14 5 9 1	260.1227	1.4×10^{-4}	192.1	
15 3 13 0		- 14 3 12 0	260.1416	1.5×10^{-4}	167.7	

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	15 5 11 0	- 14 5 10 0	260.2498	1.4×10^{-4}	187.2
	15 5 10 0	- 14 5 9 0	260.2661	1.4×10^{-4}	187.2
	13 2 11 1	- 13 1 13 0	260.3086	6.2×10^{-5}	143.3
	7 1 6 1	- 6 0 6 0	260.4376	6.7×10^{-5}	86.7
	28 3 26 2	- 28 2 27 2	261.8160	1.1×10^{-4}	351.0
	34 4 31 2	- 33 5 28 2	261.9750	4.1×10^{-5}	520.7
	23 16 7 0	- 24 15 9 1	262.0787	4.1×10^{-6}	599.0
	23 16 8 0	- 24 15 10 1	262.0787	4.1×10^{-6}	599.0
	13 2 12 2	- 12 1 11 2	262.1543	6.7×10^{-5}	81.0
	43 5 38 2	- 43 4 39 2	262.8568	1.1×10^{-4}	840.5
	35 3 32 1	- 35 3 33 1	263.0739	2.4×10^{-6}	604.2
	22 2 20 0	- 22 1 22 1	263.1656	6.4×10^{-5}	276.2
	43 11 33 1	- 42 12 31 0	263.4075	1.7×10^{-5}	996.0
	43 11 32 1	- 42 12 30 0	263.4080	1.7×10^{-5}	996.0
	32 4 29 2	- 31 5 26 2	263.4372	4.1×10^{-5}	464.7
	15 3 12 1	- 14 3 11 1	263.4725	1.6×10^{-4}	173.0
	38 8 31 0	- 38 7 31 1	263.5386	6.2×10^{-5}	752.0
	30 16 14 2	- 31 15 17 2	263.6169	1.5×10^{-5}	711.0
	30 16 15 2	- 31 15 16 2	263.6169	1.5×10^{-5}	711.0
	13 3 11 0	- 12 2 11 1	263.6955	4.4×10^{-5}	143.6
	39 6 34 0	- 38 7 32 1	263.7078	2.2×10^{-5}	751.8
CH ₃ CHO	31 10 22 4	- 32 9 24 4	259.9788	1.4×10^{-5}	886.7
	44 6 39 0	- 43 7 36 0	260.0225	1.9×10^{-5}	998.9
	32 2 30 6	- 33 1 33 6	260.1158	1.8×10^{-6}	887.2
	6 1 5 8	- 5 1 5 7	260.1739	4.1×10^{-5}	406.7
	19 3 17 0	- 19 2 18 0	260.2834	5.7×10^{-5}	196.4
	13 1 13 1	- 12 0 12 2	260.4080	7.7×10^{-5}	83.9
	13 1 13 4	- 12 0 12 5	261.8892	7.6×10^{-5}	288.3
	14 5 10 6	- 13 5 9 6	261.9566	5.7×10^{-4}	545.7
	14 5 9 6	- 13 5 8 6	261.9567	5.7×10^{-4}	545.7
	17 0 17 8	- 16 1 16 7	261.9591	7.2×10^{-5}	516.5
	10 1 10 7	- 9 0 9 8	262.1191	5.3×10^{-5}	431.0
	9 2 7 6	- 9 1 8 6	262.7285	7.8×10^{-5}	427.8
	14 0 14 3	- 13 0 13 3	262.7300	6.3×10^{-4}	302.3
	14 1 14 6	- 13 1 13 6	262.8122	5.9×10^{-4}	466.8
	17 7 10 2	- 18 6 12 2	262.8291	1.2×10^{-5}	252.0
	14 9 5 8	- 13 9 4 8	262.8903	3.8×10^{-4}	664.5
	14 0 14 2	- 13 0 13 2	262.9601	6.2×10^{-4}	95.8
	7 5 3 3	- 8 4 4 3	262.9883	5.7×10^{-6}	285.4
	7 5 2 3	- 8 4 5 3	262.9893	5.7×10^{-6}	285.4
	14 0 14 0	- 13 0 13 0	263.0040	6.2×10^{-4}	95.7
	29 4 25 5	- 29 3 26 5	263.0099	7.5×10^{-5}	645.8
	10 1 10 6	- 9 0 9 6	263.1973	7.9×10^{-5}	421.7
	23 1 22 5	- 23 0 23 5	263.3144	3.5×10^{-5}	466.0
	6 2 5 1	- 5 1 5 1	263.4217	1.1×10^{-5}	28.5
	10 2 9 7	- 10 0 10 8	263.5180	2.4×10^{-6}	440.1
	15 2 13 5	- 15 1 15 4	263.7146	3.1×10^{-6}	327.0
	7 2 6 1	- 6 1 5 2	263.7366	3.4×10^{-5}	34.9
CH ₃ OCHO	35 8 27 3	- 34 9 26 3	260.0750	5.8×10^{-6}	605.4
	44 11 34 4	- 44 10 35 4	260.1058	2.5×10^{-5}	853.7
	46 8 38 3	- 46 7 39 3	260.1156	2.5×10^{-5}	881.6
	21 10 12 1	- 21 9 13 1	260.1286	2.1×10^{-5}	202.8
	21 10 11 2	- 21 9 12 2	260.1398	2.1×10^{-5}	202.8
	21 10 11 0	- 21 9 12 0	260.1626	2.1×10^{-5}	202.8
	21 10 12 0	- 21 9 13 0	260.1645	2.1×10^{-5}	202.8
	43 11 33 0	- 43 10 34 0	260.1757	2.8×10^{-5}	644.0

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	44 11 34 0	- 44 10 35 0	260.1818	2.8×10^{-5}	670.6
	43 11 33 1	- 43 10 34 1	260.2286	2.8×10^{-5}	644.0
	44 9 36 4	- 44 8 37 4	260.2328	2.6×10^{-5}	829.2
	44 11 34 1	- 44 10 35 1	260.2341	2.8×10^{-5}	670.5
	13 4 10 4	- 12 3 10 4	260.2388	3.8×10^{-6}	251.4
	43 11 33 4	- 43 10 34 4	260.2415	2.4×10^{-5}	827.3
	21 3 18 2	- 20 3 17 2	260.2445	2.6×10^{-4}	146.8
	21 3 18 0	- 20 3 17 0	260.2551	2.6×10^{-4}	146.8
	25 10 15 3	- 25 9 16 3	260.2673	2.3×10^{-5}	445.1
	12 3 9 2	- 11 2 10 1	260.2897	4.5×10^{-6}	53.2
	25 10 16 3	- 25 9 17 3	260.2967	2.3×10^{-5}	445.1
	12 3 9 0	- 11 2 10 0	260.3608	4.5×10^{-6}	53.2
	21 8 13 2	- 20 8 12 2	260.3843	2.3×10^{-4}	179.2
	21 8 14 0	- 20 8 13 0	260.3927	2.3×10^{-4}	179.2
	21 8 14 1	- 20 8 13 1	260.4040	2.3×10^{-4}	179.2
	21 8 13 0	- 20 8 12 0	260.4153	2.3×10^{-4}	179.2
	38 11 28 4	- 38 10 28 5	260.4369	1.6×10^{-6}	705.0
	38 11 27 2	- 38 10 28 2	261.7075	2.7×10^{-5}	520.6
	21 7 14 2	- 20 7 13 2	261.7155	2.4×10^{-4}	169.6
	43 9 35 0	- 42 10 32 0	261.7196	3.6×10^{-6}	619.6
	21 6 15 5	- 20 6 14 5	261.7271	2.4×10^{-4}	348.0
	43 9 35 1	- 42 10 32 2	261.7273	3.6×10^{-6}	619.6
	21 7 14 0	- 20 7 13 0	261.7466	2.4×10^{-4}	169.5
	44 9 36 4	- 43 10 33 5	261.7591	3.4×10^{-6}	829.2
	21 6 15 3	- 20 6 14 3	261.7612	2.5×10^{-4}	348.1
	17 10 8 1	- 17 9 9 1	261.7731	1.8×10^{-5}	156.6
	17 10 7 2	- 17 9 8 2	261.7850	1.8×10^{-5}	156.6
	17 10 7 0	- 17 9 8 0	261.8223	1.8×10^{-5}	156.6
	17 10 8 0	- 17 9 9 0	261.8223	1.8×10^{-5}	156.6
	39 9 31 4	- 39 7 32 5	261.8334	6.1×10^{-6}	703.9
	42 10 33 4	- 41 11 31 4	261.8559	3.2×10^{-6}	789.0
	42 9 34 3	- 41 10 31 3	261.8562	4.4×10^{-6}	777.0
	24 10 14 5	- 24 9 15 5	261.9112	2.3×10^{-5}	430.6
	46 10 37 3	- 46 9 38 3	261.9118	2.8×10^{-5}	895.3
	38 9 30 0	- 38 7 31 0	261.9944	6.0×10^{-6}	496.2
	16 10 7 1	- 16 9 8 1	262.0360	1.7×10^{-5}	146.6
	16 10 6 2	- 16 9 7 2	262.0478	1.7×10^{-5}	146.6
	41 11 31 0	- 41 10 32 0	262.0580	2.8×10^{-5}	592.7
	38 9 30 1	- 38 7 31 2	262.0825	6.0×10^{-6}	496.2
	16 10 6 0	- 16 9 7 0	262.0882	1.7×10^{-5}	146.6
	16 10 7 0	- 16 9 8 0	262.0882	1.7×10^{-5}	146.6
	37 5 32 5	- 37 5 33 4	262.1035	8.9×10^{-6}	625.0
	41 11 31 1	- 41 10 32 1	262.1078	2.8×10^{-5}	592.7
	21 7 14 2	- 20 7 14 1	262.1113	4.6×10^{-6}	169.6
	19 4 16 4	- 18 3 15 5	262.1355	1.7×10^{-5}	309.7
	41 11 31 4	- 41 10 32 4	262.7105	2.4×10^{-5}	776.5
	43 9 35 0	- 43 7 36 0	262.7268	9.2×10^{-6}	619.6
	46 11 36 0	- 46 10 37 0	262.7278	2.9×10^{-5}	725.6
	23 10 13 5	- 23 9 14 5	262.7323	2.3×10^{-5}	416.5
	12 10 2 0	- 12 9 3 0	262.7462	1.1×10^{-5}	112.4
	12 10 3 0	- 12 9 4 0	262.7462	1.1×10^{-5}	112.4
	43 9 35 1	- 43 7 36 2	262.7498	9.2×10^{-6}	619.6
	11 10 2 1	- 11 9 3 1	262.7698	8.3×10^{-6}	105.3
	46 11 36 1	- 46 10 37 1	262.7761	2.9×10^{-5}	725.5
	11 10 1 2	- 11 9 2 2	262.7821	8.3×10^{-6}	105.3
	12 3 9 3	- 11 1 10 3	262.8302	2.4×10^{-6}	240.5

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	10 10 1 1	- 10 9 2 1	262.8348	4.7×10^{-6}	98.8
	11 10 1 0	- 11 9 2 0	262.8348	8.3×10^{-6}	105.3
	11 10 2 0	- 11 9 3 0	262.8348	8.3×10^{-6}	105.3
	10 10 0 2	- 10 9 1 2	262.8472	4.7×10^{-6}	98.8
	32 7 25 3	- 31 8 24 3	262.8826	6.5×10^{-6}	535.9
	10 10 0 0	- 10 9 1 0	262.9019	4.7×10^{-6}	98.8
	10 10 1 0	- 10 9 2 0	262.9019	4.7×10^{-6}	98.8
	39 7 33 1	- 39 6 34 1	262.9162	2.2×10^{-5}	497.3
	39 7 33 0	- 39 6 34 0	262.9513	2.2×10^{-5}	497.3
	43 7 36 3	- 43 7 37 3	262.9776	9.4×10^{-6}	790.0
	51 14 38 1	- 50 15 35 2	262.9834	2.0×10^{-6}	918.0
	44 9 36 3	- 43 10 33 3	263.0190	3.6×10^{-6}	828.7
	42 10 33 1	- 41 11 30 2	263.0193	5.5×10^{-6}	605.5
	20 10 10 3	- 20 9 11 3	263.0300	2.1×10^{-5}	377.5
	20 10 11 3	- 20 9 12 3	263.0305	2.1×10^{-5}	377.5
	22 10 13 4	- 22 9 14 4	263.0364	2.2×10^{-5}	402.5
	29 1 28 3	- 29 1 29 3	263.0528	2.8×10^{-6}	427.4
	29 1 28 3	- 29 0 29 3	263.0529	5.2×10^{-6}	427.4
	29 2 28 3	- 29 1 29 3	263.0536	5.2×10^{-6}	427.4
	29 2 28 3	- 29 0 29 3	263.0536	2.8×10^{-6}	427.4
	42 10 33 0	- 41 11 30 0	263.0820	5.5×10^{-6}	605.5
	44 11 34 1	- 43 12 31 2	263.1054	5.5×10^{-6}	670.5
	44 11 34 0	- 43 12 31 0	263.1933	5.6×10^{-6}	670.6
	51 14 38 1	- 50 15 36 1	263.3156	3.5×10^{-6}	918.0
	43 9 35 4	- 42 10 32 5	263.3286	3.9×10^{-6}	803.0
	40 6 34 3	- 40 6 35 3	263.3459	9.2×10^{-6}	704.3
	19 10 10 3	- 19 9 11 3	263.3471	2.1×10^{-5}	365.7
	19 10 9 3	- 19 9 10 3	263.3471	2.1×10^{-5}	365.7
	49 9 40 2	- 49 8 41 2	263.3632	2.7×10^{-5}	798.0
	42 8 35 3	- 42 7 36 3	263.3690	2.5×10^{-5}	765.2
	39 7 33 1	- 39 5 34 2	263.3753	9.7×10^{-6}	497.3
	49 9 40 0	- 49 8 41 0	263.3938	2.7×10^{-5}	798.0
	39 7 33 0	- 39 5 34 0	263.4086	9.7×10^{-6}	497.3
	22 10 12 5	- 22 9 13 5	263.4524	2.2×10^{-5}	403.0
	9 6 4 4	- 8 5 4 4	263.4606	2.9×10^{-5}	237.4
	29 6 23 3	- 28 7 22 3	263.4713	7.7×10^{-6}	472.9
	47 12 35 5	- 47 11 36 5	263.5168	2.5×10^{-5}	950.7
	31 8 24 4	- 31 6 25 5	263.5309	3.7×10^{-6}	523.5
	51 10 42 0	- 50 11 39 0	263.5493	2.1×10^{-6}	859.8
	37 6 32 4	- 37 5 33 4	263.5905	2.0×10^{-5}	625.0
	51 10 42 1	- 50 11 39 2	263.5969	2.1×10^{-6}	859.8
	18 10 8 3	- 18 9 9 3	263.6061	2.0×10^{-5}	354.6
	18 10 9 3	- 18 9 10 3	263.6061	2.0×10^{-5}	354.6
	21 10 12 4	- 21 9 13 4	263.6491	2.2×10^{-5}	389.6
	40 11 30 0	- 40 10 31 0	263.6666	2.9×10^{-5}	568.0
	37 6 32 4	- 37 4 33 5	263.7116	9.1×10^{-6}	625.0
	40 11 30 1	- 40 10 31 1	263.7127	2.9×10^{-5}	568.0
CH ₃ OCH ₃	20 5 16 3	- 20 4 17 3	259.9825	9.2×10^{-5}	226.6
	20 5 16 5	- 20 4 17 5	259.9826	9.2×10^{-5}	226.6
	20 5 16 1	- 20 4 17 1	259.9844	9.2×10^{-5}	226.6
	20 5 16 0	- 20 4 17 0	259.9863	9.2×10^{-5}	226.6
	24 5 20 3	- 24 4 21 3	260.0034	9.6×10^{-5}	308.9
	24 5 20 5	- 24 4 21 5	260.0034	9.6×10^{-5}	308.9
	24 5 20 1	- 24 4 21 1	260.0044	9.6×10^{-5}	308.9
	24 5 20 0	- 24 4 21 0	260.0054	9.6×10^{-5}	308.9
	19 5 15 3	- 19 4 16 3	260.3271	9.1×10^{-5}	208.3

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	19 5 15 5	- 19 4 16 5	260.3272	9.1×10^{-5}	208.3
	19 5 15 1	- 19 4 16 1	260.3292	9.1×10^{-5}	208.3
	19 5 15 0	- 19 4 16 0	260.3313	9.1×10^{-5}	208.3
	16 5 11 5	- 16 4 12 5	260.4005	8.7×10^{-5}	159.0
	16 5 11 3	- 16 4 12 3	260.4011	8.6×10^{-5}	159.0
	16 5 11 1	- 16 4 12 1	260.4032	8.7×10^{-5}	159.0
	16 5 11 0	- 16 4 12 0	260.4056	8.7×10^{-5}	159.0
	28 4 25 0	- 27 5 22 0	261.8399	2.3×10^{-5}	393.2
	28 4 25 1	- 27 5 22 1	261.8402	2.3×10^{-5}	393.2
	28 4 25 3	- 27 5 22 3	261.8404	2.3×10^{-5}	393.2
	28 4 25 5	- 27 5 22 5	261.8404	2.3×10^{-5}	393.2
	14 5 10 3	- 14 4 10 3	261.8785	1.1×10^{-5}	130.8
	14 5 10 1	- 14 4 10 1	261.8832	3.9×10^{-6}	130.8
	14 5 9 5	- 14 4 10 5	261.8941	8.6×10^{-5}	130.8
	14 5 9 3	- 14 4 10 3	261.8965	7.5×10^{-5}	130.8
	14 5 9 1	- 14 4 10 1	261.8976	8.2×10^{-5}	130.8
	14 5 9 0	- 14 4 10 0	261.8998	8.6×10^{-5}	130.8
	15 5 11 3	- 15 4 12 3	261.9530	8.4×10^{-5}	144.4
	15 5 11 5	- 15 4 12 5	261.9543	8.7×10^{-5}	144.4
	15 5 11 1	- 15 4 12 1	261.9566	8.6×10^{-5}	144.4
	15 5 11 0	- 15 4 12 0	261.9596	8.7×10^{-5}	144.4
	15 5 10 3	- 15 4 12 3	261.9814	3.7×10^{-6}	144.4
	15 5 10 1	- 15 4 12 1	261.9830	1.0×10^{-6}	144.4
	12 5 7 3	- 12 4 8 3	262.7548	2.8×10^{-5}	106.2
	12 5 8 1	- 12 4 8 1	262.7616	2.3×10^{-5}	106.2
	12 5 7 5	- 12 4 8 5	262.7637	8.2×10^{-5}	106.2
	12 5 8 3	- 12 4 8 3	262.7688	5.4×10^{-5}	106.2
	12 5 7 1	- 12 4 8 1	262.7690	5.9×10^{-5}	106.2
	12 5 7 0	- 12 4 8 0	262.7699	8.2×10^{-5}	106.2
	8 2 6 5	- 7 1 7 5	262.7712	3.3×10^{-5}	38.6
	8 2 6 3	- 7 1 7 3	262.7712	3.3×10^{-5}	38.6
	8 2 6 1	- 7 1 7 1	262.7740	3.3×10^{-5}	38.6
	8 2 6 0	- 7 1 7 0	262.7769	3.3×10^{-5}	38.6
	12 5 7 3	- 12 4 9 3	262.8843	5.5×10^{-5}	106.2
	12 5 8 5	- 12 4 9 5	262.8893	8.2×10^{-5}	106.2
	12 5 8 1	- 12 4 9 1	262.8902	5.9×10^{-5}	106.2
	12 5 8 0	- 12 4 9 0	262.8954	8.2×10^{-5}	106.2
	12 5 7 1	- 12 4 9 1	262.8976	2.3×10^{-5}	106.2
	12 5 8 3	- 12 4 9 3	262.8982	2.8×10^{-5}	106.2
	27 5 23 3	- 27 4 24 3	262.9642	1.0×10^{-4}	380.2
	27 5 23 5	- 27 4 24 5	262.9642	1.0×10^{-4}	380.2
	27 5 23 1	- 27 4 24 1	262.9645	1.0×10^{-4}	380.2
	27 5 23 0	- 27 4 24 0	262.9649	1.0×10^{-4}	380.2
	11 5 6 3	- 11 4 7 3	263.0353	2.6×10^{-5}	95.2
	11 5 7 1	- 11 4 7 1	263.0428	2.8×10^{-5}	95.2
	11 5 6 5	- 11 4 7 5	263.0441	8.0×10^{-5}	95.2
	11 5 7 3	- 11 4 7 3	263.0494	5.3×10^{-5}	95.2
	11 5 6 1	- 11 4 7 1	263.0500	5.1×10^{-5}	95.2
	11 5 6 0	- 11 4 7 0	263.0504	8.0×10^{-5}	95.2
	11 5 6 3	- 11 4 8 3	263.1021	5.3×10^{-5}	95.2
	11 5 7 5	- 11 4 8 5	263.1075	8.0×10^{-5}	95.2
	11 5 7 1	- 11 4 8 1	263.1079	5.1×10^{-5}	95.2
	11 5 7 0	- 11 4 8 0	263.1138	8.0×10^{-5}	95.2
	11 5 6 1	- 11 4 8 1	263.1151	2.8×10^{-5}	95.2
	11 5 7 3	- 11 4 8 3	263.1163	2.6×10^{-5}	95.2
	9 5 4 3	- 9 4 6 3	263.3887	6.4×10^{-6}	76.1

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	9 5 5 1	- 9 4 5 1	263.3991	1.5×10^{-5}	76.1
	9 5 4 5	- 9 4 5 5	263.4011	7.3×10^{-5}	76.1
	9 5 5 3	- 9 4 6 3	263.4037	6.6×10^{-5}	76.1
	9 5 4 1	- 9 4 5 1	263.4066	5.8×10^{-5}	76.1
	9 5 4 0	- 9 4 5 0	263.4078	7.3×10^{-5}	76.1
	9 5 4 3	- 9 4 5 3	263.4114	6.6×10^{-5}	76.1
	9 5 5 5	- 9 4 6 5	263.4140	7.3×10^{-5}	76.1
	9 5 5 1	- 9 4 6 1	263.4151	5.8×10^{-5}	76.1
	9 5 5 0	- 9 4 6 0	263.4206	7.3×10^{-5}	76.1
	9 5 4 1	- 9 4 6 1	263.4226	1.5×10^{-5}	76.1
	9 5 5 3	- 9 4 5 3	263.4263	6.4×10^{-6}	76.1
	8 5 3 3	- 8 4 5 3	263.4921	1.1×10^{-6}	67.9
	8 5 4 1	- 8 4 4 1	263.5038	3.8×10^{-6}	67.9
	8 5 3 5	- 8 4 4 5	263.5071	6.7×10^{-5}	67.9
	8 5 4 3	- 8 4 5 3	263.5074	6.6×10^{-5}	67.9
	8 5 3 1	- 8 4 4 1	263.5115	6.3×10^{-5}	67.9
	8 5 3 3	- 8 4 4 3	263.5118	6.6×10^{-5}	67.9
	8 5 4 5	- 8 4 5 5	263.5121	6.7×10^{-5}	67.9
	8 5 3 0	- 8 4 4 0	263.5140	6.7×10^{-5}	67.9
	8 5 4 1	- 8 4 5 1	263.5146	6.3×10^{-5}	67.9
	8 5 4 0	- 8 4 5 0	263.5189	6.7×10^{-5}	67.9
	8 5 3 1	- 8 4 5 1	263.5222	3.8×10^{-6}	67.9
	8 5 4 3	- 8 4 4 3	263.5271	1.1×10^{-6}	67.9
	7 5 3 3	- 7 4 4 3	263.5784	6.0×10^{-5}	60.6
	7 5 2 5	- 7 4 3 5	263.5795	6.0×10^{-5}	60.6
	7 5 3 5	- 7 4 4 5	263.5812	6.0×10^{-5}	60.6
	7 5 2 3	- 7 4 3 3	263.5822	6.0×10^{-5}	60.6
	7 5 2 1	- 7 4 3 1	263.5828	5.9×10^{-5}	60.6
	7 5 3 1	- 7 4 4 1	263.5848	5.9×10^{-5}	60.6
	7 5 2 0	- 7 4 3 0	263.5865	6.0×10^{-5}	60.6
	7 5 3 0	- 7 4 4 0	263.5881	6.0×10^{-5}	60.6
	6 5 2 3	- 6 4 3 3	263.6249	4.9×10^{-5}	54.3
	6 5 1 5	- 6 4 2 5	263.6266	4.9×10^{-5}	54.3
	6 5 2 5	- 6 4 3 5	263.6271	4.9×10^{-5}	54.3
	6 5 1 3	- 6 4 2 3	263.6288	4.9×10^{-5}	54.3
	6 5 1 1	- 6 4 2 1	263.6294	4.8×10^{-5}	54.3
	6 5 2 1	- 6 4 3 1	263.6313	4.8×10^{-5}	54.3
	6 5 1 0	- 6 4 2 0	263.6337	4.9×10^{-5}	54.3
	6 5 2 0	- 6 4 3 0	263.6342	4.9×10^{-5}	54.3
	5 5 1 3	- 5 4 2 3	263.6534	3.1×10^{-5}	48.8
	5 5 0 5	- 5 4 1 5	263.6553	3.1×10^{-5}	48.8
	5 5 1 5	- 5 4 2 5	263.6553	3.1×10^{-5}	48.8
	5 5 0 3	- 5 4 1 3	263.6573	3.1×10^{-5}	48.8
	5 5 0 1	- 5 4 1 1	263.6579	3.1×10^{-5}	48.8
	5 5 1 1	- 5 4 2 1	263.6599	3.1×10^{-5}	48.8
	5 5 0 0	- 5 4 1 0	263.6625	3.1×10^{-5}	48.8
	5 5 1 0	- 5 4 2 0	263.6625	3.1×10^{-5}	48.8
CH ₃ COCH ₃	38 26 12 0	- 38 23 15 0	260.0197	2.3×10^{-5}	615.2
	10 9 2 1	- 9 6 3 1	260.0832	4.7×10^{-6}	49.8
	25 19 7 0	- 25 16 10 1	260.0881	2.4×10^{-6}	277.8
	29 27 2 0	- 28 28 1 0	260.1037	1.0×10^{-6}	404.6
	29 27 3 0	- 28 28 0 0	260.1037	1.0×10^{-6}	404.6
	23 16 8 1	- 22 17 5 1	260.1095	3.5×10^{-5}	229.8
	28 25 4 0	- 27 26 2 1	260.1376	3.3×10^{-6}	370.0
	27 23 5 0	- 26 24 2 0	260.1771	7.1×10^{-6}	337.4
	27 23 4 0	- 26 24 3 0	260.1771	7.1×10^{-6}	337.4

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	25 19 7 1	- 25 16 10 1	260.1828	4.4×10^{-6}	277.8
	28 17 11 0	- 28 14 14 0	260.1892	7.0×10^{-6}	330.6
	34 24 11 1	- 34 21 13 2	260.1953	1.7×10^{-6}	497.8
	9 9 0 0	- 8 6 3 0	260.2293	1.2×10^{-6}	42.7
	44 14 30 0	- 44 13 31 0	260.2497	3.0×10^{-4}	690.3
	44 15 30 0	- 44 14 31 0	260.2497	3.0×10^{-4}	690.3
	44 14 30 0	- 44 13 31 1	260.2689	3.4×10^{-5}	690.3
	44 14 30 0	- 44 14 31 1	260.2689	3.4×10^{-5}	690.3
	44 15 30 0	- 44 13 31 1	260.2689	3.9×10^{-4}	690.3
	44 15 30 0	- 44 14 31 1	260.2689	3.9×10^{-4}	690.3
	29 27 3 0	- 28 28 0 1	260.2727	1.0×10^{-6}	404.5
	28 17 11 0	- 28 14 14 1	260.2796	7.0×10^{-6}	330.6
	44 14 30 1	- 44 14 31 2	260.2881	3.0×10^{-4}	690.2
	44 15 30 1	- 44 13 31 2	260.2881	3.0×10^{-4}	690.2
	44 14 30 1	- 44 14 31 1	260.2881	3.0×10^{-4}	690.2
	44 15 30 1	- 44 13 31 1	260.2881	3.0×10^{-4}	690.2
	27 19 9 1	- 27 16 12 2	260.3305	6.5×10^{-6}	315.8
	28 25 3 0	- 27 26 1 1	260.3326	3.3×10^{-6}	370.0
	36 25 11 0	- 36 22 14 1	260.3392	2.0×10^{-5}	555.0
	28 17 11 1	- 28 14 14 1	260.3691	6.9×10^{-6}	330.5
	28 17 11 1	- 28 14 14 2	260.3772	6.9×10^{-6}	330.5
	28 25 4 1	- 27 26 2 2	260.4195	3.3×10^{-6}	369.9
	29 27 3 1	- 28 28 0 2	260.4414	1.0×10^{-6}	404.4
	38 26 12 1	- 38 23 15 2	261.7077	2.0×10^{-5}	615.2
	18 9 9 0	- 18 6 12 0	261.7099	1.0×10^{-6}	134.3
	26 21 6 0	- 25 22 4 1	261.7369	1.3×10^{-5}	306.6
	30 22 8 1	- 30 19 11 2	261.7414	4.8×10^{-6}	393.3
	41 11 30 1	- 41 11 31 2	261.7560	2.6×10^{-4}	576.3
	41 12 30 1	- 41 10 31 2	261.7560	2.6×10^{-4}	576.3
	41 11 30 1	- 41 10 31 1	261.7561	2.6×10^{-4}	576.3
	41 12 30 1	- 41 11 31 1	261.7561	2.6×10^{-4}	576.3
	30 22 8 0	- 30 19 11 1	261.7703	5.6×10^{-6}	393.3
	41 11 30 0	- 41 10 31 1	261.7955	1.5×10^{-4}	576.4
	41 11 30 0	- 41 11 31 1	261.7955	1.5×10^{-4}	576.4
	41 12 30 0	- 41 10 31 1	261.7955	2.0×10^{-4}	576.4
	41 12 30 0	- 41 11 31 1	261.7955	2.0×10^{-4}	576.4
	10 9 1 0	- 9 6 3 1	261.7975	2.4×10^{-6}	49.8
	20 7 13 1	- 19 8 12 2	261.8181	4.9×10^{-4}	151.2
	20 7 13 1	- 19 8 12 1	261.8182	4.9×10^{-4}	151.2
	20 8 13 1	- 19 7 12 2	261.8191	4.9×10^{-4}	151.2
	20 8 13 1	- 19 7 12 1	261.8192	4.9×10^{-4}	151.2
	41 11 30 0	- 41 10 31 0	261.8347	2.6×10^{-4}	576.4
	41 12 30 0	- 41 11 31 0	261.8347	2.6×10^{-4}	576.4
	30 22 8 0	- 30 19 11 0	261.8621	7.7×10^{-6}	393.3
	20 7 13 0	- 19 8 12 1	261.9017	4.9×10^{-4}	151.1
	20 8 13 0	- 19 7 12 1	261.9027	4.9×10^{-4}	151.1
	40 27 13 0	- 40 24 16 0	261.9051	2.9×10^{-5}	678.6
	21 18 3 1	- 21 15 6 1	261.9131	1.1×10^{-6}	206.3
	20 7 13 0	- 19 8 12 0	261.9852	4.9×10^{-4}	151.1
	20 8 13 0	- 19 7 12 0	261.9862	4.9×10^{-4}	151.1
	26 21 5 1	- 25 22 3 2	261.9892	1.3×10^{-5}	306.6
	26 21 6 1	- 25 22 3 1	262.0656	1.3×10^{-5}	306.5
	26 21 5 1	- 25 22 4 1	262.0681	1.3×10^{-5}	306.5
	26 21 6 1	- 25 22 4 2	262.1439	1.3×10^{-5}	306.5
	10 9 1 1	- 9 6 3 2	262.1591	2.6×10^{-6}	49.9
	40 10 30 1	- 40 10 31 2	262.1740	2.4×10^{-4}	540.1

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	40 11 30 1	- 40 9 31 2	262.1740	2.4×10^{-4}	540.1
	40 10 30 1	- 40 10 31 1	262.1741	2.4×10^{-4}	540.1
	40 11 30 1	- 40 9 31 1	262.1741	2.4×10^{-4}	540.1
	50 23 28 1	- 49 24 25 1	262.7522	2.6×10^{-6}	964.3
	39 10 30 0	- 39 9 31 0	262.7529	2.3×10^{-4}	504.8
	39 9 30 0	- 39 8 31 0	262.7529	2.3×10^{-4}	504.8
	50 22 28 0	- 49 25 25 1	262.7640	4.1×10^{-6}	964.3
	50 23 28 0	- 49 24 25 1	262.7644	4.1×10^{-6}	964.3
	21 6 15 1	- 20 7 14 2	262.7689	5.7×10^{-4}	158.3
	21 7 15 1	- 20 6 14 2	262.7689	5.7×10^{-4}	158.3
	21 6 15 1	- 20 7 14 1	262.7689	5.7×10^{-4}	158.3
	21 7 15 1	- 20 6 14 1	262.7689	5.7×10^{-4}	158.3
	50 22 28 0	- 49 25 25 0	262.7764	2.5×10^{-6}	964.4
	50 23 28 0	- 49 24 25 0	262.7767	2.6×10^{-6}	964.4
	40 27 13 0	- 40 24 16 1	262.8043	3.7×10^{-5}	678.6
	21 6 15 0	- 20 7 14 1	262.8419	5.7×10^{-4}	158.2
	21 7 15 0	- 20 6 14 1	262.8419	5.7×10^{-4}	158.2
	30 17 13 1	- 29 20 10 1	262.8726	6.8×10^{-6}	374.2
	30 17 13 0	- 29 20 10 1	262.9092	6.8×10^{-6}	374.3
	38 8 30 1	- 38 8 31 2	262.9134	2.1×10^{-4}	470.2
	38 9 30 1	- 38 7 31 2	262.9134	2.1×10^{-4}	470.2
	38 8 30 1	- 38 8 31 1	262.9135	2.1×10^{-4}	470.2
	38 9 30 1	- 38 7 31 1	262.9135	2.1×10^{-4}	470.2
	21 6 15 0	- 20 7 14 0	262.9147	5.7×10^{-4}	158.2
	21 7 15 0	- 20 6 14 0	262.9147	5.7×10^{-4}	158.2
	30 17 13 0	- 29 20 10 0	262.9170	6.8×10^{-6}	374.3
	30 17 13 1	- 29 20 10 2	262.9385	6.7×10^{-6}	374.2
	23 16 7 1	- 22 17 5 2	262.9661	2.7×10^{-5}	230.0
	14 11 4 0	- 13 10 3 0	262.9821	4.6×10^{-4}	90.4
	21 18 3 1	- 21 15 6 2	262.9920	1.0×10^{-6}	206.4
	38 8 30 0	- 38 7 31 1	263.0452	2.3×10^{-5}	470.3
	38 8 30 0	- 38 8 31 1	263.0452	3.9×10^{-5}	470.3
	38 9 30 0	- 38 7 31 1	263.0452	2.2×10^{-4}	470.3
	38 9 30 0	- 38 8 31 1	263.0452	2.4×10^{-4}	470.3
	21 18 4 1	- 21 15 7 2	263.1102	1.0×10^{-6}	206.2
	38 8 30 0	- 38 7 31 0	263.1764	2.1×10^{-4}	470.3
	38 9 30 0	- 38 8 31 0	263.1764	2.1×10^{-4}	470.3
	14 11 4 1	- 13 10 4 2	263.3286	4.4×10^{-4}	90.4
	14 11 3 1	- 13 10 3 2	263.3983	4.4×10^{-4}	90.6
	37 7 30 0	- 37 6 31 1	263.4085	5.1×10^{-6}	436.6
	37 7 30 0	- 37 7 31 1	263.4085	5.1×10^{-6}	436.6
	37 8 30 0	- 37 6 31 1	263.4085	2.2×10^{-4}	436.6
	37 8 30 0	- 37 7 31 1	263.4085	2.2×10^{-4}	436.6
	18 10 9 0	- 17 9 8 1	263.4583	3.2×10^{-4}	134.4
	21 18 3 0	- 21 15 6 0	263.5077	1.1×10^{-6}	206.4
	14 11 3 0	- 13 10 3 1	263.5148	4.0×10^{-4}	90.5
	40 27 13 1	- 40 24 16 1	263.5282	2.9×10^{-5}	678.6
	36 6 30 1	- 36 5 31 2	263.5343	1.7×10^{-4}	403.8
	36 7 30 1	- 36 6 31 2	263.5343	1.7×10^{-4}	403.8
	36 6 30 1	- 36 6 31 1	263.5344	1.7×10^{-4}	403.8
	36 7 30 1	- 36 5 31 1	263.5344	1.7×10^{-4}	403.8
	37 7 30 0	- 37 7 31 0	263.5783	1.9×10^{-4}	436.6
	37 8 30 0	- 37 6 31 0	263.5783	1.9×10^{-4}	436.6
	18 10 9 0	- 17 9 8 0	263.6265	3.2×10^{-4}	134.4
	23 16 7 0	- 22 17 6 0	263.6422	3.6×10^{-5}	230.0
	14 11 4 0	- 13 10 4 1	263.6745	3.9×10^{-4}	90.4
	14 10 4 1	- 13 9 5 1	263.6827	3.0×10^{-4}	88.0

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	10 8 3 1	- 9 5 5 2	263.7026	2.1×10^{-6}	47.5
	25 19 7 0	- 24 20 4 0	263.7264	2.2×10^{-5}	277.8
	36 6 30 0	- 36 5 31 1	263.7477	4.5×10^{-5}	403.8
	36 6 30 0	- 36 6 31 1	263.7477	4.5×10^{-5}	403.8
	36 7 30 0	- 36 5 31 1	263.7477	1.5×10^{-4}	403.8
	36 7 30 0	- 36 6 31 1	263.7477	1.5×10^{-4}	403.8
aGg'(CH ₂ OH) ₂	46 11 35 1	- 45 12 34 1	259.9713	1.3×10^{-5}	593.8
	47 3 44 0	- 47 3 45 1	259.9736	8.1×10^{-6}	546.3
	47 4 44 0	- 47 2 45 1	259.9737	8.1×10^{-6}	546.3
	55 14 41 1	- 55 13 43 0	260.0722	1.2×10^{-5}	855.4
	26 12 14 0	- 25 12 13 1	260.1167	3.4×10^{-4}	243.3
	26 12 15 0	- 25 12 14 1	260.1167	3.4×10^{-4}	243.3
	17 5 13 0	- 16 4 13 1	260.1561	2.1×10^{-5}	87.8
	30 13 18 1	- 30 12 18 0	260.2226	8.7×10^{-6}	311.9
	30 13 17 1	- 30 12 19 0	260.2226	8.7×10^{-6}	311.9
	47 14 34 0	- 47 13 34 1	260.2259	9.0×10^{-6}	651.5
	47 14 33 0	- 47 13 35 1	260.2706	8.8×10^{-6}	651.5
	29 5 24 1	- 28 6 23 1	260.2757	3.6×10^{-5}	231.5
	26 11 16 0	- 25 11 15 1	260.3799	3.5×10^{-4}	232.1
	26 11 15 0	- 25 11 14 1	260.3799	3.5×10^{-4}	232.1
	46 6 41 1	- 45 7 39 0	260.4379	2.3×10^{-6}	547.7
	23 5 18 0	- 22 5 18 0	261.7385	2.2×10^{-6}	149.9
	24 13 11 1	- 24 12 13 0	261.7715	8.0×10^{-6}	230.7
	24 13 12 1	- 24 12 12 0	261.7715	8.0×10^{-6}	230.7
	24 5 19 1	- 23 5 18 0	261.8151	4.2×10^{-4}	162.5
	53 14 40 0	- 52 15 37 0	261.8547	1.0×10^{-5}	801.3
	53 14 40 1	- 52 15 37 1	261.8655	1.4×10^{-5}	801.5
	59 9 51 1	- 59 8 52 1	261.8778	4.0×10^{-5}	914.0
	18 4 15 0	- 17 3 15 1	261.8919	9.2×10^{-6}	92.3
	26 8 19 0	- 25 8 18 1	261.9036	4.0×10^{-4}	204.5
	53 14 39 0	- 52 15 38 0	261.9265	1.0×10^{-5}	801.3
	59 9 51 0	- 59 8 52 0	261.9386	4.6×10^{-5}	913.6
	53 14 39 1	- 52 15 38 1	261.9416	1.4×10^{-5}	801.5
	23 13 10 1	- 23 12 12 0	261.9447	7.8×10^{-6}	218.9
	23 13 11 1	- 23 12 11 0	261.9447	7.8×10^{-6}	218.9
	60 8 52 0	- 60 8 53 1	261.9497	1.7×10^{-5}	942.7
	26 8 18 0	- 25 8 17 1	261.9712	4.0×10^{-4}	204.5
	34 9 26 0	- 34 7 27 1	262.0007	2.2×10^{-6}	333.6
	50 14 36 1	- 50 13 37 1	262.0294	5.5×10^{-5}	724.4
	23 5 18 1	- 22 5 18 1	262.0416	3.6×10^{-6}	150.3
	25 3 22 1	- 24 3 21 0	262.0459	4.7×10^{-4}	167.7
	45 14 32 0	- 45 13 32 1	262.0479	9.2×10^{-6}	605.6
	45 14 31 0	- 45 13 33 1	262.0631	9.1×10^{-6}	605.6
	50 14 36 0	- 50 13 37 0	262.0948	5.4×10^{-5}	724.1
	22 13 10 1	- 22 12 10 0	262.0986	7.5×10^{-6}	207.6
	22 13 9 1	- 22 12 11 0	262.0986	7.5×10^{-6}	207.6
	14 6 8 1	- 13 5 8 0	262.1074	1.2×10^{-5}	69.7
	34 5 29 1	- 33 6 27 0	262.1490	2.4×10^{-6}	311.8
	58 12 47 1	- 58 10 48 0	262.1613	8.3×10^{-6}	917.2
	12 7 5 1	- 11 6 5 0	262.7122	1.4×10^{-5}	62.7
	12 7 6 1	- 11 6 6 0	262.7128	1.4×10^{-5}	62.7
	26 7 20 0	- 25 7 19 1	262.7212	4.0×10^{-4}	197.4
	18 5 14 0	- 17 4 13 0	262.7275	2.6×10^{-5}	96.7
	25 18 7 1	- 24 18 6 0	262.7483	2.1×10^{-4}	319.0
	25 18 8 1	- 24 18 7 0	262.7483	2.1×10^{-4}	319.0
	15 13 2 1	- 15 12 4 0	262.7550	3.9×10^{-6}	142.3

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	15 13 3 1	- 15 12 3 0	262.7550	3.9×10^{-6}	142.3
	25 13 12 1	- 24 13 11 0	262.7564	3.2×10^{-4}	243.0
	25 13 13 1	- 24 13 12 0	262.7564	3.2×10^{-4}	243.0
	51 5 46 1	- 51 4 47 1	262.7644	3.1×10^{-5}	664.3
	51 6 46 1	- 51 5 47 1	262.7705	3.1×10^{-5}	664.3
	14 13 1 1	- 14 12 3 0	262.8042	2.9×10^{-6}	134.9
	14 13 2 1	- 14 12 2 0	262.8042	2.9×10^{-6}	134.9
	25 19 6 1	- 24 19 5 0	262.8144	1.9×10^{-4}	337.2
	25 19 7 1	- 24 19 6 0	262.8144	1.9×10^{-4}	337.2
	26 3 24 1	- 25 3 23 0	262.8239	4.3×10^{-4}	174.2
	44 14 31 0	- 44 13 31 1	262.8422	9.2×10^{-6}	583.4
	13 13 1 1	- 13 12 1 0	262.8455	1.6×10^{-6}	128.1
	44 14 30 0	- 44 13 32 1	262.8508	9.2×10^{-6}	583.4
	18 5 14 1	- 17 4 13 1	262.8516	1.6×10^{-5}	97.0
	25 12 13 1	- 24 12 12 0	262.8661	3.4×10^{-4}	230.8
	25 12 14 1	- 24 12 13 0	262.8661	3.4×10^{-4}	230.8
	10 8 2 1	- 9 7 2 0	262.8770	1.7×10^{-5}	58.8
	10 8 3 1	- 9 7 3 0	262.8770	1.7×10^{-5}	58.8
	51 5 46 0	- 51 4 47 0	262.8941	3.2×10^{-5}	664.0
	25 20 5 1	- 24 20 4 0	262.8963	1.6×10^{-4}	356.3
	25 20 6 1	- 24 20 5 0	262.8963	1.6×10^{-4}	356.3
	51 6 46 0	- 51 5 47 0	262.9008	3.2×10^{-5}	664.0
	26 2 24 1	- 25 2 23 0	262.9441	4.4×10^{-4}	174.2
	24 8 17 0	- 24 6 18 1	262.9878	1.3×10^{-6}	179.2
	25 21 4 1	- 24 21 3 0	262.9924	1.3×10^{-4}	376.4
	25 21 5 1	- 24 21 4 0	262.9924	1.3×10^{-4}	376.4
	25 11 15 1	- 24 11 14 0	263.0395	3.6×10^{-4}	219.6
	25 11 14 1	- 24 11 13 0	263.0395	3.6×10^{-4}	219.6
	50 5 45 1	- 50 5 46 0	263.0857	1.2×10^{-5}	640.1
	50 6 45 1	- 50 4 46 0	263.0967	1.2×10^{-5}	640.1
	25 22 3 1	- 24 22 2 0	263.1018	1.0×10^{-4}	397.4
	25 22 4 1	- 24 22 3 0	263.1018	1.0×10^{-4}	397.4
	25 10 16 1	- 24 10 15 0	263.3025	3.7×10^{-4}	209.3
	25 10 15 1	- 24 10 14 0	263.3026	3.7×10^{-4}	209.3
	49 14 35 1	- 49 13 36 1	263.3298	5.6×10^{-5}	699.7
	25 24 1 1	- 24 24 0 0	263.3565	3.5×10^{-5}	442.4
	25 24 2 1	- 24 24 1 0	263.3565	3.5×10^{-5}	442.4
	49 14 35 0	- 49 13 36 0	263.3831	5.4×10^{-5}	699.5
	29 1 29 0	- 28 1 28 1	263.3921	4.5×10^{-4}	196.9
	29 0 29 0	- 28 0 28 1	263.3921	4.5×10^{-4}	196.9
	27 7 20 0	- 27 4 23 0	263.3926	1.4×10^{-6}	211.0
	26 7 19 0	- 25 7 18 1	263.4155	4.1×10^{-4}	197.5
	49 14 36 1	- 49 13 37 1	263.4359	5.6×10^{-5}	699.7
	26 6 21 0	- 25 6 20 1	263.4496	3.8×10^{-4}	191.4
	27 6 21 0	- 27 3 24 0	263.4667	1.7×10^{-6}	205.7
	25 5 21 1	- 24 5 20 0	263.4812	4.4×10^{-4}	173.3
	49 14 36 0	- 49 13 37 0	263.4838	5.3×10^{-5}	699.4
	42 9 34 1	- 41 10 32 0	263.4896	5.6×10^{-6}	487.2
	43 14 30 0	- 43 13 30 1	263.5672	9.3×10^{-6}	561.7
	43 14 29 0	- 43 13 31 1	263.5721	9.3×10^{-6}	561.7
	44 10 35 1	- 43 11 33 0	263.6747	9.7×10^{-6}	539.0
	47 11 37 0	- 46 12 35 1	263.6974	6.4×10^{-6}	616.8
	25 9 17 1	- 24 9 16 0	263.6982	3.9×10^{-4}	200.1
	25 9 16 1	- 24 9 15 0	263.7007	3.9×10^{-4}	200.1
	36 5 31 1	- 35 6 29 0	263.7327	1.8×10^{-6}	347.0
gGg'(CH₂OH)₂	60 8 52 0	- 60 8 53 1	259.9804	2.7×10^{-5}	936.2

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	50 14 37 1	- 50 13 38 1	260.0226	1.2×10^{-4}	718.7
	50 14 37 0	- 50 13 38 0	260.0388	9.8×10^{-5}	718.6
	47 6 42 0	- 46 6 40 0	260.0511	5.8×10^{-6}	566.1
	57 4 53 0	- 56 5 51 1	260.0513	1.7×10^{-5}	799.2
	57 5 53 0	- 56 6 51 1	260.0516	1.7×10^{-5}	799.2
	30 7 24 0	- 30 4 26 1	260.0601	8.1×10^{-6}	252.2
	28 0 28 1	- 27 1 27 1	260.1916	1.7×10^{-4}	183.2
	28 1 28 1	- 27 0 27 1	260.1917	1.7×10^{-4}	183.2
	28 0 28 0	- 27 1 27 0	260.1931	1.7×10^{-4}	183.1
	28 1 28 0	- 27 0 27 0	260.1932	1.7×10^{-4}	183.1
	48 6 43 0	- 47 7 41 1	260.2289	2.0×10^{-5}	588.9
	54 6 48 1	- 54 6 49 0	260.2348	5.6×10^{-6}	747.1
	54 7 48 1	- 54 5 49 0	260.2368	8.1×10^{-6}	747.1
	53 14 40 0	- 52 15 37 0	260.2665	2.2×10^{-5}	795.2
	53 14 40 1	- 52 15 37 1	260.2737	2.5×10^{-5}	795.2
	49 5 45 1	- 49 3 46 0	260.2761	5.1×10^{-6}	600.5
	49 4 45 1	- 49 4 46 0	260.2762	5.0×10^{-6}	600.5
	53 14 39 0	- 52 15 38 0	260.3312	2.5×10^{-5}	795.2
	53 14 39 1	- 52 15 38 1	260.3391	2.3×10^{-5}	795.2
	18 2 16 1	- 17 2 16 1	260.4196	4.2×10^{-6}	87.3
	35 7 28 1	- 34 8 27 1	261.7538	2.8×10^{-5}	335.9
	48 5 43 0	- 47 7 41 0	261.7594	3.6×10^{-6}	588.9
	15 6 9 0	- 14 5 9 1	261.7612	1.0×10^{-4}	76.2
	49 14 35 1	- 49 13 37 0	261.7776	8.8×10^{-5}	694.2
	45 3 43 1	- 45 2 44 1	261.7792	2.8×10^{-5}	488.4
	45 2 43 1	- 45 1 44 1	261.7792	2.8×10^{-5}	488.4
	51 13 39 0	- 50 14 37 1	261.7989	3.4×10^{-5}	731.2
	45 3 43 0	- 45 2 44 0	261.8089	2.8×10^{-5}	488.3
	45 2 43 0	- 45 1 44 0	261.8089	2.8×10^{-5}	488.3
	30 7 24 1	- 30 4 26 0	261.8350	8.3×10^{-6}	252.3
	57 8 50 1	- 57 6 51 0	261.9016	1.4×10^{-5}	839.2
	26 7 19 1	- 26 4 22 1	261.9779	4.5×10^{-6}	196.1
	41 8 34 1	- 40 9 32 0	261.9959	3.7×10^{-5}	454.8
	15 6 10 0	- 14 5 10 1	262.0159	1.1×10^{-4}	76.1
	43 2 42 1	- 43 1 43 1	262.0203	1.5×10^{-5}	435.8
	43 1 42 1	- 43 0 43 1	262.0203	1.5×10^{-5}	435.8
	48 6 43 0	- 47 6 41 0	262.0213	4.4×10^{-6}	588.9
	51 13 38 0	- 50 14 36 1	262.0401	2.2×10^{-5}	731.2
	43 1 42 0	- 43 0 43 0	262.0499	1.5×10^{-5}	435.7
	43 2 42 0	- 43 1 43 0	262.0499	1.5×10^{-5}	435.7
	26 7 19 0	- 26 4 22 0	262.0555	3.2×10^{-6}	196.0
	47 4 44 1	- 47 2 45 0	262.0701	4.1×10^{-6}	543.3
	47 3 44 1	- 47 3 45 0	262.0702	4.1×10^{-6}	543.3
	12 4 9 0	- 11 1 10 0	262.1539	2.6×10^{-6}	45.9
	52 6 47 1	- 52 5 48 1	262.7561	6.2×10^{-5}	684.3
	52 5 47 1	- 52 4 48 1	262.7563	6.2×10^{-5}	684.3
	48 14 35 1	- 48 13 35 0	262.7790	8.2×10^{-5}	670.1
	58 4 54 0	- 57 5 52 1	262.7833	1.7×10^{-5}	826.0
	58 5 54 0	- 57 6 52 1	262.7835	1.8×10^{-5}	826.0
	52 6 47 0	- 52 5 48 0	262.7850	6.2×10^{-5}	684.3
	52 5 47 0	- 52 4 48 0	262.7851	6.2×10^{-5}	684.3
	49 5 44 0	- 48 6 42 1	262.7944	2.3×10^{-5}	612.1
	56 11 45 1	- 56 10 47 0	262.8056	3.1×10^{-5}	846.1
	54 10 44 1	- 53 10 43 0	262.8323	7.1×10^{-6}	769.2
	48 14 34 1	- 48 13 36 0	262.8480	8.7×10^{-5}	670.1
	15 6 10 1	- 14 5 9 1	262.9243	8.5×10^{-5}	76.2

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	48 6 43 1	- 47 7 41 0	262.9262	2.2×10^{-5}	588.9
	15 6 10 0	- 14 5 9 0	262.9573	7.7×10^{-5}	76.1
	51 13 39 1	- 50 14 37 0	262.9728	2.3×10^{-5}	731.3
	25 4 21 1	- 24 4 20 0	263.0163	1.6×10^{-4}	170.5
	45 3 43 1	- 45 1 44 0	263.0586	2.9×10^{-6}	488.4
	45 2 43 1	- 45 2 44 0	263.0586	2.9×10^{-6}	488.4
	26 3 23 1	- 25 3 22 0	263.0676	9.8×10^{-5}	178.9
	25 5 20 0	- 24 5 19 1	263.0725	1.6×10^{-4}	173.6
	25 7 19 0	- 25 4 21 1	263.0877	3.3×10^{-6}	183.1
	17 5 13 0	- 16 4 13 1	263.1057	9.6×10^{-5}	87.1
	30 6 25 0	- 30 3 27 1	263.1423	3.9×10^{-6}	246.1
	48 5 43 1	- 47 6 41 0	263.1708	2.1×10^{-5}	588.9
	32 5 27 0	- 32 4 29 1	263.1942	6.3×10^{-6}	276.0
	43 2 42 1	- 43 0 43 0	263.3325	1.6×10^{-6}	435.8
	44 10 34 0	- 43 11 32 1	263.4029	3.7×10^{-5}	535.1
	46 14 33 0	- 46 13 33 1	263.5562	8.7×10^{-5}	623.6
	26 18 8 0	- 25 18 7 1	263.5570	8.8×10^{-5}	328.6
	26 18 9 0	- 25 18 8 1	263.5570	8.8×10^{-5}	328.6
	26 17 10 0	- 25 17 9 1	263.5577	9.7×10^{-5}	311.6
	26 17 9 0	- 25 17 8 1	263.5577	9.7×10^{-5}	311.6
	26 19 7 0	- 25 19 6 1	263.5670	7.8×10^{-5}	346.5
	26 19 8 0	- 25 19 7 1	263.5670	7.8×10^{-5}	346.5
	26 16 10 0	- 25 16 9 1	263.5730	1.1×10^{-4}	295.6
	26 16 11 0	- 25 16 10 1	263.5730	1.1×10^{-4}	295.6
	46 14 32 0	- 46 13 34 1	263.5813	8.6×10^{-5}	623.6
	26 20 6 0	- 25 20 5 1	263.5846	6.9×10^{-5}	365.4
	26 20 7 0	- 25 20 6 1	263.5846	6.9×10^{-5}	365.4
	9 9 0 1	- 8 8 1 1	263.5918	1.9×10^{-4}	61.4
	9 9 1 1	- 8 8 0 1	263.5918	1.9×10^{-4}	61.4
	26 21 5 0	- 25 21 4 1	263.6069	5.8×10^{-5}	385.3
	26 21 6 0	- 25 21 5 1	263.6069	5.8×10^{-5}	385.3
	26 15 11 0	- 25 15 10 1	263.6074	1.1×10^{-4}	280.6
	26 15 12 0	- 25 15 11 1	263.6074	1.1×10^{-4}	280.6
	13 7 7 1	- 12 6 6 1	263.6167	1.1×10^{-4}	68.3
	13 7 6 1	- 12 6 7 1	263.6186	1.1×10^{-4}	68.3
	9 9 0 0	- 8 8 1 0	263.6196	1.9×10^{-4}	61.3
	9 9 1 0	- 8 8 0 0	263.6196	1.9×10^{-4}	61.3
	26 22 4 0	- 25 22 3 1	263.6311	4.8×10^{-5}	406.1
	26 22 5 0	- 25 22 4 1	263.6311	4.8×10^{-5}	406.1
	13 7 7 0	- 12 6 6 0	263.6457	1.1×10^{-4}	68.2
	13 7 6 0	- 12 6 7 0	263.6476	1.1×10^{-4}	68.2
	27 2 25 1	- 26 3 24 1	263.6485	8.9×10^{-5}	185.6
	26 23 3 0	- 25 23 2 1	263.6545	3.6×10^{-5}	427.8
	26 23 4 0	- 25 23 3 1	263.6545	3.6×10^{-5}	427.8
	27 2 25 0	- 26 3 24 0	263.6585	1.6×10^{-4}	185.5
	11 8 4 1	- 10 7 3 1	263.6667	1.4×10^{-4}	63.4
	11 8 3 1	- 10 7 4 1	263.6667	1.4×10^{-4}	63.4
	26 14 12 0	- 25 14 11 1	263.6670	1.2×10^{-4}	266.5
	26 14 13 0	- 25 14 12 1	263.6670	1.2×10^{-4}	266.5
	26 24 2 0	- 25 24 1 1	263.6744	2.5×10^{-5}	450.6
	26 24 3 0	- 25 24 2 1	263.6744	2.5×10^{-5}	450.6
	26 25 1 0	- 25 25 0 1	263.6877	1.3×10^{-5}	474.3
	26 25 2 0	- 25 25 1 1	263.6877	1.3×10^{-5}	474.3
	11 8 4 0	- 10 7 3 0	263.6951	1.4×10^{-4}	63.3
	11 8 3 0	- 10 7 4 0	263.6951	1.4×10^{-4}	63.3
	26 13 13 0	- 25 13 12 1	263.7599	1.3×10^{-4}	253.4

Table E.3. continued.

Species	Transition		Frequency (GHz)	A_{ij} (s^{-1})	E_{up} (K)
	(J K L M)	- (J K L M)			
	26 13 14 0	- 25 13 13 1	263.7599	1.3×10^{-4}	253.4
	43 9 35 1	- 42 10 32 1	263.7620	2.4×10^{-5}	504.6
CH ₂ OHCHO	27 11 16	- 27 10 17	260.1657	2.8×10^{-4}	283.9
	27 11 17	- 27 10 18	260.1817	2.8×10^{-4}	283.9
	45 9 37	- 45 8 38	261.7386	3.0×10^{-4}	632.0
	25 11 14	- 25 10 15	261.7955	2.7×10^{-4}	254.2
	25 11 15	- 25 10 16	261.7990	2.7×10^{-4}	254.2
	41 12 29	- 41 11 30	261.8305	3.3×10^{-4}	569.0
	25 1 24	- 24 2 23	262.0482	4.6×10^{-4}	170.8
	25 2 24	- 24 2 23	262.0513	6.9×10^{-6}	170.8
	25 1 24	- 24 1 23	262.0537	6.9×10^{-6}	170.8
	25 2 24	- 24 1 23	262.0568	4.6×10^{-4}	170.8
	43 7 36	- 43 6 37	262.7212	2.7×10^{-4}	570.1
	26 0 26	- 25 1 25	262.7943	5.3×10^{-4}	173.0
	26 1 26	- 25 1 25	262.7943	7.1×10^{-6}	173.0
	26 0 26	- 25 0 25	262.7943	7.1×10^{-6}	173.0
	26 1 26	- 25 0 25	262.7943	5.3×10^{-4}	173.0
	43 10 34	- 42 11 31	262.7994	6.2×10^{-5}	593.1
	51 14 37	- 50 15 36	262.9430	6.9×10^{-5}	861.3
	23 11 12	- 23 10 13	263.0488	2.6×10^{-4}	226.9
	23 11 13	- 23 10 14	263.0495	2.6×10^{-4}	226.9
	38 5 33	- 38 4 34	263.3786	2.2×10^{-4}	433.8
	30 1 29	- 30 0 30	263.4168	5.6×10^{-5}	240.8
	30 2 29	- 30 1 30	263.4168	5.6×10^{-5}	240.8
	22 11 11	- 22 10 12	263.5580	2.5×10^{-4}	214.1
	22 11 12	- 22 10 13	263.5582	2.5×10^{-4}	214.1
	46 8 38	- 46 7 39	263.5591	3.0×10^{-4}	657.2
	22 6 16	- 21 6 15	263.7526	6.6×10^{-6}	164.3
	38 6 33	- 38 5 34	263.7558	2.2×10^{-4}	433.8
H ₂ CCO	13 1 13	- 12 1 12	260.1920	2.0×10^{-4}	100.5
	13 8 5	- 12 8 4	261.8202	1.3×10^{-4}	918.9
	13 8 6	- 12 8 5	261.8202	1.3×10^{-4}	918.9
	13 7 6	- 12 7 5	262.0435	1.5×10^{-4}	725.0
	13 7 7	- 12 7 6	262.0435	1.5×10^{-4}	725.0
	13 2 11	- 12 2 10	262.7609	2.0×10^{-4}	140.5
t-HCOOH	12 0 12	- 11 0 11	262.1036	1.9×10^{-4}	82.8
	20 3 17	- 20 2 18	263.0581	5.1×10^{-6}	257.2
	3 2 2	- 2 1 1	263.7624	3.4×10^{-6}	19.2